Biosensors and Nano-Bioelectronics

Lecture I



Prof. Chenzhong Li
Nanobioengineering&Bioelectronics
Lab, Department of Biomedical
Engineering, Florida International
University, E-mail: licz@fiu.edu

Outlines

- Introduction of the lecture
- Terms and definition
- Rational of a biosensor
- Types of biosensor
- Applications of biosensors
- Electrochemistry and biosensors
- Nanotechnology in biosensor

"An important player in 21st century engineering will be the 'biotraditional engineer,' the recipient of a traditional engineer's training and a modicum of exposure to life science." M.H. Friedman, J. Biomechanical Eng, V123, December 2001

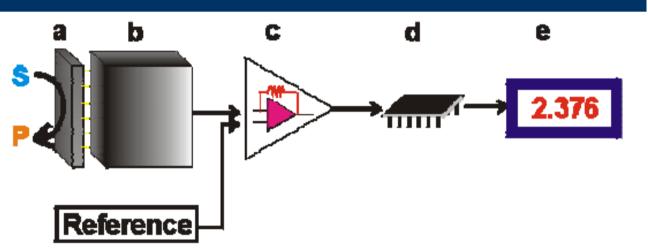
What is biosensor?

Chemical Sensors:

"A chemical sensor is a device that transforms chemcial information, ranging from the concentration of a specific sample component to total composition analysis, into an analytically useful signal" – IUPAC

Biosensors: are analytical tools for the analysis of bio-material samples to gain an understanding of their <u>bio-composition</u>, <u>structure</u> and <u>function</u> by converting a biological response into an electrical signal. The analytical devices composed of a biological recognition element directly interfaced to a signal transducer which together relate the concentration of an analyte (or group of related analytes) to a measurable response.

Biosensor Components



Schematic diagram showing the main components of a biosensor. The bio-reaction (a) converts the substrate to product. This reaction is determined by the *transducer* (b) which converts it to an electrical signal. The output from the transducer is amplified (c), processed (d) and displayed (e).

(http://www.lsbu.ac.uk/biology/enztech/biosensors.html)

Selective Elements and Transducers

Selective elements.

Transducers

```
(Current, potential,
                                                electrochemical:
synthetic ionophores
                                                   - potentiometric Resistance, impedance)
synthetic carriers
supramolecular structures, clusters
                                                   - amperometric
                                                   -- conductimetric
solid layers: metals
                                                   - voltammetric, polarographic
   - metal oxides, crystals
                                                   - impedimetric, capacitive

    polymers, conducting polymers

    piezoelectric

                                                                          (florescence.
organisms
                                                   ntical: light scattering, etc.),
- transmission / absorbance / reflection
microorganisms
                                                optical:
plant and animal tissues
                                                   - dispersion, interferometric
cells
                                                   - polarimetric
organelles

    circular dichroism, ellipsometry

membranes, bilayers and monolayers

    scattering

enzymes
                                                   - emission intensity, photon counting
receptors
                                                     (luminescence) decay time
antibodies
                                                                 (Thermal, temperature)
nucleic acids
                                                calorimetric
                                                acoustic / gravimetric: (Mass Sensitive)
natural organic and inorganic molecules
micelles, reversed micelles
                                                   - surface photo-acoustic wave

    guartz microbalance
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Ref: Spichiger-Keller U.E., "Chemical Sensors and Biosensors for Medical and Biological Applications, Wiley-VCH, 1998

Defining events in the history of biosensor development

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1916	First report on the immobilisation of proteins: adsorption of invertase on activated charcoal
1922	First glass pH electrode
1956	Invention of the oxygen electrode (Clark)
1962	First description of a biosensor: an amperometric enzyme electrode for glucose (Clark)
1969	First potentiometric biosensor: urease immobilised on an ammonia electrode to detect urea
1970	Invention of the Ion-Selective Field-Effect Transistor (ISFET) (Bergveld)
1972/5	First commercial biosensor: Yellow Springs Instruments glucose biosensor
1975	First microbe-based biosensor First immunosensor: ovalbumin on a platinum wire Invention of the pO2 / pCO2 optode
1976	First bedside artificial pancreas (Miles)

Biosensor History (cont.)

1980	First fibre optic pH sensor for <i>in vivo</i> blood gases (Peterson)
1982	First fibre optic-based biosensor for glucose
1983	First surface plasmon resonance (SPR) immunosensor
1984	First mediated amperometric biosensor: ferrocene used with glucose oxidase for the detection of glucose
1987	Launch of the MediSense ExacTech™ blood glucose biosensor
1990	Launch of the Pharmacia BIACore SPR-based biosensor system
1992	i-STAT launches hand-held blood analyser
1996	Glucocard launched
1996	Abbott acquires MediSense for \$867 million
1998	Launch of LifeScan FastTake blood glucose biosensor
1998	Merger of Roche and Boehringer Mannheim to form Roche Diagnostics
2001	LifeScan purchases Inverness Medical's glucose testing business for \$1.3billion

1999-current	BioNMES, Quantum dots, Nanoparticles, Nanocantilever, Nanowire and Nanotube
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Type of Biosensors (by analytes)

Types of Biological Recognition Name of the BIOSENSOR

Elements

Enzymes Enzyme electrode

Proteins

Antibodies Immunosensor

DNA DNA sensor

Organelles

Microbial cells Microbial sensor

Plant and animal tissues

Types of Biosensor (by detection mode)

Types of Transducers	Measured Property
Electrochemical	Potentiometric
	Amperometric
	Voltametric
Electrical	Surface conductivity
	Electrolyte conductivity
Optical	Fluoresence
	Adsorption
	Reflection
Mass sensitive	Rezonans frequency of piezocrytals
Thermal	Heat of reaction
	Heat of adsorpsion

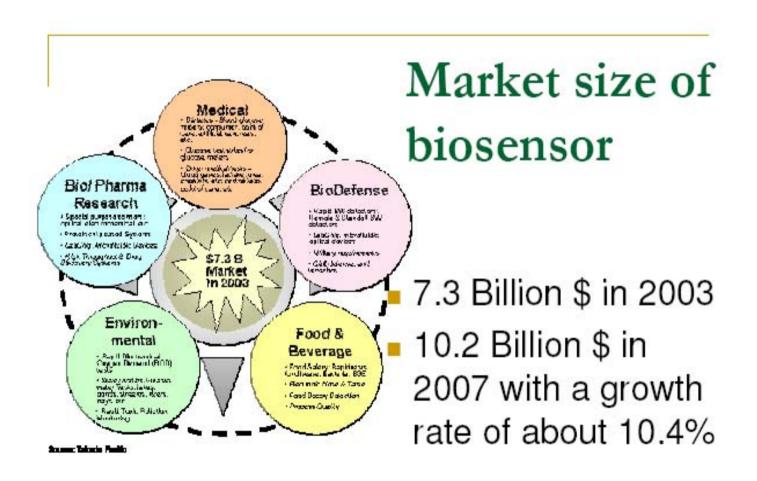
Typical Sensing Techniques for Biosensors

- Fluorescence
- DNA Microarray
- SPR Surface plasmon resonance
- Impedance spectroscopy
- SPM (Scanning probe microscopy, AFM, STM)
- QCM (Quartz crystal microbalance)
- SERS (Surface Enhanced Raman Spectroscopy)
- Electrochemical

Application of Biosensor

- Applications
- Study of biomolecules and how they interact with
- one another
- E.g. Biospecific interaction analysis (BIA)
- Drug Development
- In- home medical diagnosis
- Environmental field monitoring
- Scientific crime detection
- Quality control in small food factory
- Food Analysis

Biosensor Market



Biomedical Diagnostics

- Doctors increasingly rely on testing
- Needs: rapid, cheap, and "low tech"
- Done by technicians or patients
- Some needs for in-vivo operation, with feedback

Glucose-based on glucose oxidase Cholesterol - based on cholesterol oxidase Antigen-antibody sensors - toxic substances, pathogenic bacteria Small molecules and ions in living things: H+, K+, Na+, NO, CO₂, H₂O₂ DNA hybridization, sequencing, mutants and damage

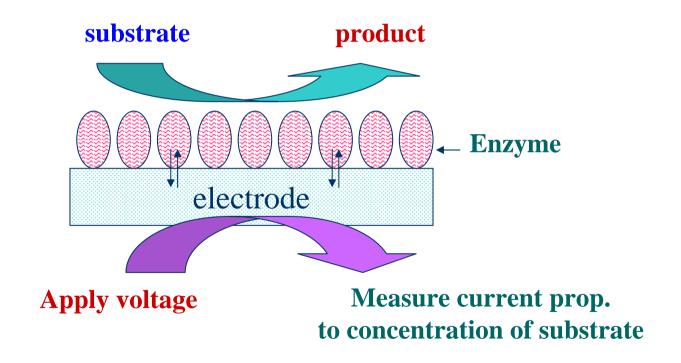
Commercial Glucose Sensors

- Biggest biosensor success story!
- Diabetic patients monitor blood glucose at home
- First made by Clark in 1962, now 5 or more commercial test systems
- Rapid analysis from single drop of blood
- Enzyme-electrochemical device on a slide

Basic Characteristics of a Biosensor

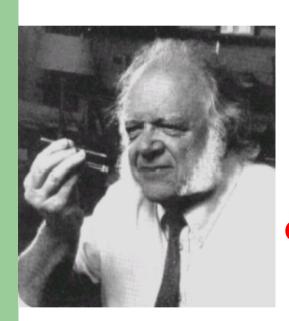
- 1. LINEARITY: Maximum linear value of the sensor calibration curve. Linearity of the sensor must be high for the detection of high substrate concentration.
- **2. SENSITIVITY:** The value of the electrode response per substrate concentration.
- **3. SELECTIVITY:** Interference of chemicals must be minimised for obtaining the correct result.
- **4. RESPONSE TIME:** The necessary time for having 95% of the response.

Principle of Electrochemical Biosensors



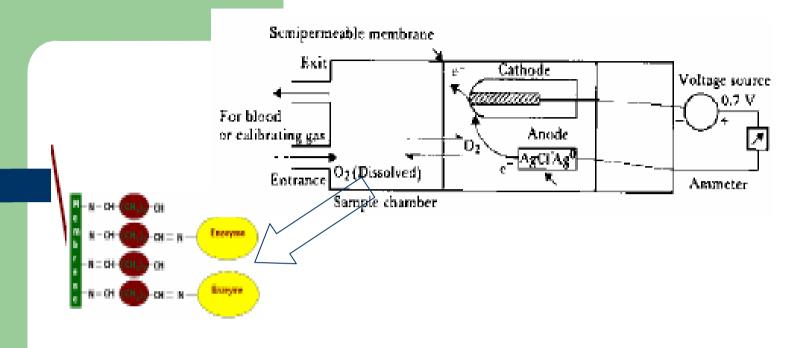
Electrochemical Glucose Biosensor

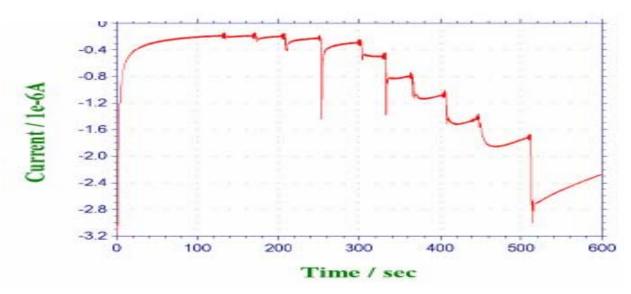




Glucose +
$$O_2$$
 Gluconic Acid + H_2O_2
 H_2O_2 $O_2O_2O_3O_3O_4$ O_2O_4 $O_2O_2O_3O_4$ O_2O_4 O_4 O

The first and the most widespreadly used commercial biosensor: the blood glucose biosensor – developed by *Leland C. Clark in* 1962

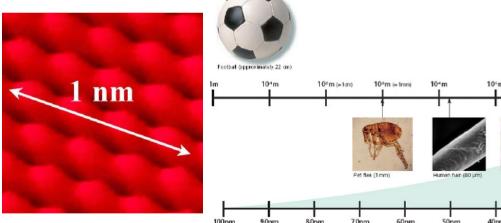




Richard Feynman's (1918-1988) 1959 Talk "There's Plenty of Room at the bottom"

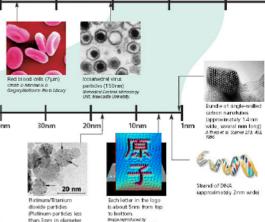
What is Nano?

A nanometre is 1/1,000,000,000 (1 billionth) of a metre, which is around 1/50,000 of the diameter of a human hair or the space occupied by 3-4 atoms placed end-to-end.



A few carbon atoms on the surface of highly oriented pyrolytic graphite (HOPG). Image obtained by Scanning Tunneling Microscope (STM).

Figure 2.1. Length scale showing the nanometre in context. The length scale at the top ranges from 1m to 10^{10} m, and illustrates the size of a football compared to a carbon 60 (C_{QQ}) molecule, also known as a buckyball. For comparison the world is approximately one hundred million times larger than a football, which is in turn one hundred million times larger than a buckyball. The section from 10^{2} m (100 nm) to 10^{2} m (1 nm) is expanded below. The lengthscale of interest for nanoscience and nanotechnologies is from 100 nm down to the atomic scale - approximately 0.2 nm.



are indicated by arrows

on titanium di oxide) P Shobel et al J Catal 222, 294

What Is Nanotechnology?

(Definition from the NNI)

- Research and technology development aimed to understand and control matter at dimensions of approximately 1 - 100 nanometer – the nanoscale
- Ability to understand, create, and use structures, devices and systems that have fundamentally new properties and functions because of their nanoscale structure
- Ability to image, measure, model, and manipulate matter on the nanoscale to exploit those properties and functions
- Ability to integrate those properties and functions into systems spanning from nano- to macro-scopic scales

The First Nanotechnology



The First Nanotechnologists

Size*: 50 nm

Shape: sphere

Color reflected:

Size*: 100 nm

Shape: sphere Color reflected:

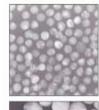
Ancient stained-glass makers knew that by putting varying, tiny amounts of gold and silver in the glass, they could produce the red and yellow found in stained-glass windows. Similarly, today's scientists and engineers have found that it takes only small amounts of a nanoparticle, precisely placed, to change a material's physical properties.

Gold particles in glass Size*: 25 nm Shape: sphere Color reflected: 100 nanometers = 0.0001 millimeter

Silver particles in glass



Had medieval artists been able to control the size and shape of the nanoparticles, they would have been able to use the two metals to produce other colors. Examples:

















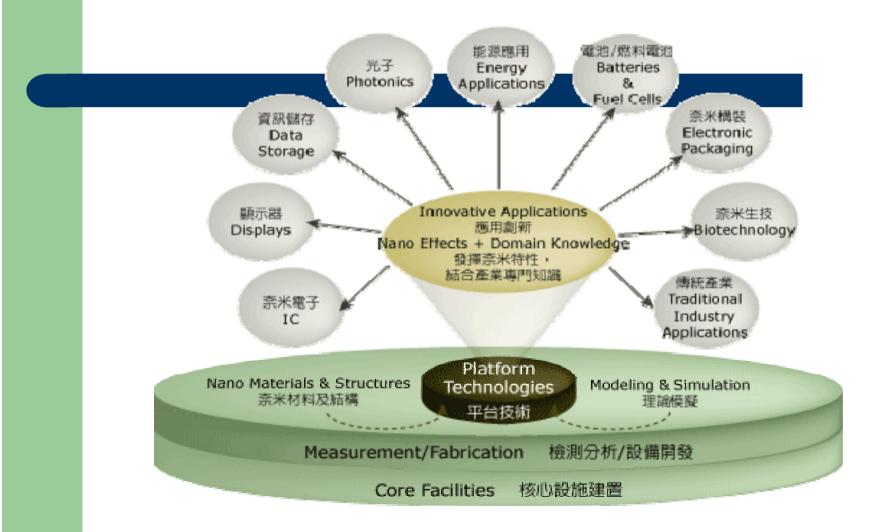


Source: Dr. Chad A. Mirian, Institute of Nanotechnology, Northwestern University

*Approximate



Application of Nanotech



Nanotech in Daily Life

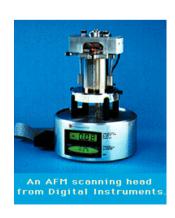


Current Successes in Nanotechnology



Tools In Nanotechnology

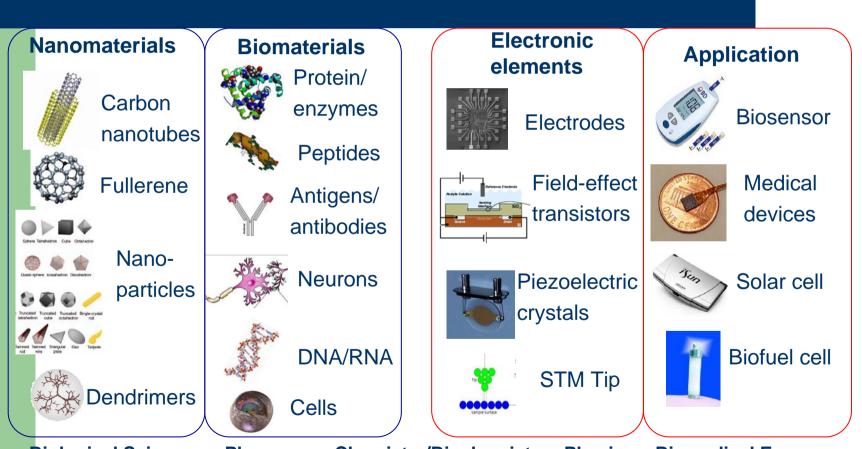
- The main tools used in nanotechnology are four main microscopes
- Transmission Electron Microscope (TEM)
- Atomic Force Microscope (AFM)
- Scanning Tunneling Microscope (STM)
- Scanning Electron Microscope (SEM)





Nano-Biotechnology

Current, Potential, Impedance, Electrical power



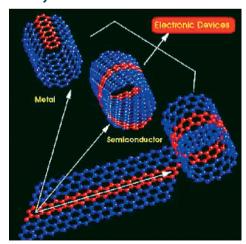
Biological Sciences – Pharmacy – Chemistry/Biochemistry – Physics – Biomedical Eng. – Electrical Eng. – Mechanical Eng. – Material Eng. – Bioinformatics

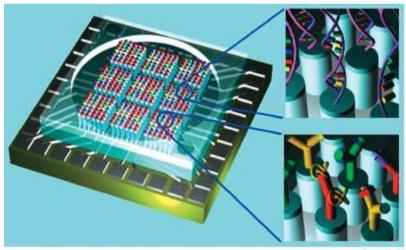
Nanotechnology will enable us to design sensors that are much smaller, less power hungry, and more sensitive than current micro- or macrosensors.



➤ Nano Materials: Carbon Nanotube-Electrodes; Metallic Nanoparticles-sensor probes and electrodes; Nanorod-sensor probes; Magnetic Particles-sensor probes; Nanowires-FET sensing system, quantam dot (AsSe, CdSe, etc.)

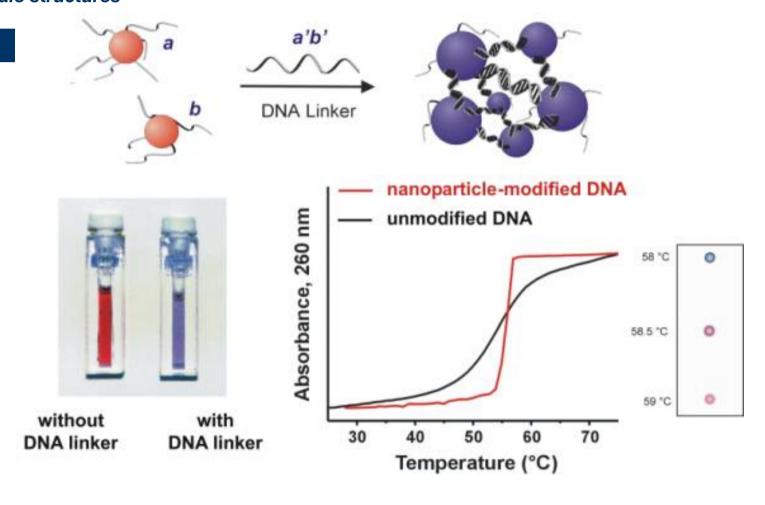
➤ Bio-Nanomaterial Hybrids: DNA-Np; DNA-CNTs; Drug-Nps, Peptide-CNTs, etc.



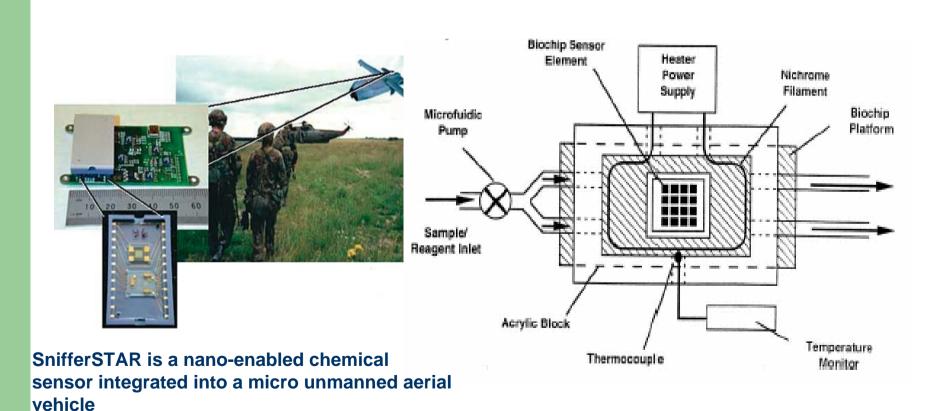


Integration of nano-scale technologies could lead to tiny, low-power, smart sensors that could be manufactured cheaply in large numbers.

sensing the interaction of a small number of molecules, processing and transmitting the data with a small number of electrons, and storing the information in nanometer-scale structures



- Nano/Micro-Electro-Mechanical Systems (N/MEMS) for Sensor Fabrication
- BioMEMS/BioNEMS, Lab-on –Chip, Microfluidic System, Sensor Arrays, Implantable Sensor



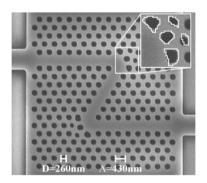
Nanofabrication (Top-Down; Bottom-

Up Nanofabrication

! Nanofabrication methods can be divided into two categories:

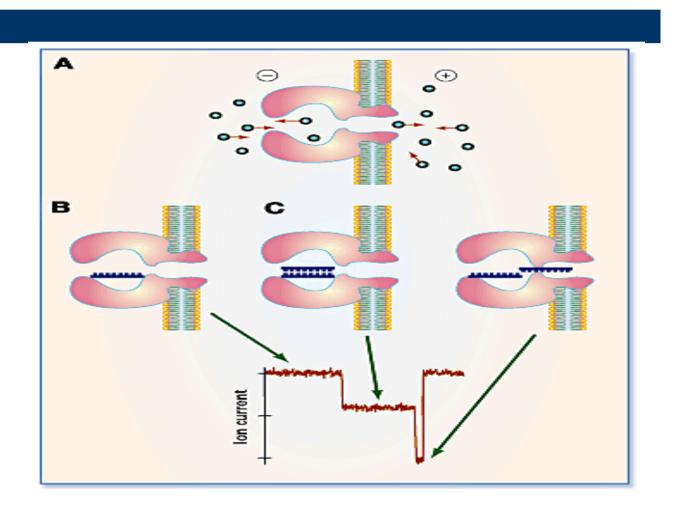
- "Top down" approach
- Micron scale lithography: optical, ultra-violet, Focused lon Beam
- •Electron-beam lithography 10-100 nm





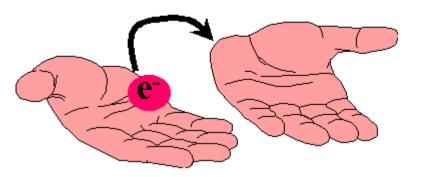
- "Bottom up" approach
- Chemical self-assembly: Man-made synthesis (e.g.
 carbon nanotubes); DNA SAMs, Biological synthesis (DNA, proteins)

Nanopore Technology



Electrochemistry

Where there is oxidation, there is reduction



Substance oxidized loses electron(s)

Substance reduced gains electron(s)

Introduction

Electrochemistry can be broadly defined as the study of charge-transfer phenomena. As such, the field of electrochemistry includes a wide range of different chemical and physical phenomena. These areas include (but are not limited to): battery chemistry, photosynthesis, ion-selective electrodes, coulometry, and many biochemical processes. Although wide ranging, electrochemistry has found many practical applications in analytical measurements.