

FUNDAMENTALS OF NANOELECTRONICS

Basic Concepts

1. The New Perspective

Energy Band Model

3. What and Where

is the Voltage?

4. Heat & Electricity:

Second Law & Information



2.1. Introduction

2.2. $E(p)$ or $E(k)$ relation

2.3. Counting States

2.4. Density of states

2.5. Number of modes

2.6. Electron density (n)

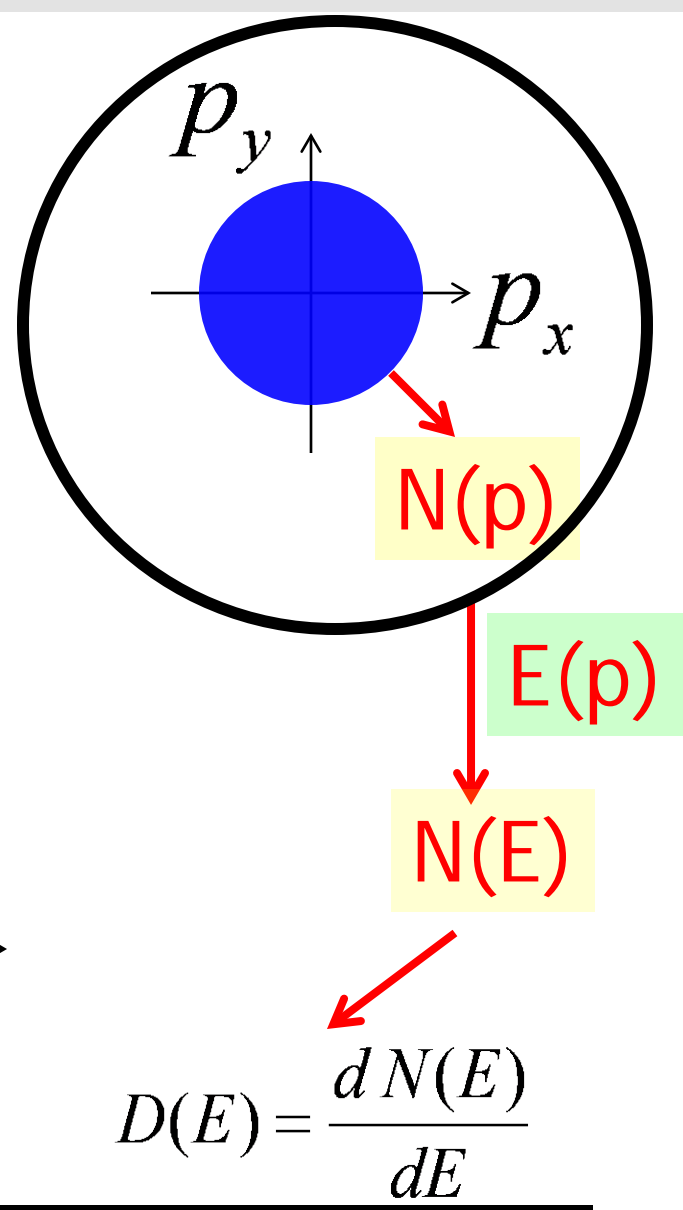
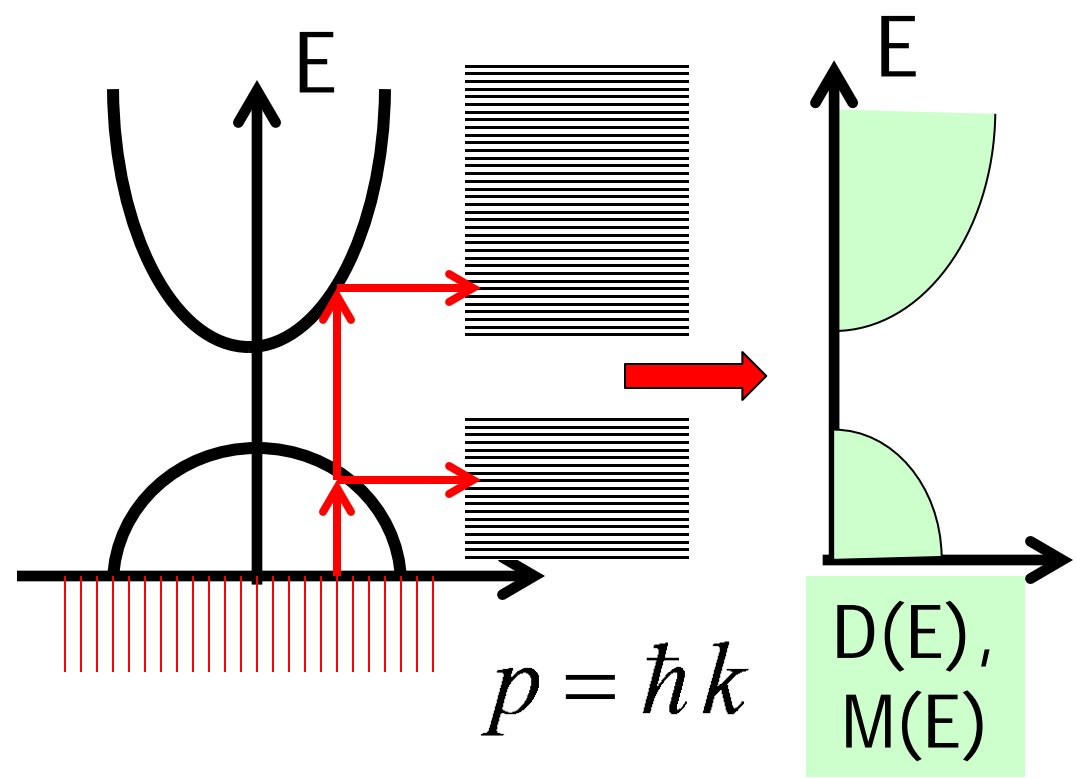
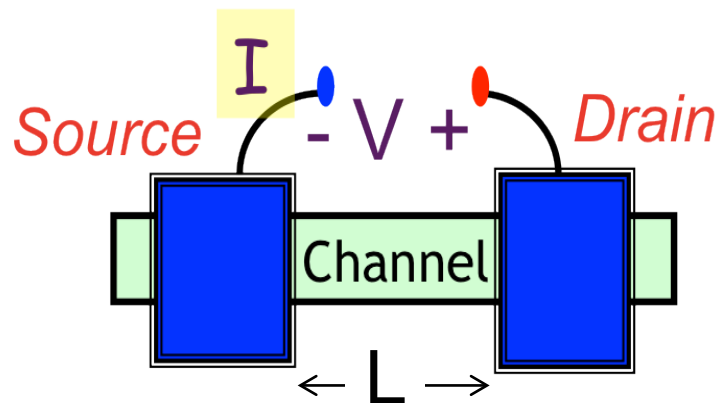
2.7. Conductivity vs n

2.8. Quantum Capacitance

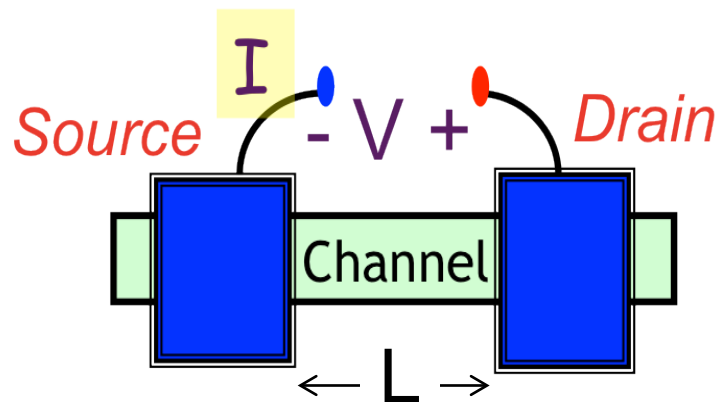
2.9. The Nanotransistor

2.10. Summing up ..

2.3a Counting states

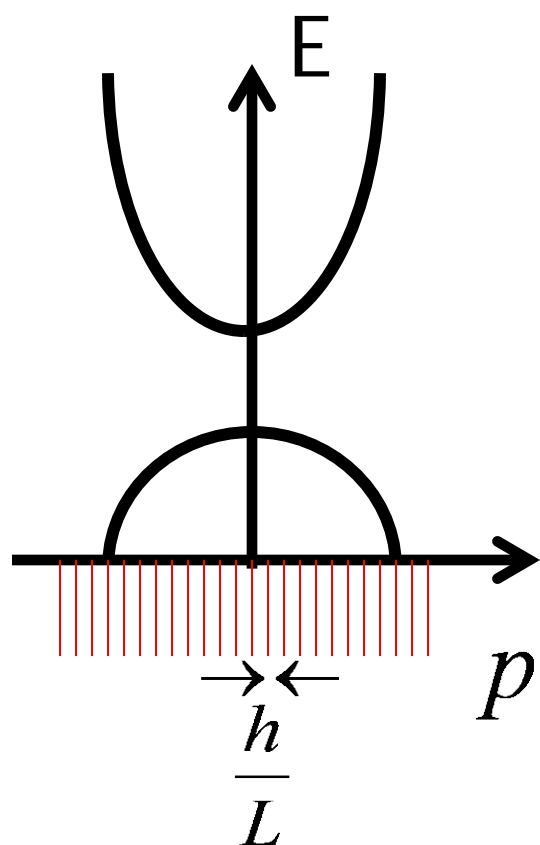


2.3b Counting states



$$\lambda_{\text{De Broglie}} = \frac{h}{p}$$

$$N(p) = \frac{2p}{h/L}$$



$$L = n \frac{h}{p_n}$$

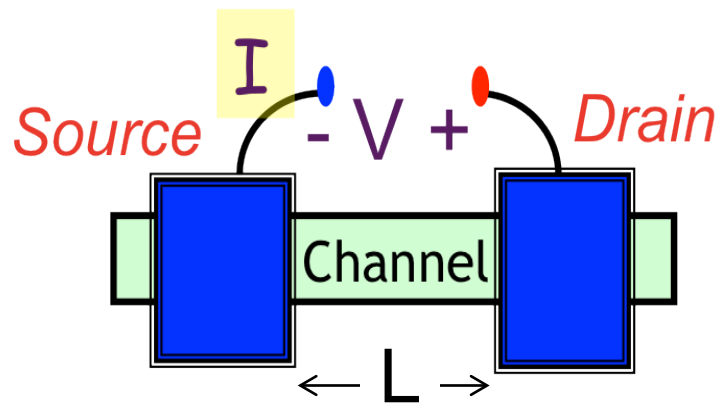
$$\rightarrow p_n = n \frac{h}{L}$$

$$\rightarrow k_n = \frac{n}{L} \frac{h}{\hbar} = n \frac{2\pi}{L}$$

$$p = \hbar k$$

$$\rightarrow \hbar \frac{2\pi}{\lambda_{\text{De Broglie}}} = \frac{h}{\lambda_{\text{De Broglie}}}$$

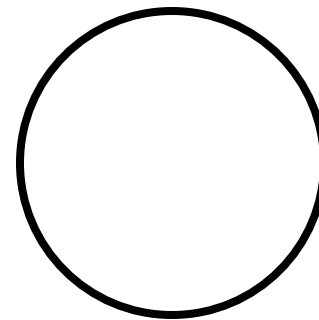
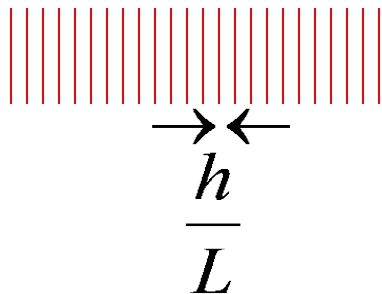
2.3c Counting states



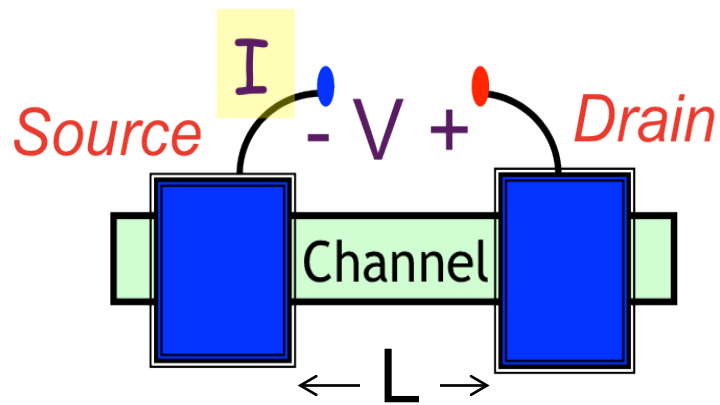
Periodic
Boundary
Condition
PBC

$$L = n \frac{h}{p_n} \rightarrow p_n = n \frac{h}{L}$$

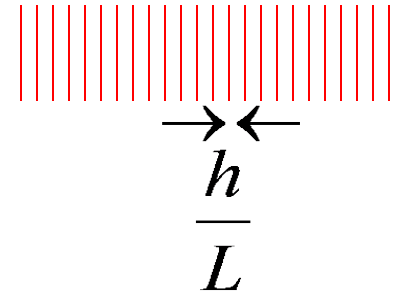
1D $N(p) = \frac{2p}{h/L}$



2.3d Counting states



1D $N(p) = \frac{2p}{h/L}$



2D $N(p) = \frac{\pi p^2}{(h/L)(h/W)}$

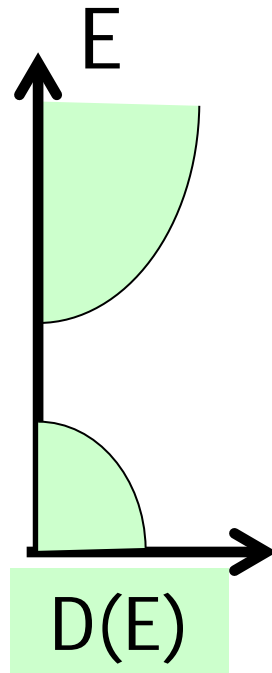
$$N(p) = \left(\frac{p}{h}\right)^d \left\{ \begin{array}{ccc} 1D & 2D & 3D \\ 2L & \pi WL & \frac{4\pi}{3} AL \end{array} \right\}$$

Periodic
Boundary
Condition
PBC

3D $N(p) = \frac{4\pi p^3 / 3}{(h/L)(h^2/A)}$

Coming up next ..

$$N(p) = \left(\frac{p}{h}\right)^d \begin{cases} 1D & 2D & 3D \\ 2L & \pi WL & \frac{4\pi}{3} AL \end{cases}$$



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