

Organic photovoltaics:

*Challenges and opportunities for
theory, modeling, & simulation*

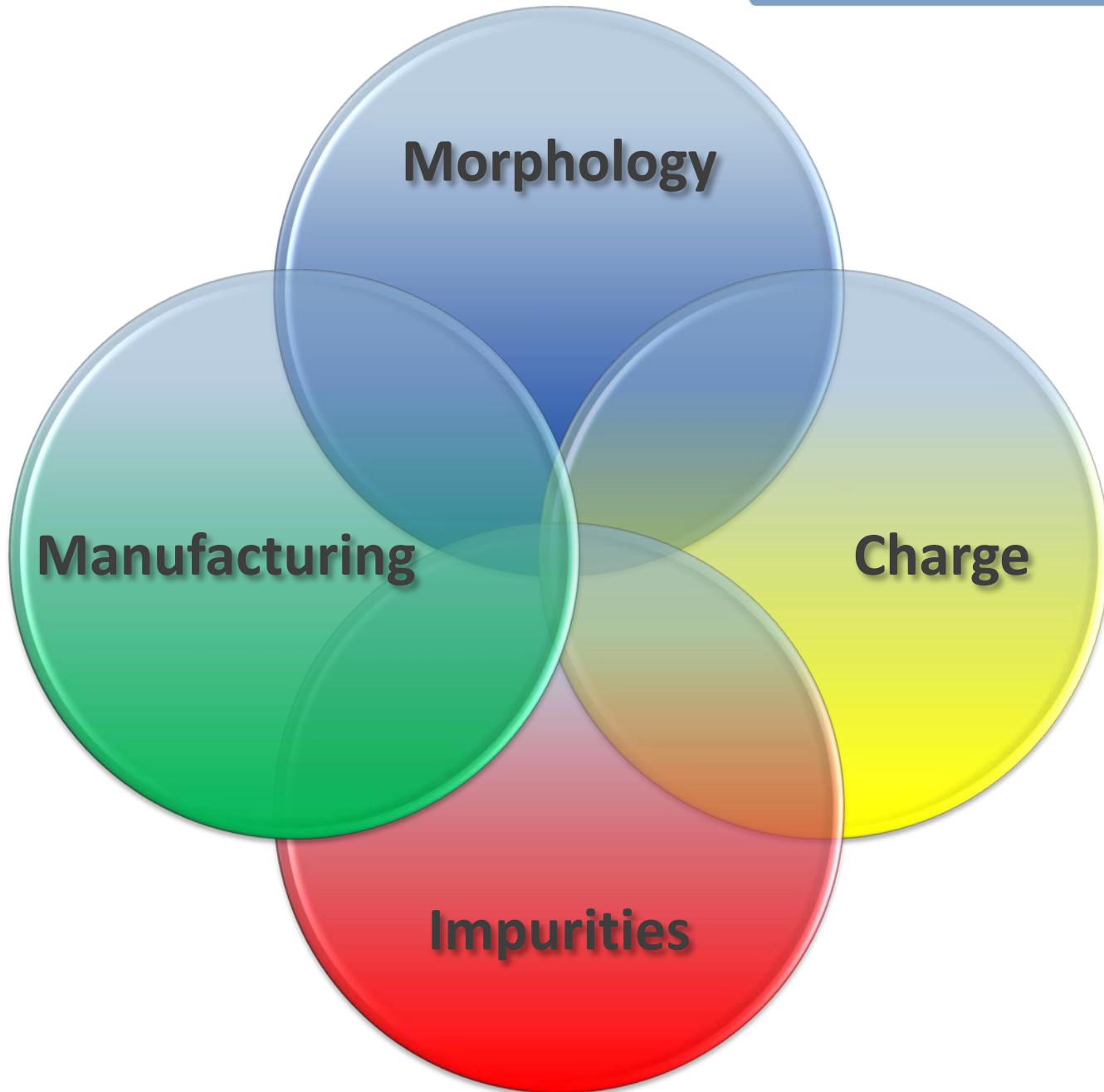
Seth Darling



Center for
Nanoscale
Materials

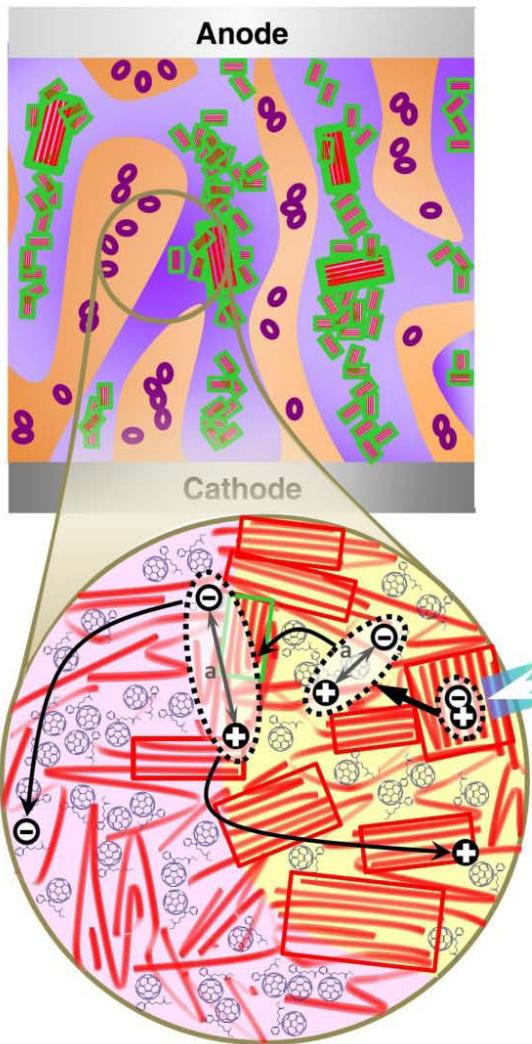


THE INSTITUTE FOR
MOLECULAR
ENGINEERING



Morphology

Hierarchical morphologies in the active layer



- ↔ Sun light
- ↔ PTB7 copolymers
- ↔ PC₆₁BM
- ↔ PTB7 copolymer crystallites
- ↔ PC₆₁BM crystallites
- ↔ PTB7-rich domains
- ↔ PC₆₁BM-rich domains
- ↔ Exciton
- ↔ Bound polaron pair with a separation distance of a
- ↔ Fully dissociated charge carriers
- ↔ Electron transfer from PTB7 to PC₆₁BM
- ↔ Bound polaron pair diffusion/migration of dissociated charges

Nano Letters 11 (2011) 3707
Energy Environ. Sci. 5 (2012) 8045

Morphology

- How and when do each of these morphological features form?



in solution



during deposition

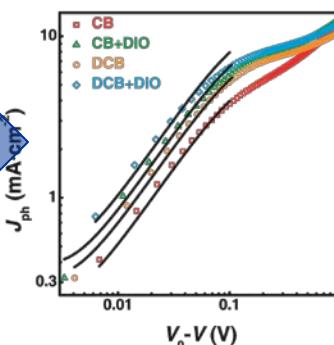
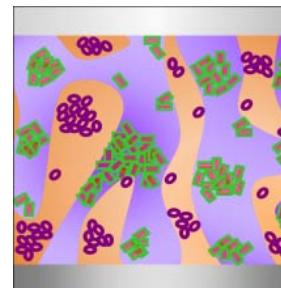


during annealing

Morphology

Charge

- Correlation of morphology and performance: Charge separation

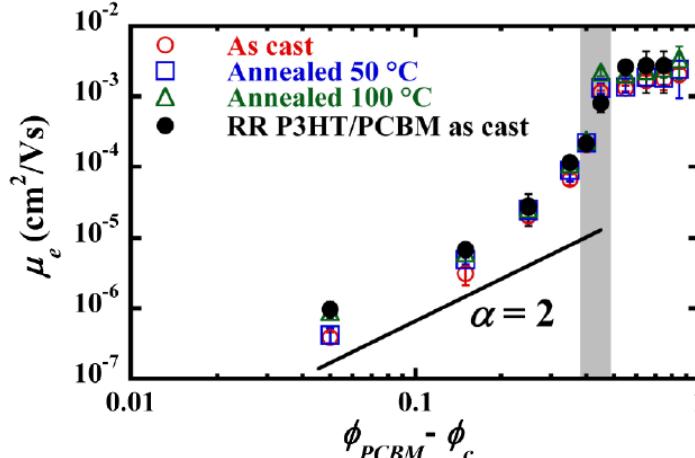


Morphology

Charge

- Correlation of morphology and performance: Charge (polaron) transport
 - Carrier mobilities depend monotonically on miscibility for amorphous blends and can be described with percolation theory

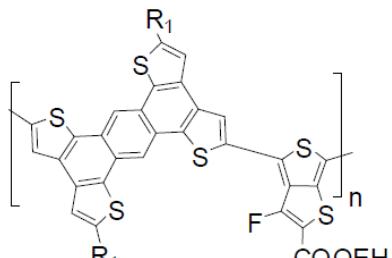
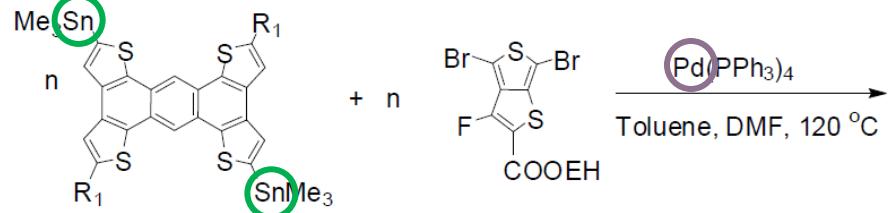
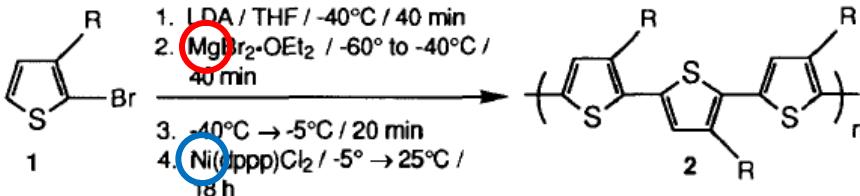
$$(\phi - \phi_c)^\alpha \leftarrow \begin{matrix} \text{dimensionality factor} \\ \text{percolation volume fraction} \end{matrix}$$



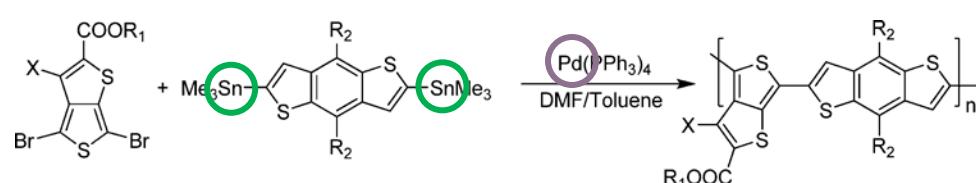
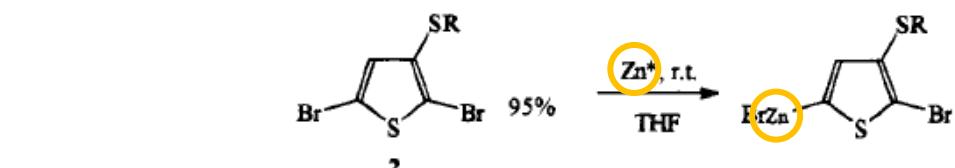
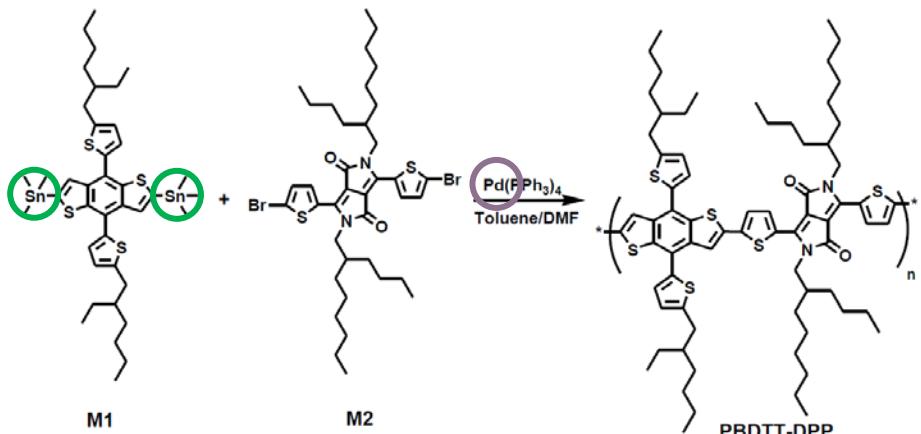
Phys Rev Lett 108 (2012) 026601

- Crystallinity adds complexity: good models for “real” OPV films remain elusive

Impurities



PTTATT-X

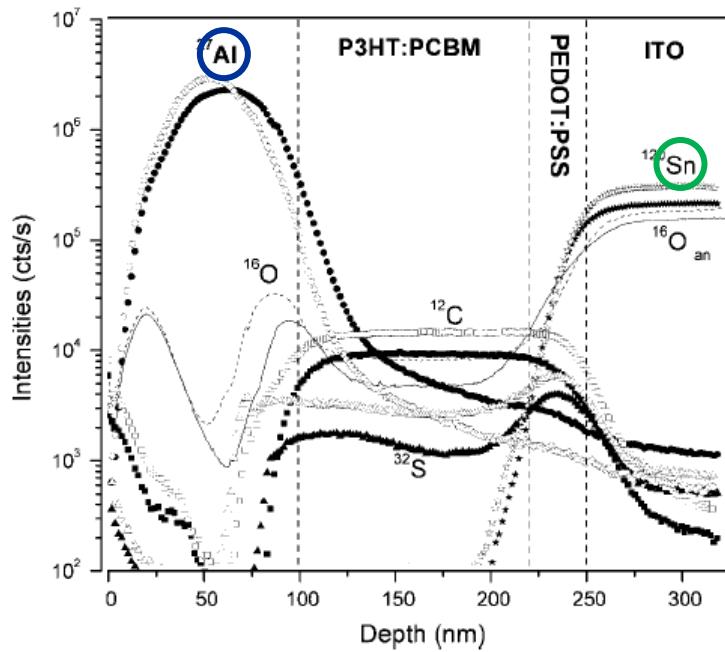


PTB1: X = H, R₁ = n-dodecyl, R₂ = n-octyloxy
PTB2: X = H, R₁ = 2-ethylhexyl, R₂ = n-octyloxy
PTB3: X = H, R₁ = 2-ethylhexyl, R₂ = n-octyl

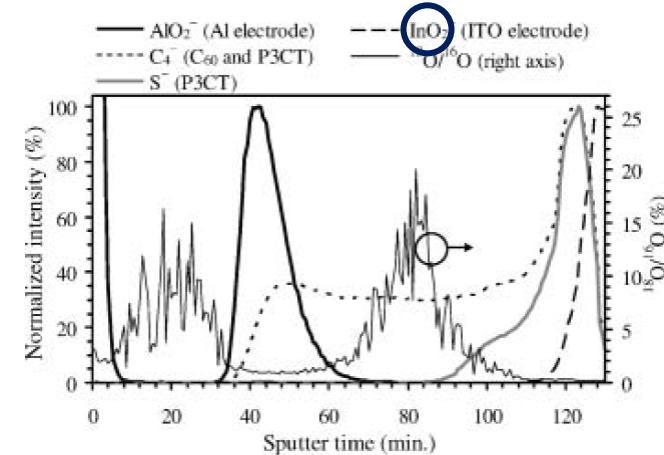
PTB4: X = F, R₁ = n-octyl, R₂ = 2-ethylhexyloxy
PTB5: X = H, R₁ = n-octyl, R₂ = 2-ethylhexyloxy
PTB6: X = H, R₁ = 2-butyloctyl, R₂ = n-octyloxy

Impurities

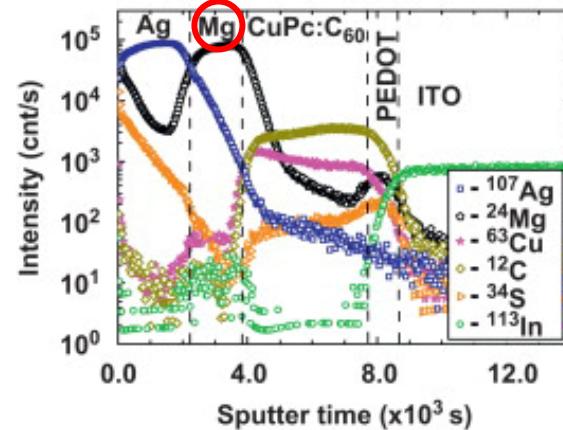
- Diffusion of electrode materials into active layer



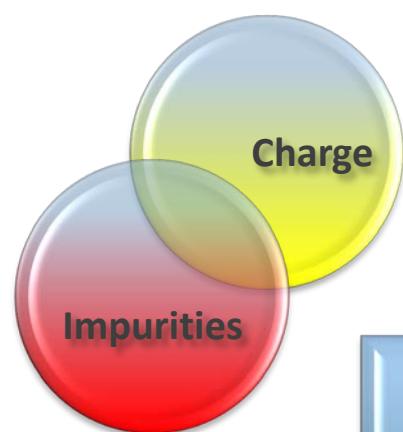
Surf Interface Anal **42** (2010) 1010



Prog Photovolt: Res Appl **15** (2007) 697



Sol Energy Mater Sol Cells **95** (2011) 1489



Charge

Impurities

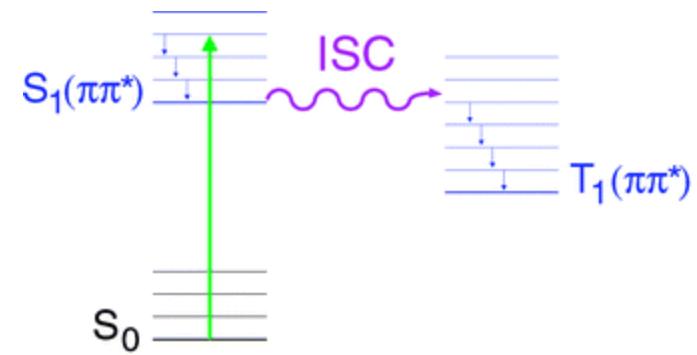
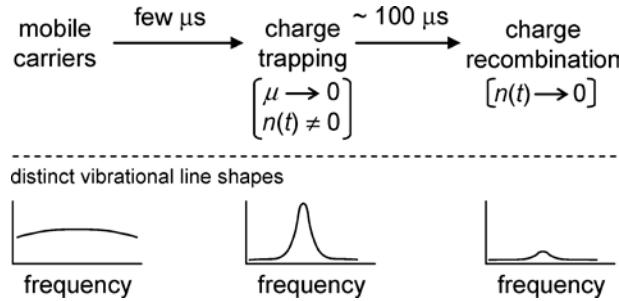
- What, if any, role do these impurities play?

Bad?

- Charge trapping

Good?

- Triplet conversion



J Phys Chem C **114** (2010) 5344

- Work function gradient?

- Performance gap for OPVs is far larger than for traditional PVs

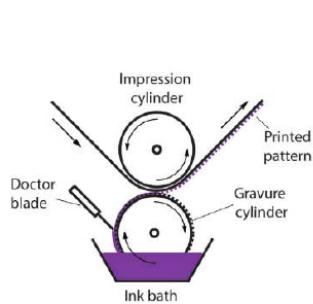
PV technology	Lab record cell PCE*	Commercial module PCE	% Decrease
c-Si	25.0	20.1 (SunPower)	19.6
p-Si	20.4	14.6 (Schott)	28.4
a-Si	10.1	5.1 (Uni-Solar)	49.5
CIGS	19.6	12.5 (Q-Cells)	36.2
CdTe	16.7	12.5 (First Solar)	25.1
DSSC	11.0	1.74 (G24 Innovations)	84.2
OPV	10.0	1.65 (Konarka)	83.5

* *Prog. Photovolt.: Res. Appl.* **20** (2011) 12

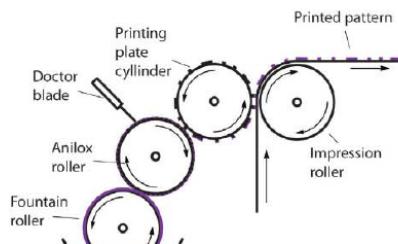
Morphology

Manufacturing

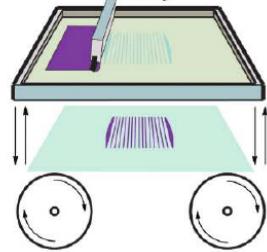
- Deposition techniques are connected with morphology
 - Model kinetics of evolution during solvent evaporation
 - Model role of solvents & additives (viscosity tuning)



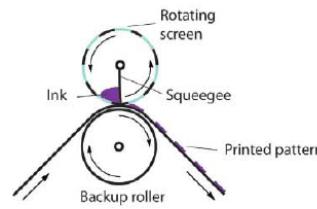
Gravure Printing



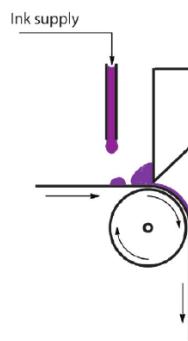
Flexographic Printing



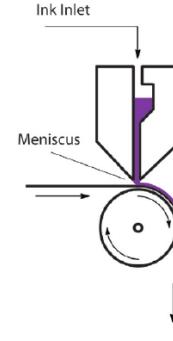
Screen printing



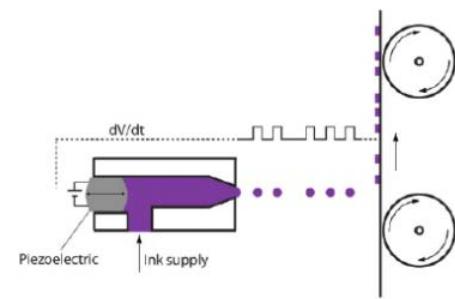
Rotary screen printing



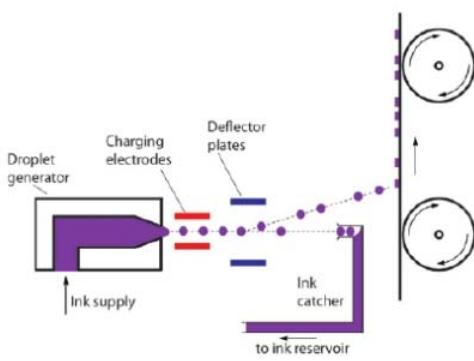
Knife Coating



Slot Die Coating



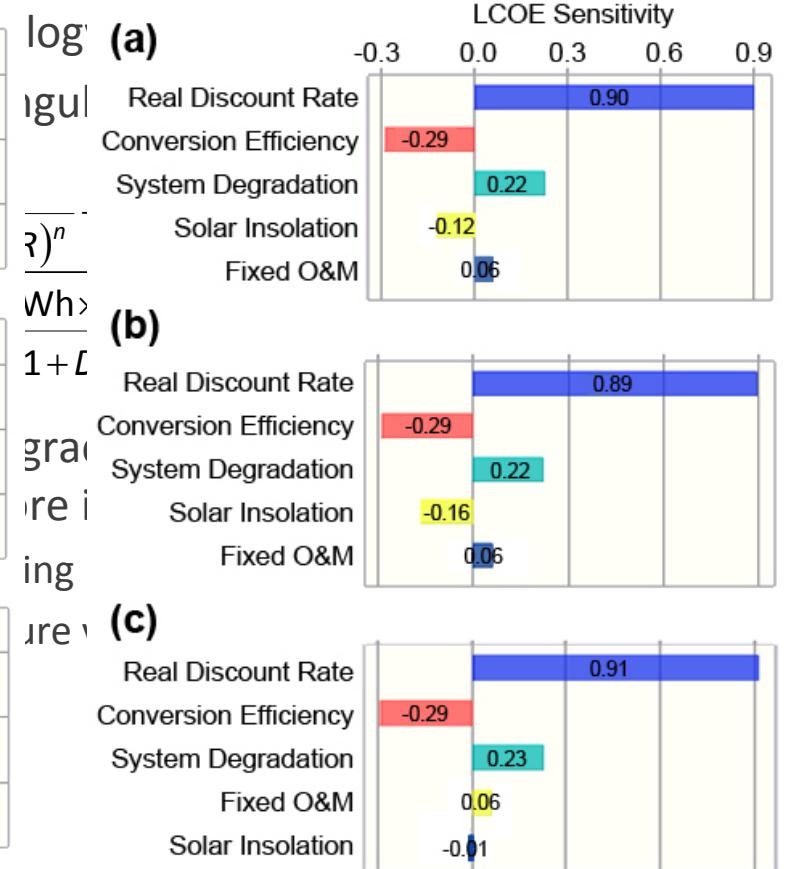
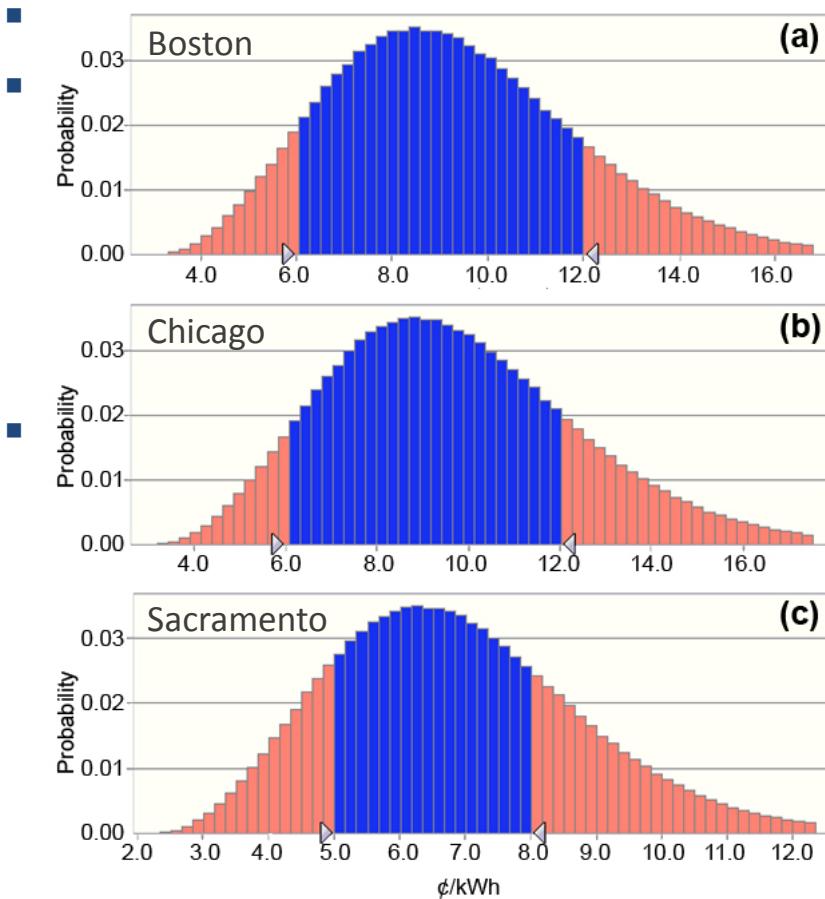
Piezoelectric inkjet
(Droplet on demand, DOD)



Continuous inkjet

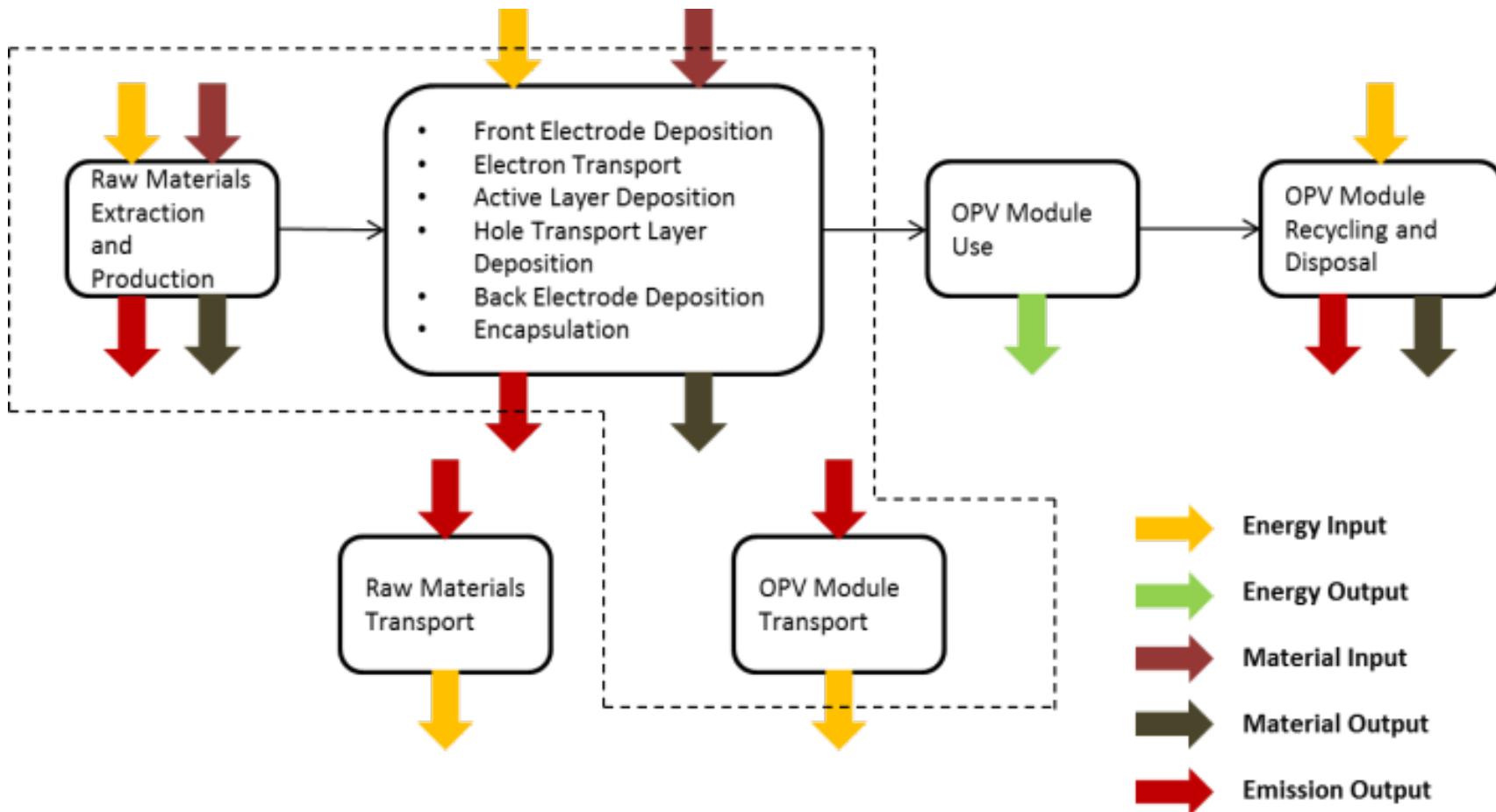
Mater. Today 15 (2012) 36
MRS Bulletin 30 (2005) 50

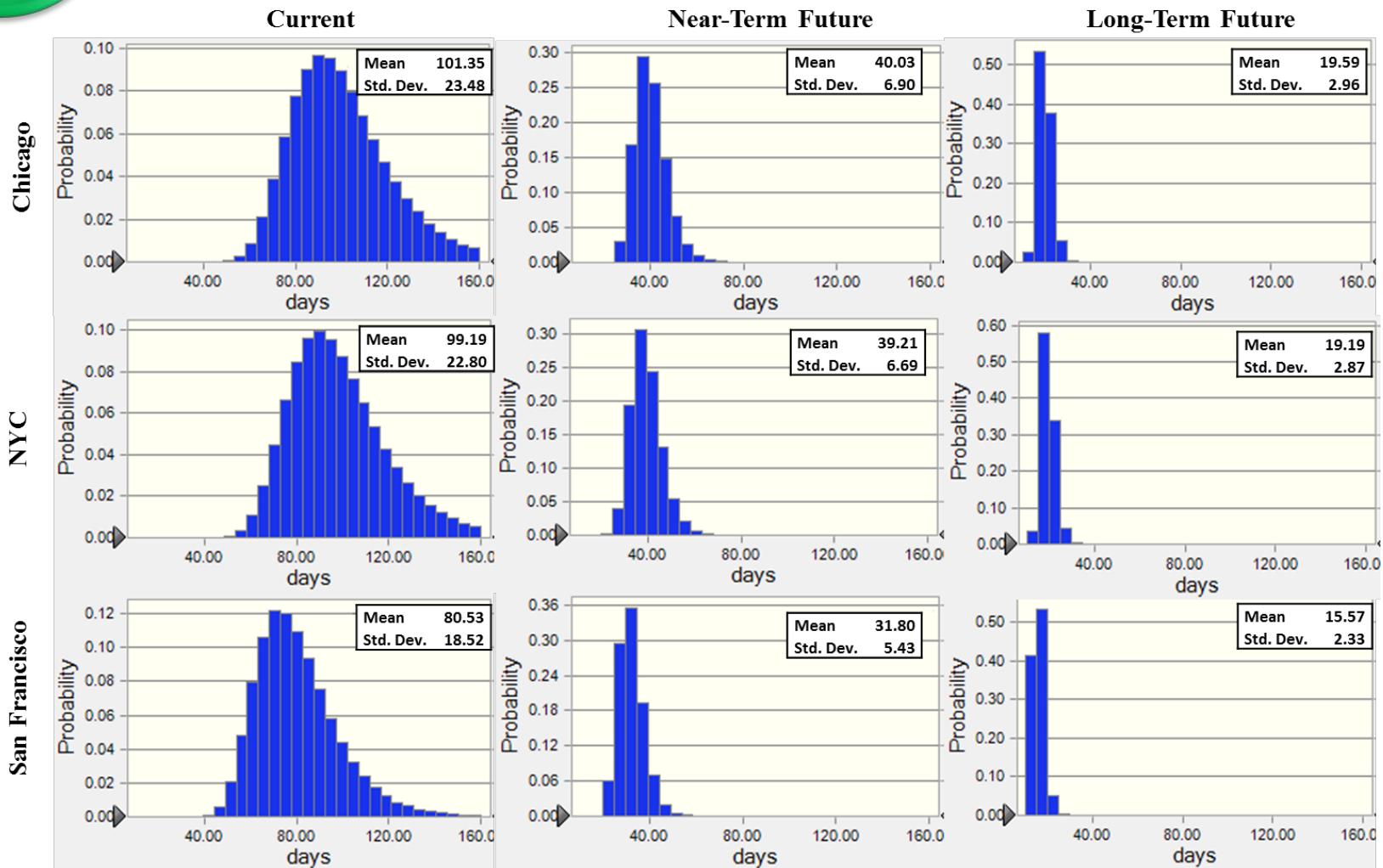
Levelized cost of energy (LCOE)

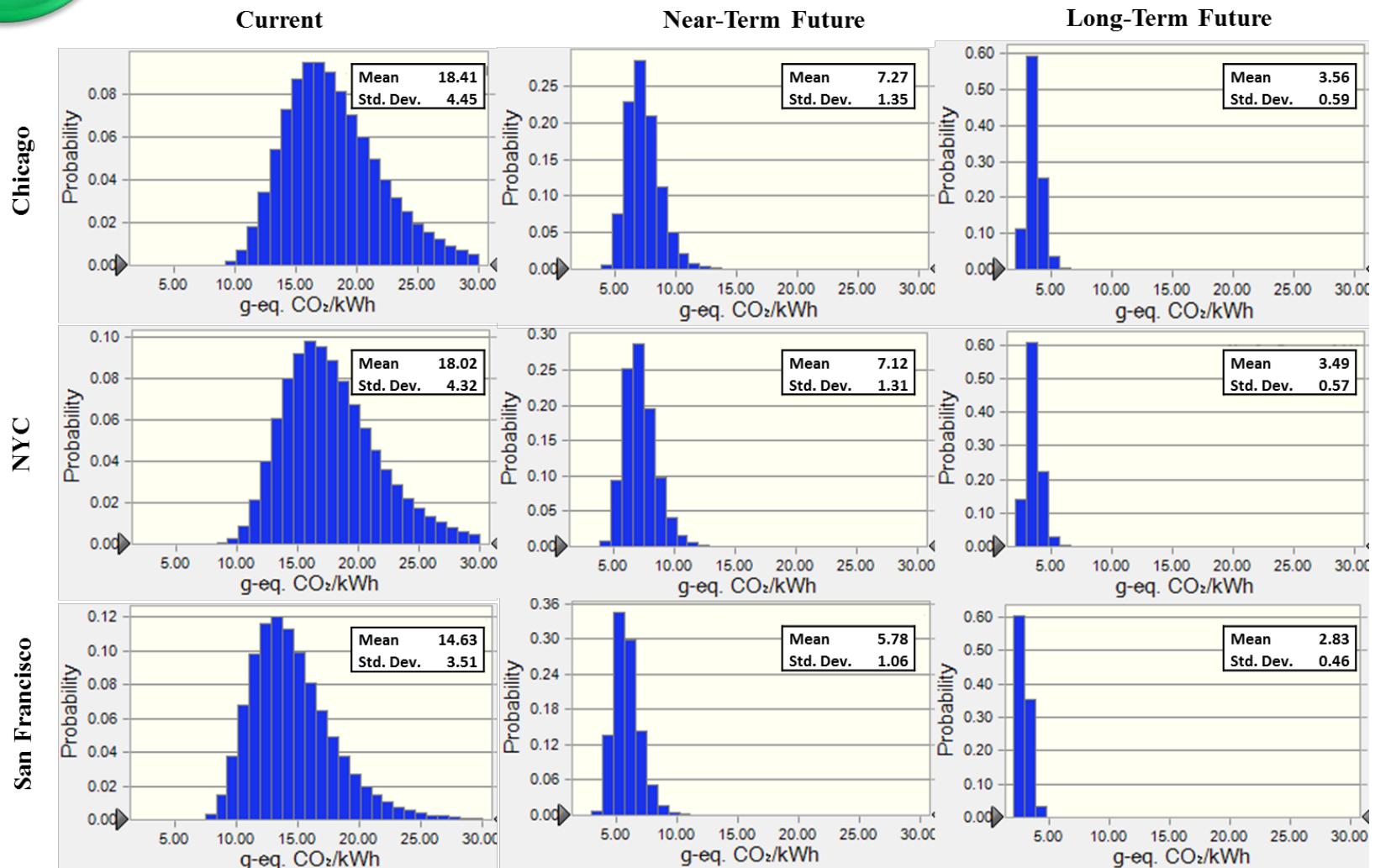


- Growing need for analogous analysis for OPVs



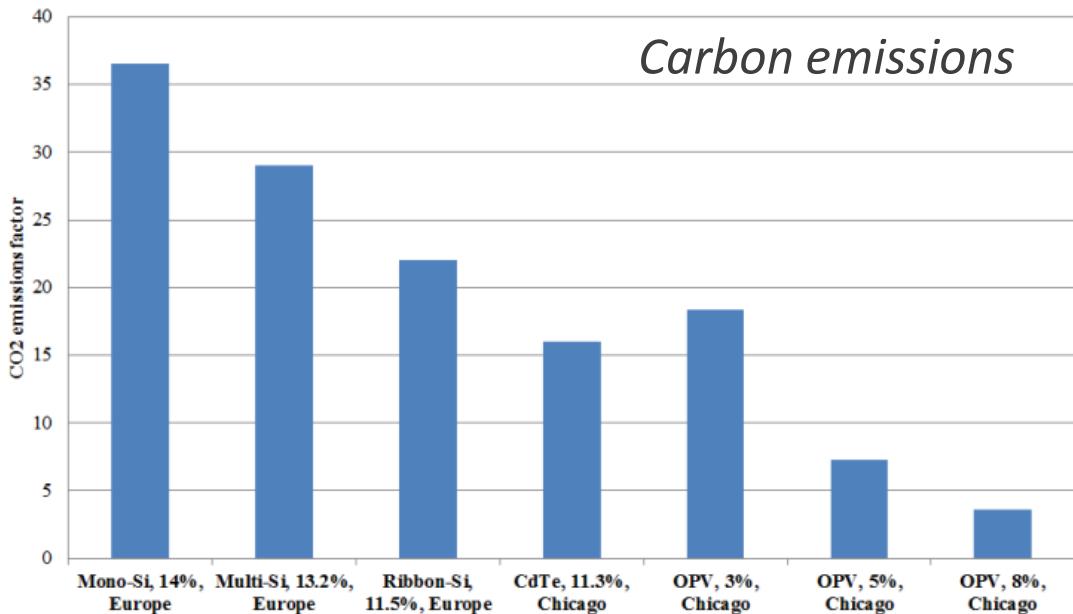
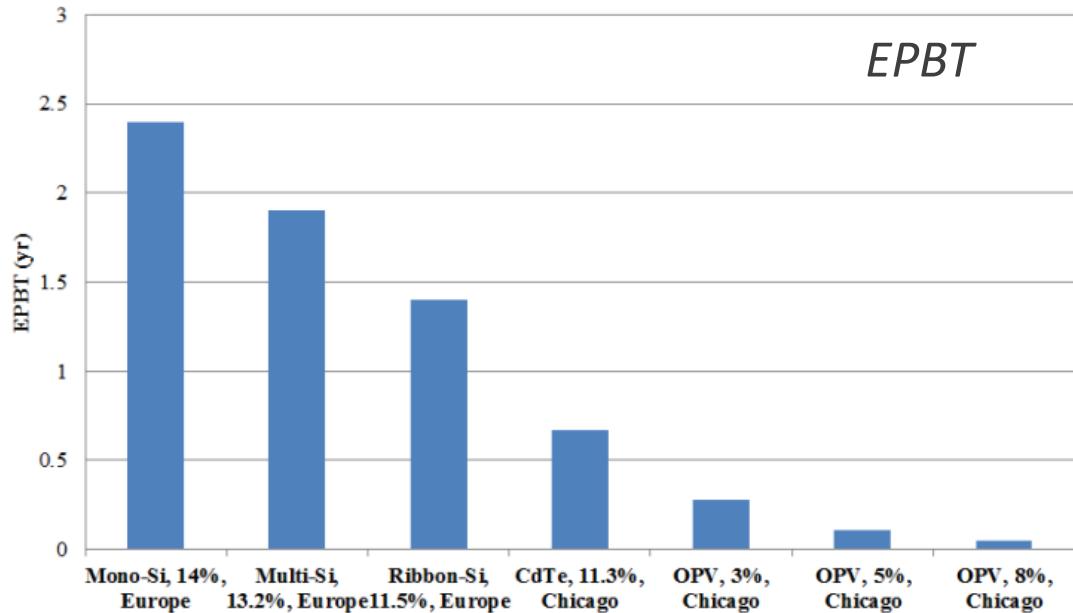
Life cycle analysis (LCA)

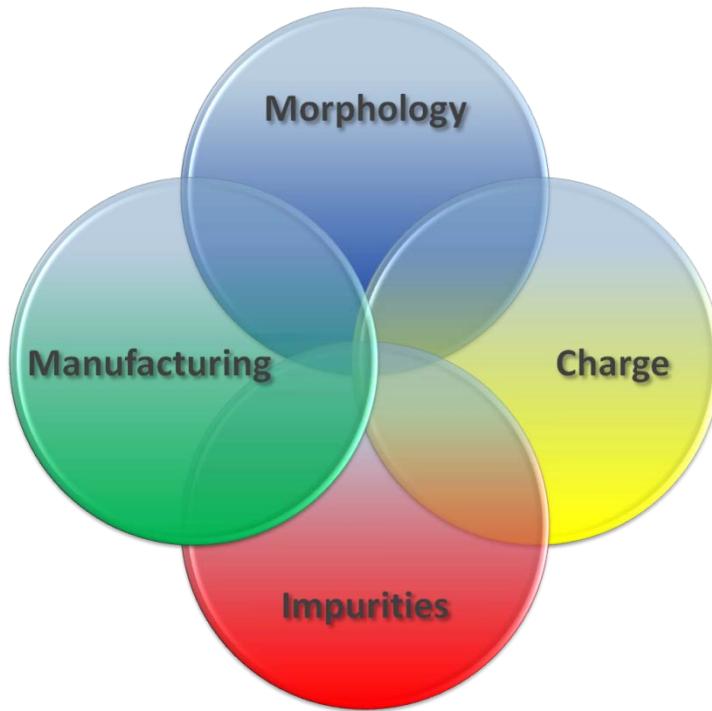
Energy payback time (EPBT)

Carbon emissions

Manufacturing

- Making the case for OPV as a commercial technology will require careful LCA analyses
- Initial studies are encouraging





- Morphology formation mechanisms
- Morphology-function relationships
- Charge separation process
- Role of impurities/dopants
- Life cycle analysis