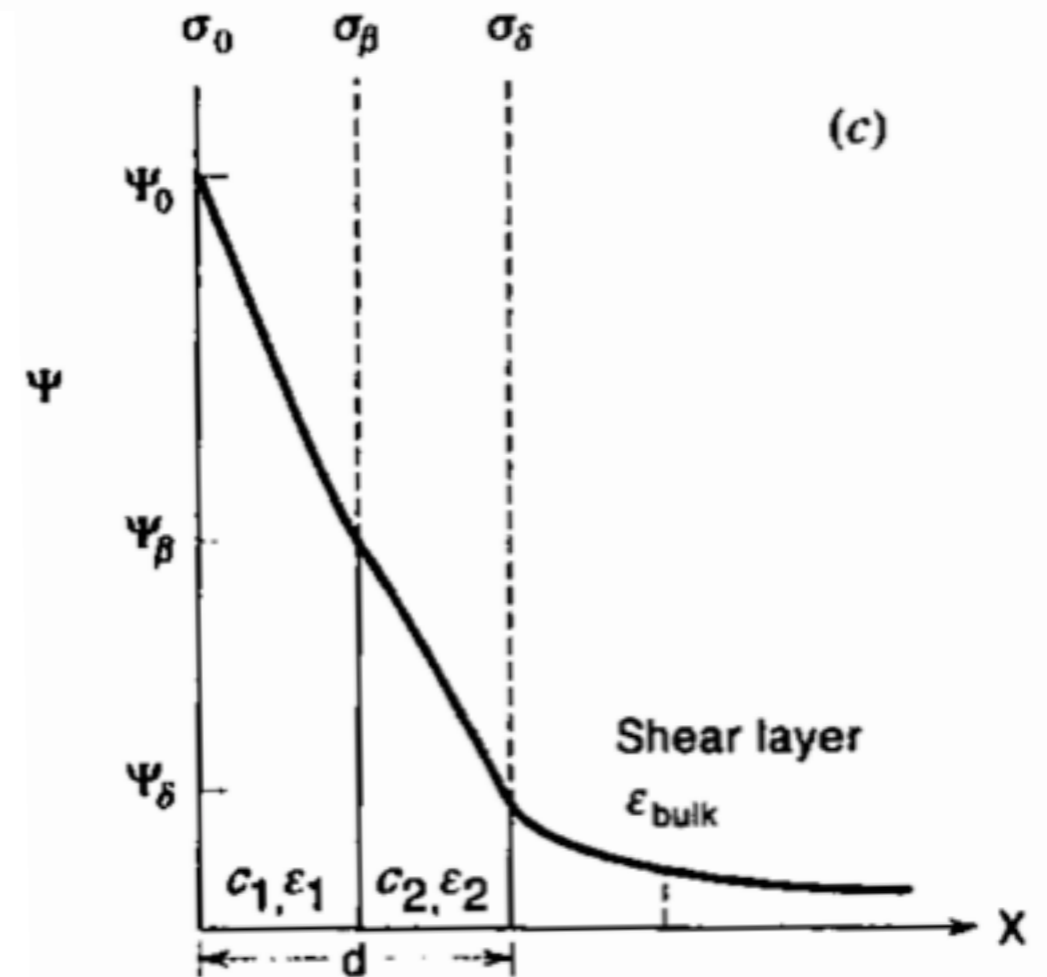
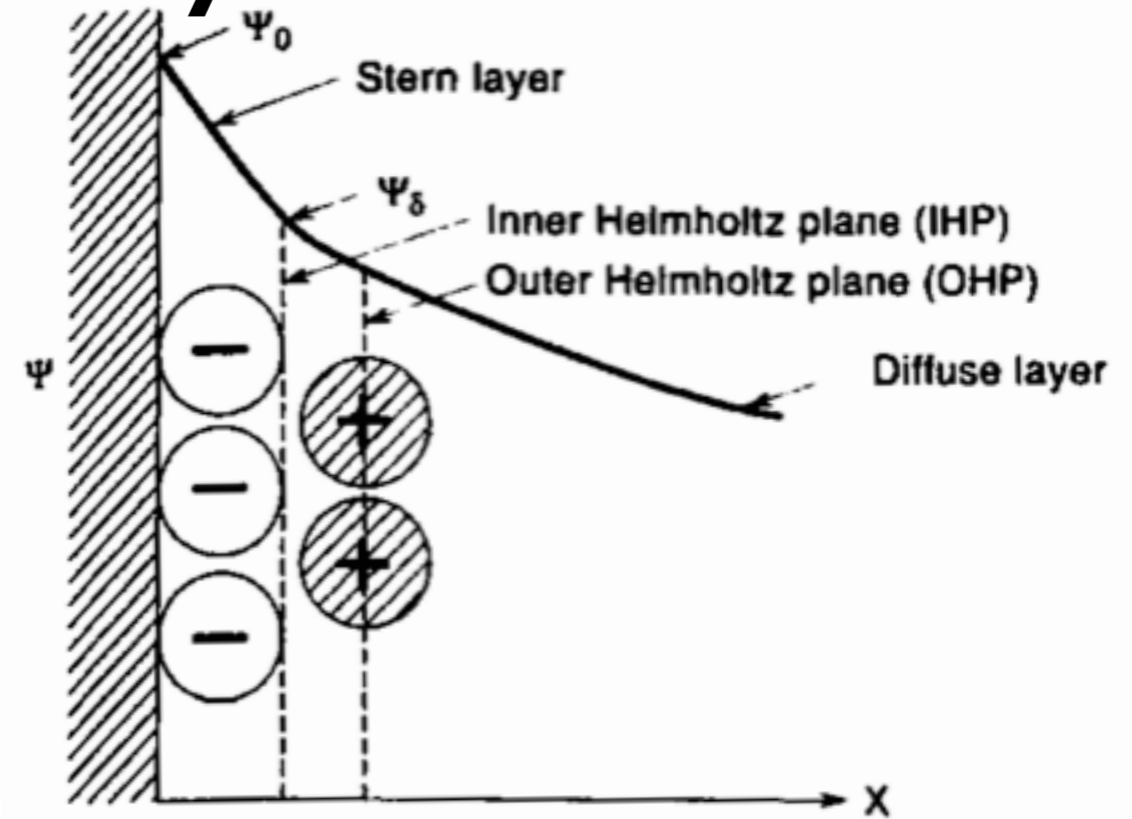
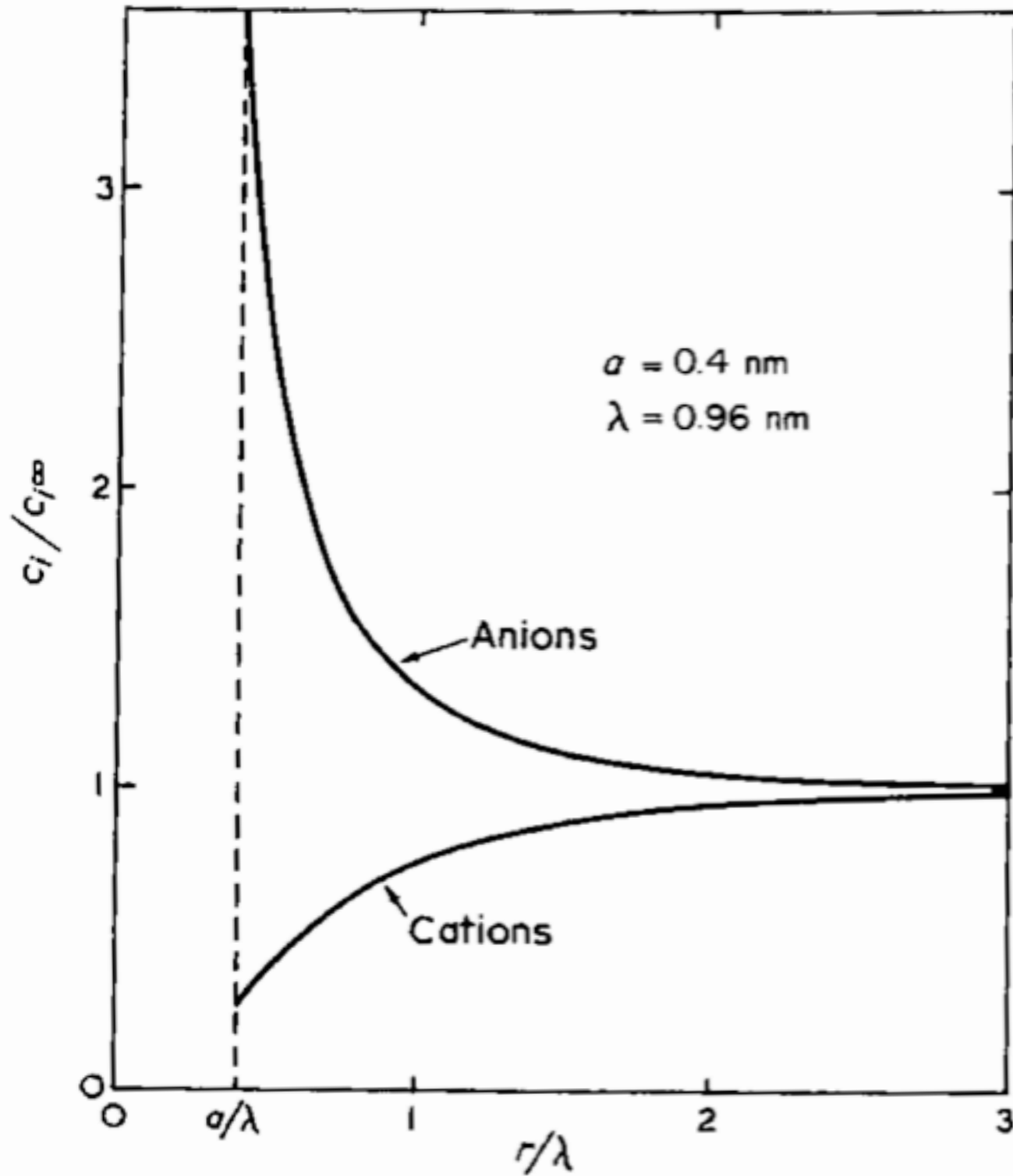


# The Electrode Interface II

Lecture 17

R. Edwin García  
redwing@purdue.edu

# The Electrode-Electrolyte Interface



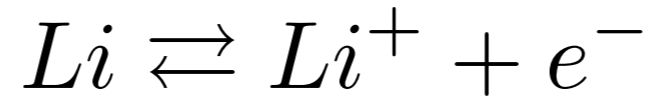
# The Electrode-Electrolyte Interface

$$\nabla^2 \phi = \frac{z e c^*}{\epsilon_r \epsilon_0} \sinh \frac{z e \phi}{k_b T} - \frac{\rho_o(x, y, z)}{\epsilon_r \epsilon_0} \longrightarrow \text{non-linear equation}$$

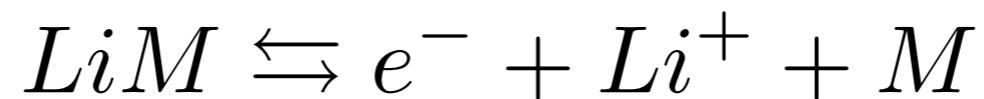
$$\nabla^2 \phi = \frac{z^2 e^2 c^*}{\epsilon_r \epsilon_0 k_b T} \phi - \frac{\rho_o(x, y, z)}{\epsilon_r \epsilon_0} \longrightarrow \text{linearized equation}$$

$$\kappa^2 = \frac{z^2 e^2 c^*}{\epsilon_r \epsilon_0 k_b T} \quad \text{Debye constant}$$

# The Butler-Volmer Relation



Detailing the rate at which Li intercalates

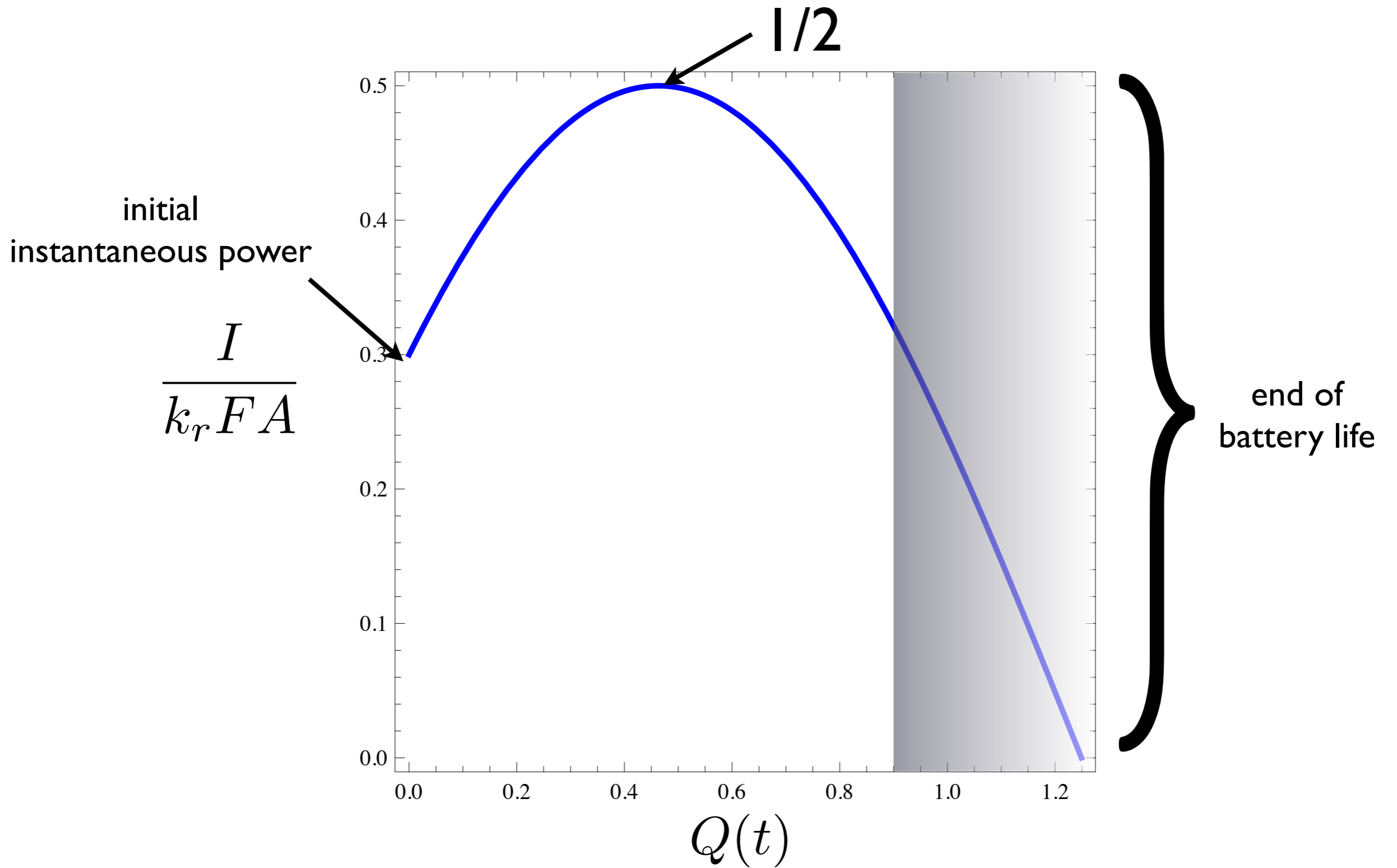


$$i_o = Fk_r(c_T - c_s)^{\alpha_a} c_s^{\alpha_c}$$

The simplest interfacial kinetics

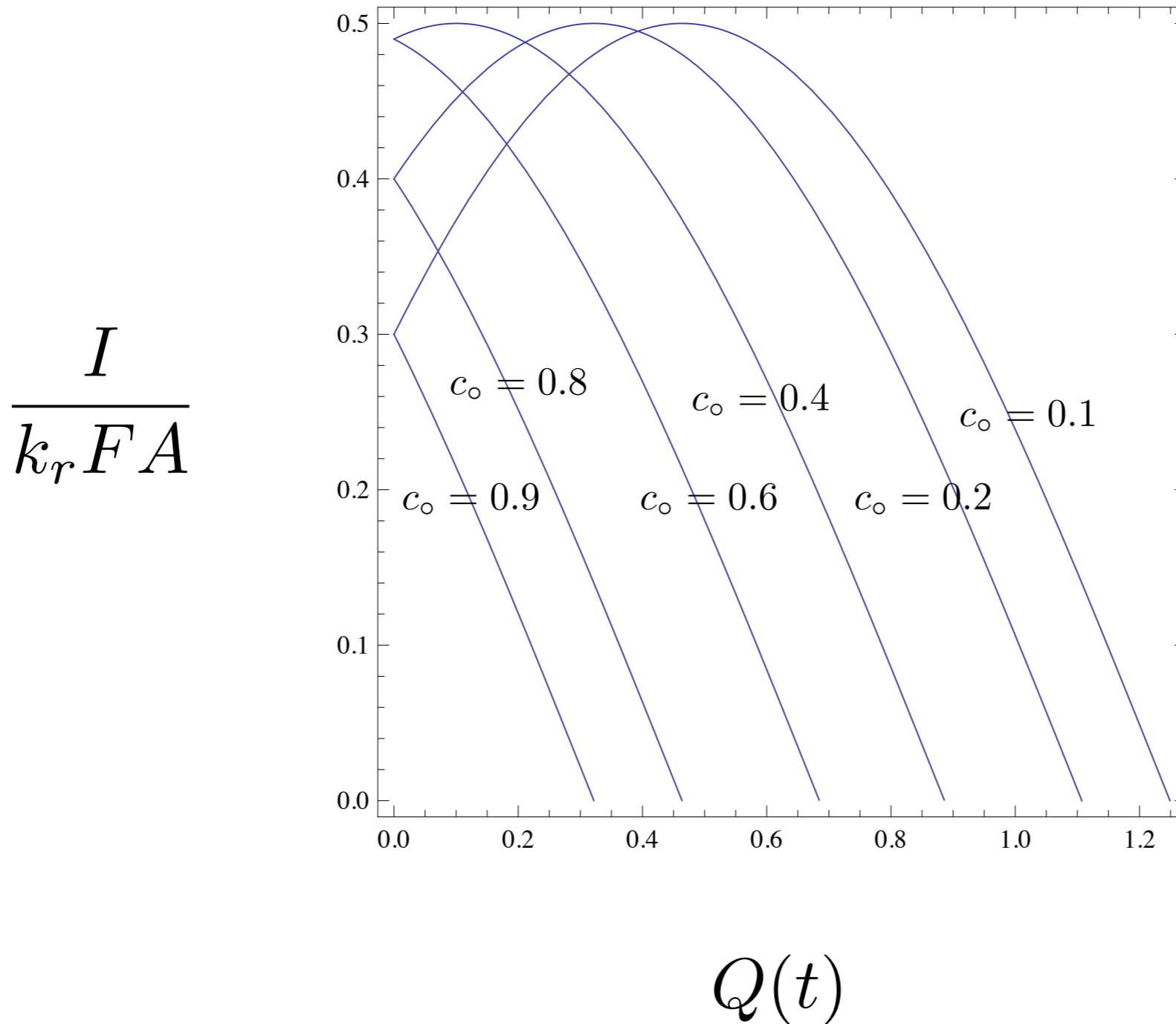
$$\vec{J} \cdot \hat{n} = i_o \left( \exp\left(\frac{\alpha_a F z \eta}{RT}\right) - \exp\left(-\frac{\alpha_c F z \eta}{RT}\right) \right)$$

# Effect on Power Density



$$\frac{\partial c}{\partial t} = A(\eta, T) c^{1/2} (1 - c)^{1/2}$$

# Effect on the Electrode Material

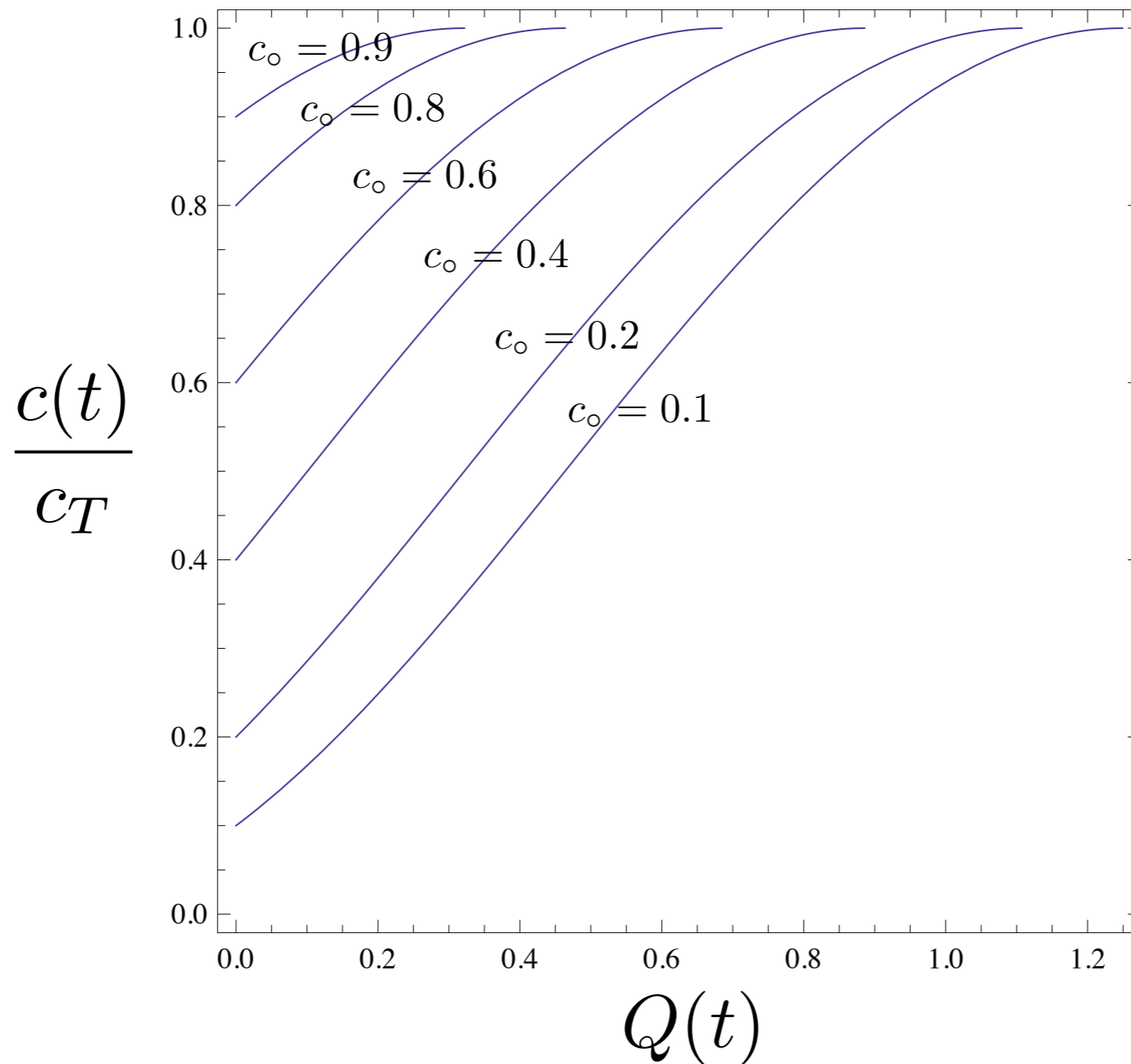


$$\frac{\partial c}{\partial t} = A(\eta, T) c^{1/2} (1 - c)^{1/2}$$



# Effect on the Battery

$$\frac{\partial c}{\partial t} = A(\eta, T)c^{1/2}(1-c)^{1/2}$$





# Tafel Kinetics

Butler-Volmer Kinetics:

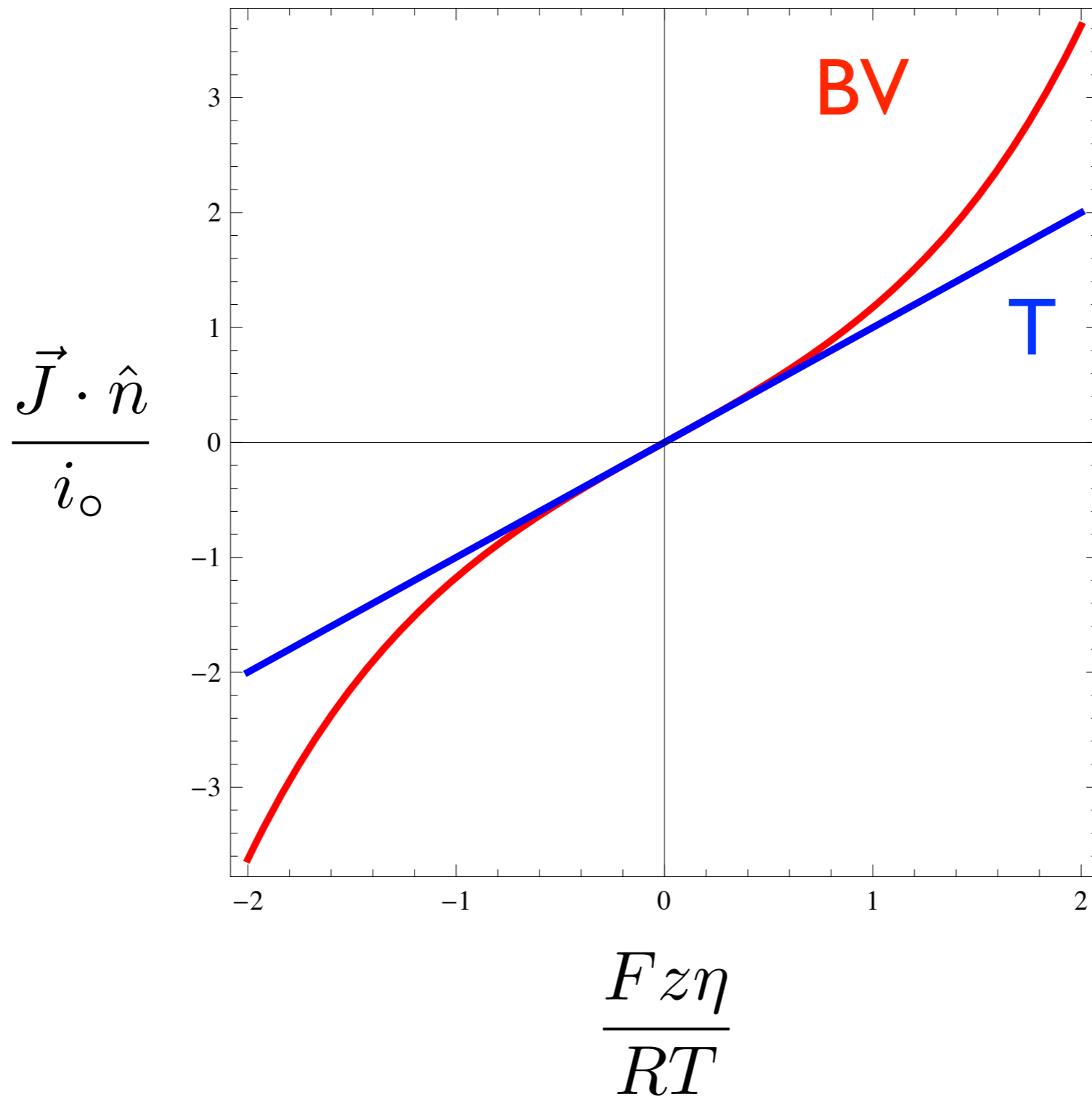
$$\vec{J} \cdot \hat{n} = i_o \left( \exp\left(\frac{\alpha_a F z \eta}{RT}\right) - \exp\left(-\frac{\alpha_c F z \eta}{RT}\right) \right)$$

$$i_o = F k_r (c_T - c_s)^{\alpha_a} c_s^{\alpha_c} c_e^{\alpha_a}$$

if  $\alpha_a = \alpha_c = 1/2$  and  $\frac{F z \eta}{RT} \ll 1$

$$\vec{J} \cdot \hat{n} = i_o \frac{F z \eta}{RT}$$

# Tafel Kinetics



# Exponent<sup>®</sup>

## Understanding Lithium-Ion Degradation and Failure Mechanisms by Cross-Section Analysis

Quinn Horn

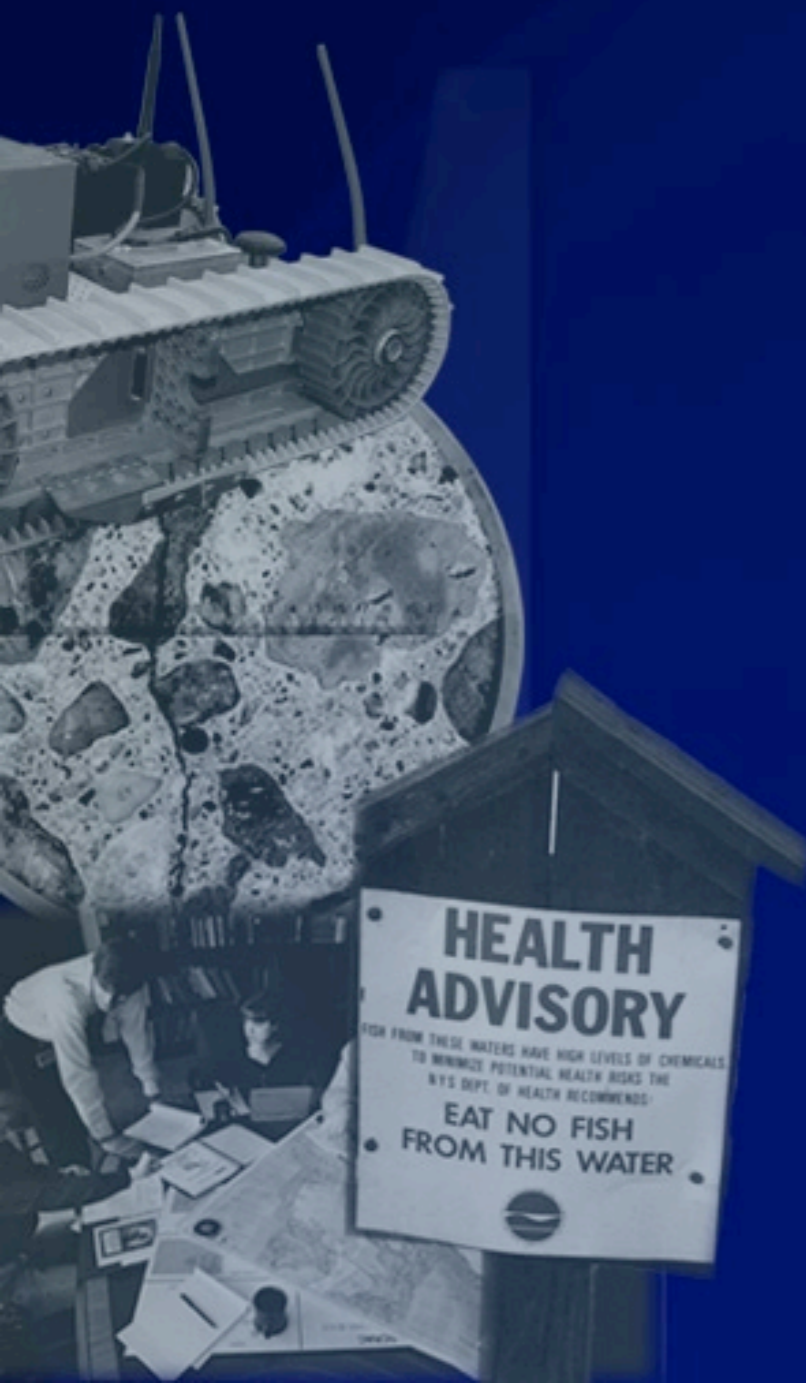
Kevin White

Exponent, Inc.

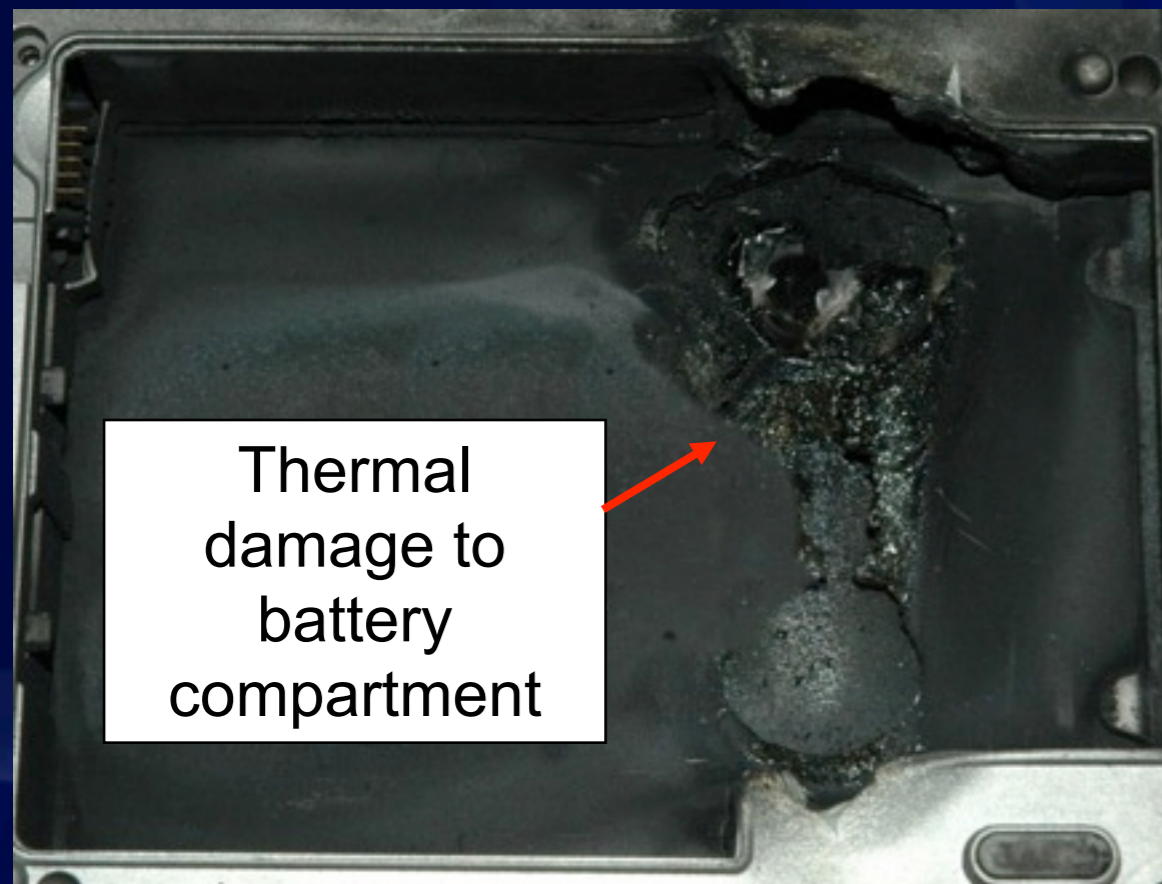
Natick, Massachusetts

211th Meeting of the Electrochemical Society

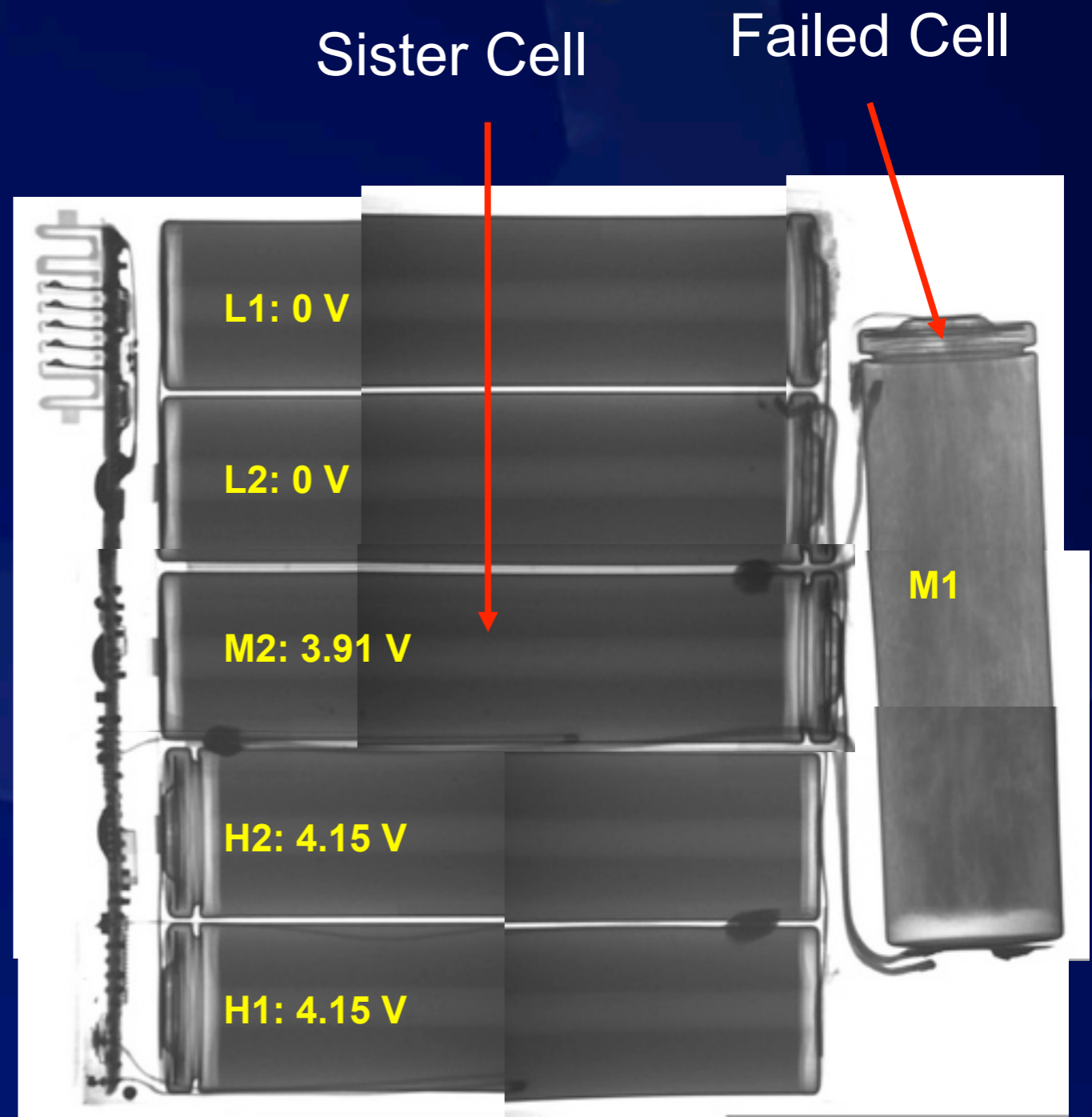
May 8, 2007



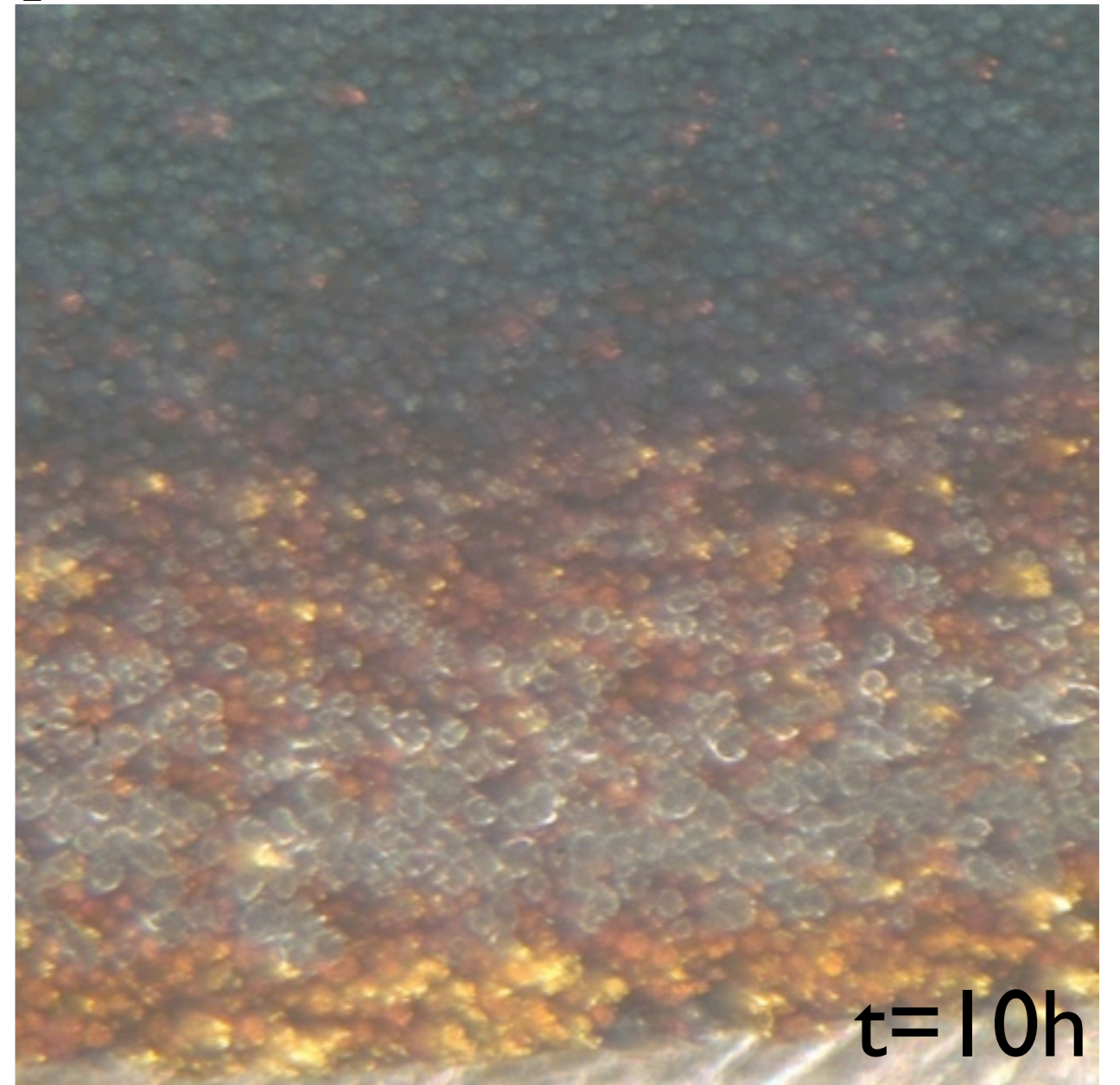
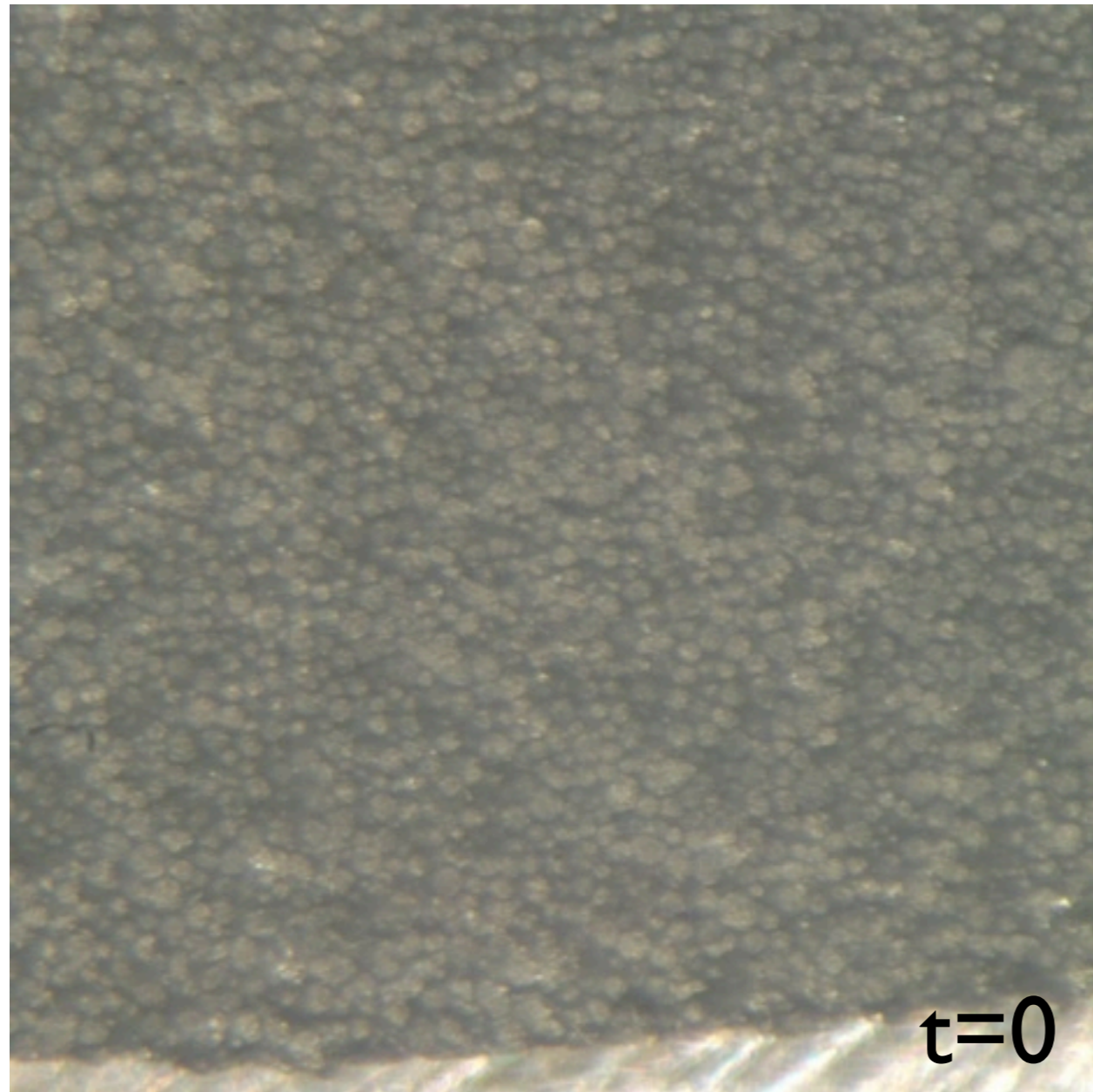
# Field Failure of a Notebook Battery Pack

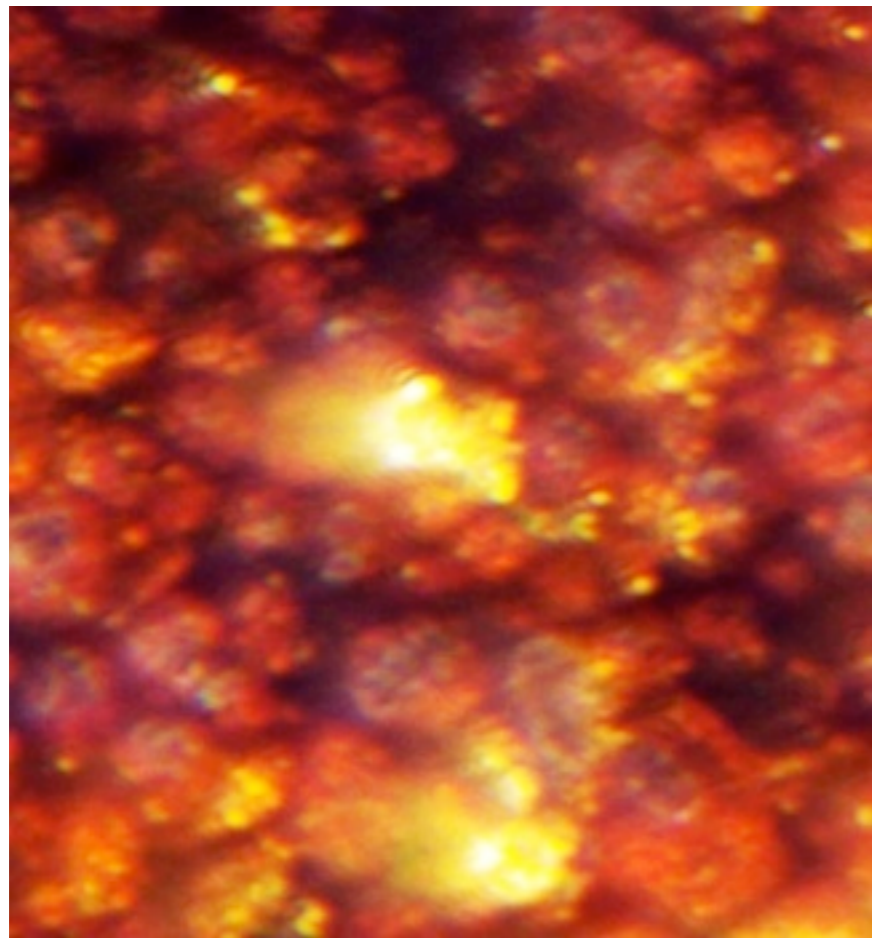


Cell M2 is the “sister cell” to the failed cell (i.e. M1 and M2 were in parallel and therefore experienced the same voltage conditions)

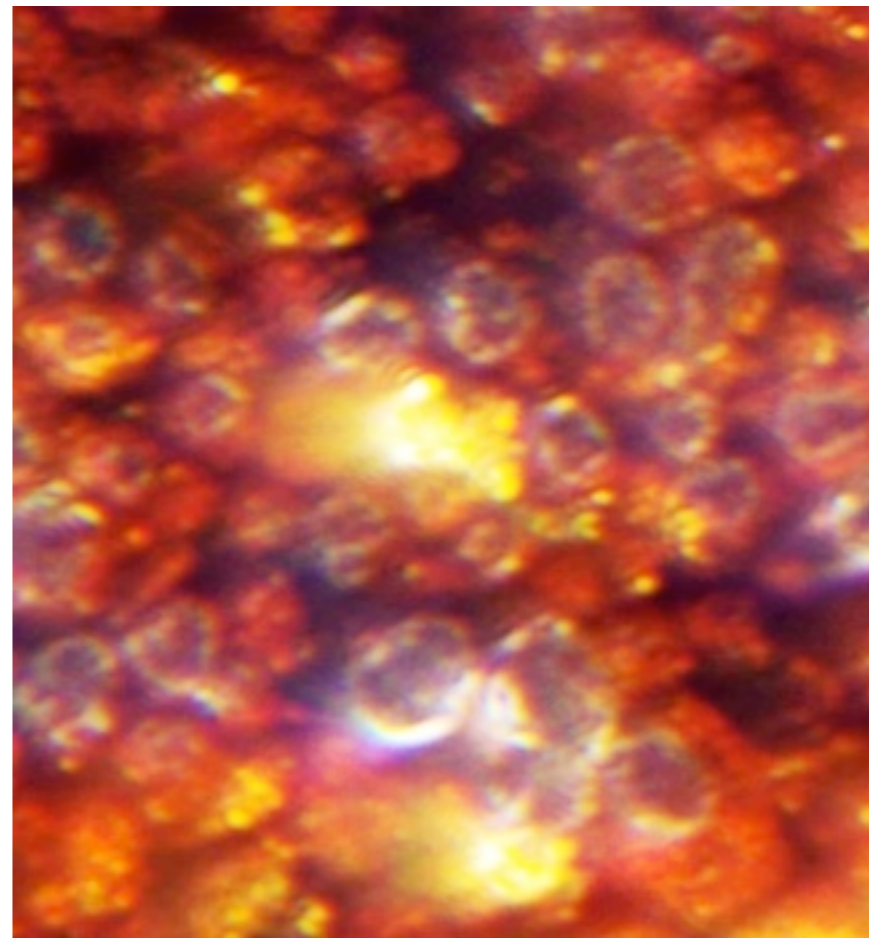


# In Situ Experiments

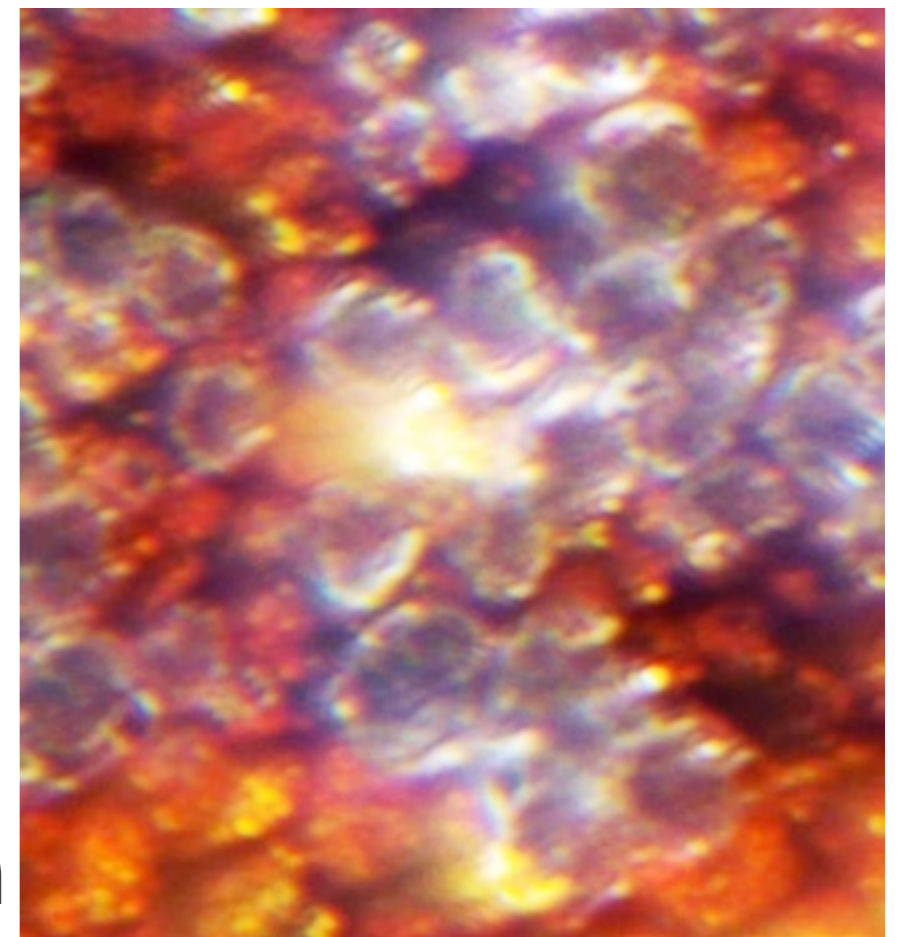




**incubation**

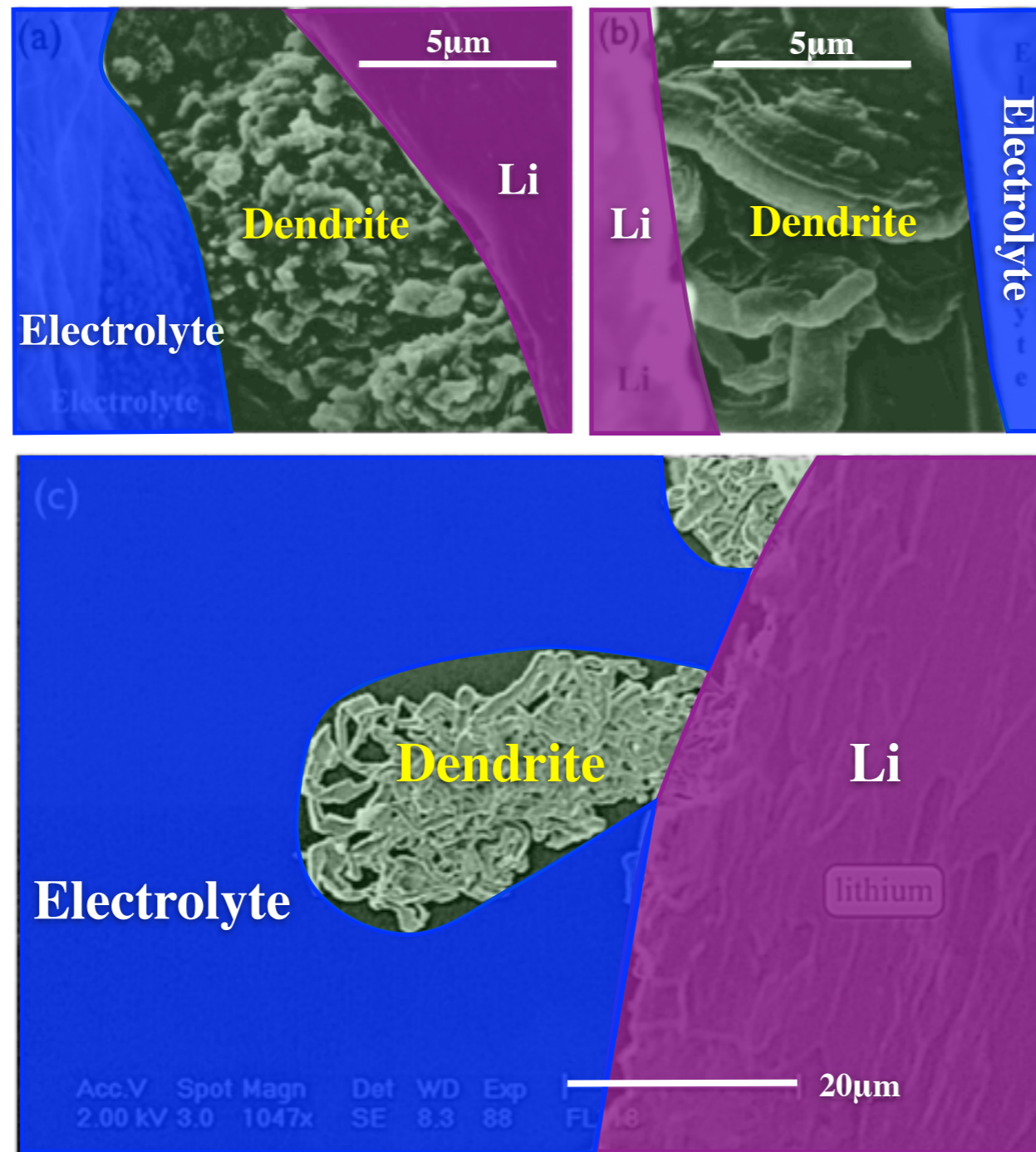


**nucleation**



**growth**

# Lithium Dendrites

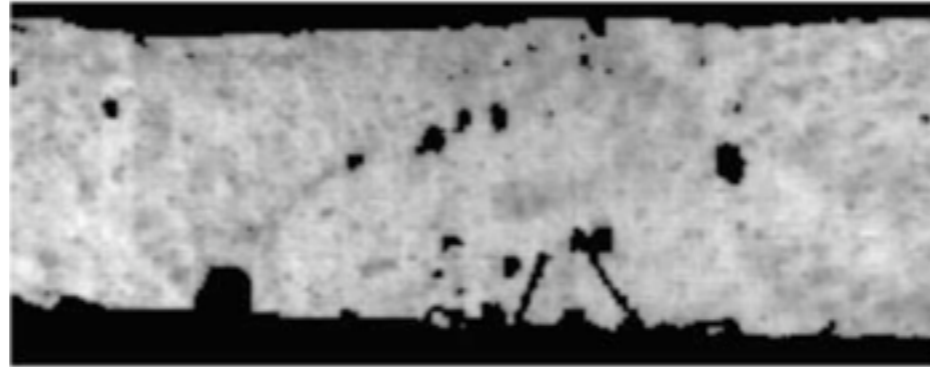


M. Dolle, L. Sannier, B. Beaudoin, M. Trentin, and J. M. Tarascon. Live Scanning Electron Microscope Observations of Dendritic Growth in Lithium/Polymer Cells. *Electrochemical and Solid-State Letters*, 5:A286-A289, 2002

F. Orsini, A. Du Pasquier, B. Beaudoin, J. M. Tarascon, M. Trentin, N. Langenhuisen, E. De Beer, and P. Notten. In Situ Scanning Electron Microscopy (SEM) Observation of Interfaces within Plastic Lithium Batteries. *Journal of Power Sources*, 76:19-29,1998

# Tip-Controlled Dendritic Growth

t=56 hours

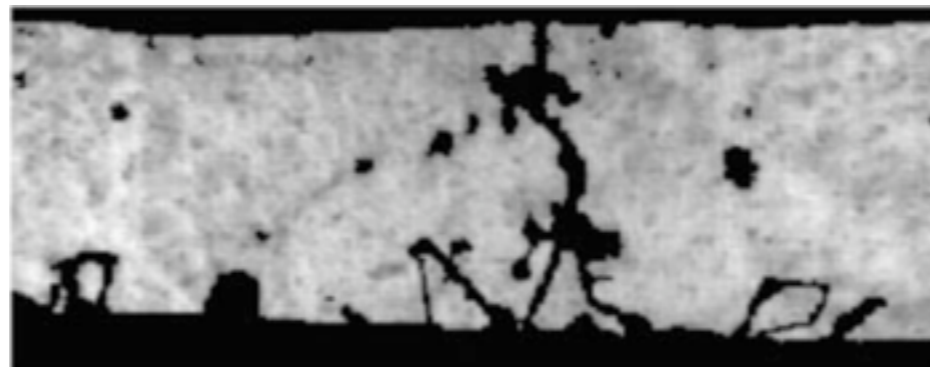


1.2mm

t=83 hours



t=100 hours

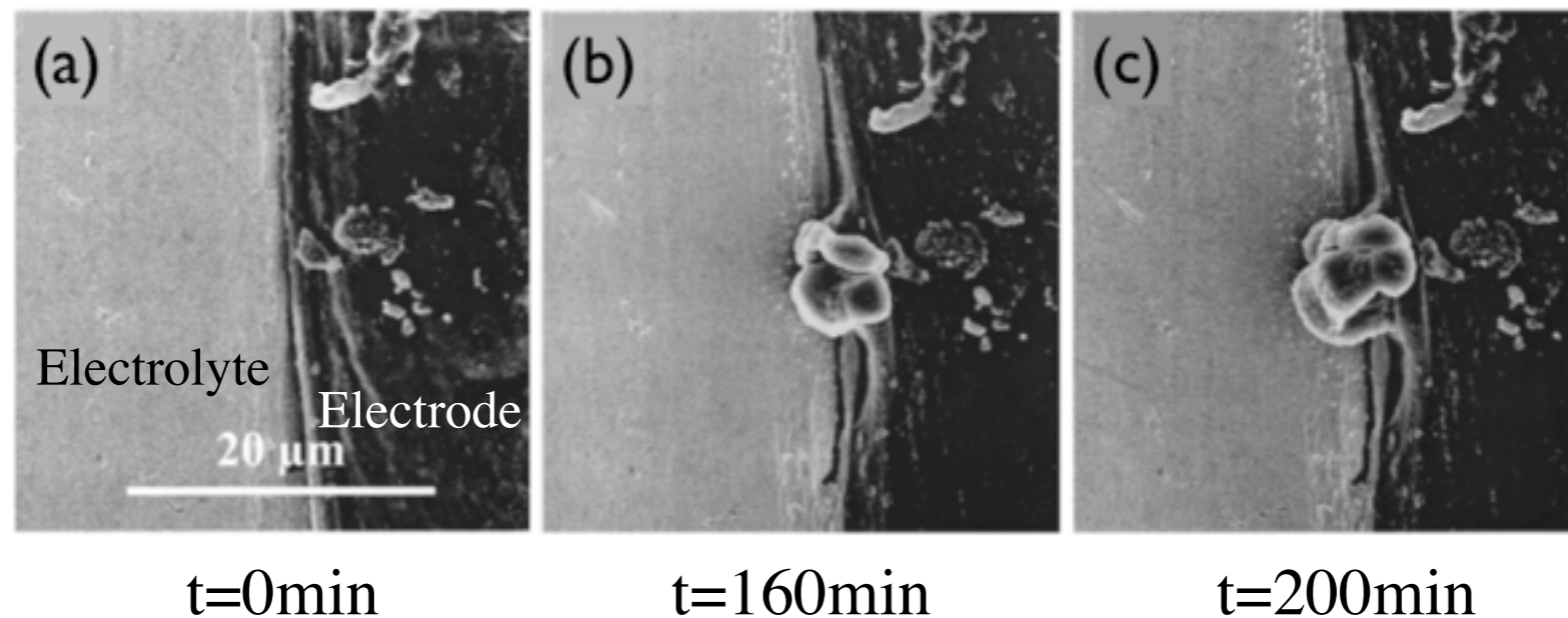
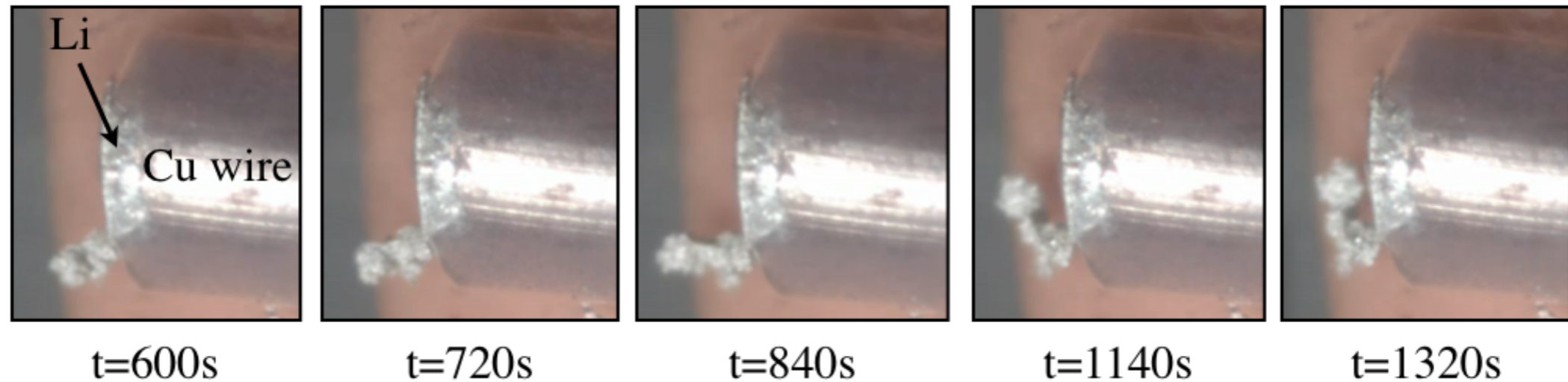


Slide courtesy of Zhiwen Liang

M. Rosso, C. Brissot, A. Teyssot, M. Dolle, L. Sannier, J. M. Tarascon, R. Bouchet, and S. Lascaud.  
Dendrite Short-circuit and Fuse Effect on Li/Polymer/Li Cells. *Electrochimica Acta*, 51:5334-5340, 2006



# Base-Controlled Dendritic Growth



O. Crowther and A. C. West. Effect of Electrolyte Composition on Lithium Dendrite Growth. *Journal of The Electrochemical Society*, 155(11):A806-A811, 2008

M. Dolle, L. Sannier, B. Beaudoin, M. Trentin, and J. M. Tarascon. Live Scanning Electron Microscope Observations of Dendritic Growth in Lithium/Polymer Cells. *Electrochemical and Solid-State Letters*, 5:A286-A289, 2002

# Dendrite Growth Mechanisms

- **Tip-Controlled:**
  - Dendrite grows at the tip.
  - Base remains the same.
- **Base-Controlled:**
  - Dendrite is extruded at the base.
  - Tip remains the same.