Online simulations via nanoHUB: Binding and electronic structure of Si

In this tutorial:

• Use density functional theory calculations to explore:

- How the electronic structure of Si develops
- Understand binding in the Si crystal

David Guzman, Sam Reeve, and Ale Strachan

strachan@purdue.edu School of Materials Engineering & Birck Nanotechnology Center Purdue University West Lafayette, Indiana USA



Launch the **DFTMatProp** tool in nanoHUB

From your *My HUB* page launch DFT calculations with Quantum Espresso

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Electronic structure and binding in Si

- DFT calculations using the DFTMatProp tool in nanoHUB
- Explain the formation of the band structure in Si starting from the simple electronic structure of a Si atom

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Step 1: Basic Input

- select "E-K Diagram" from task menu
- then "Semiconductor" from material type
- followed by "Si(Diamond)" in semiconductors.
- Finally switch to "yes" the advanced options



Electronic structure of Si perfect crystal

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Band structure will be shown along the following path (tat contains the valence band maxima and conduction band minima)

$$K_{path} = L - \Gamma - X - \Gamma$$



Explore the electronic band structure of Si

Assignment

- Identify the valence band maximum and conduction band minima
- Is Si direct or indirect band gap semiconductor?
- (Note that DFT tends to underestimate band gaps)



Electronic structure of an isolated Si atom

Analyze the band structure of the 2 isolated silicon atoms

NOTICE: No dispersion, i.e. no k-dependence

- Non-interacting atoms
- k determines how WF varies from unit cell to unit cell does not affect the energy as atoms are far way

Now let us look at the electronic properties of silicon molecule. In the following we will start with a guess for the separation between Si atoms, compute the GGA predicted bonding distance, and compute the electronic structure of such system. To get started launch a new DFTMAtProp window.

Step 1: Basic Input

- select "E-K Diagram" from task menu
- then "Semiconductor" from material type
- followed by "Si(Diamond)" in semiconductors.
- Finally switch to "yes" the advanced options

Step 2: Geometric input

Change the atomic structure and cell vectors fields as follows:

Basic Input Geometric Input Energy Expression					
	Atomic Structure:	si 0.4 0.5 0.5 si 0.6 0.5 0.5			
	Cell Vectors (A):				
		0.00 0.00 15.0			
Equation of State Calculations: 🔵 📰 🔄 no					

Si₂ molecule:

To model a molecule in a code that implements DFT based on plane waves and periodic boundaries, we need to increase the size of the simulation box to avoid interactions with adjacent images

The molecule is located in the middle of the simulation box and the atoms are initially separated by 3Å

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SCF Convergence Criterion (Ry):	1E-6			
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Step 4: CLICK SIMULATE

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From single atom and molecule to crystal

<section-header>

- To better understand the band structure of Si we will perform a series of simulations with varying lattice parameters
- From a large lattice parameter with isolated atoms to the equilibrium lattice parameter of the crystal

Electronic structure vs. lattice parameter

Geometric input

Switch the EOS calculation to YES and change minimum volume to 1, maximum volume to 20, and steps to 20

Remember to set the lattice parameters back to the original value or re-start the tool

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Electronic structure vs. lattice parameter

The simulation will take several minutes since it will run 20 different calculations at different unit cell volumes.

Large lattice parameter (V~6.5V₀) weak interactions

- s and p bands start to overlap in energy
- s/p splitting similar to bonding/antibonding splitting
- No gap

A band gap opens up

Inverse separation between Si atoms

By-product: equation of state

- The energy-volume relationship obtained is called equation of state
- From it, we can obtain the equilibrium volume (minimum energy) and bulk modulus of the system (at T=0K in this case)

15 points form $0.8V_0$ to $2V_0$

Additional Questions

- Consider the Si₂ molecule. For each energy level in the band diagram plot, indicate its multiplicity (how many states have the same energy) and their electron occupation indicating electrons and their spin with arrows.
- At what lattice parameter does a band gap separating occupied and unoccupied states form?
- Think about the development of the band structure in terms of the s and p picture in the figure below

Additional notes: the diamond structure

Silicon crystalizes in the cubic Fd-3m space group every Si atom is bonded to four equivalent Si atoms forming corner-sharing tetrahedra Conventional cell

 $\vec{a} = a_0(1, 0, 0)$ $\vec{b} = a_0(0, 1, 0)$ $\vec{c} = a_0(0, 0, 1)$ $Si_1: 0.00, 0.00, 0.00$ $Si_2: 0.25, 0.75 0.75$ $Si_3: 0.50, 0.00 0.50$ $Si_4: 0.00, 0.50 0.50$ $Si_5: 0.50, 0.50 0.00$ $Si_6: 0.75, 0.25 0.75$ $Si_7: 0.75, 0.75 0.25$ $Si_8: 0.25, 0.25 0.25$

Primitive cell

 $\vec{a} = a_0(0, \frac{1}{2}, \frac{1}{2})$ $\vec{b} = a_0(\frac{1}{2}, 0, \frac{1}{2})$ $\vec{c} = a_0(\frac{1}{2}, \frac{1}{2}, 0)$

 $Si_1: 0.00, 0.00, 0.00$ $Si_2: 0.25, 0.25, 0.25$

Additional notes: exchange and correlation

Energy Expression You may explore the same results with LDA – by changing the XC functional

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SCF Convergence Criterion (Ry): 1E-6								
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Additional notes: reciprocal space

Band Structure/DOS Leave the defaults in all fields.

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We selected a short K-path to plot the electron band dispersion which contains the valence band maxima and conduction band minima of Si diamond electronic structure

 $K_{path} = L - \Gamma - X - \Gamma$

