

Appendix B

# NanoTreatment Cards



*Note:* Print these cards on cardstock and laminate them for repeated use.

## Water Quality Problem 1

Cleaning water for drinking, cooking, and bathing in a first-world country



What contaminants affect this water?



Bacteria



Pesticides

What other factors do you need to consider?

- Literate population
- Expects clear water = clean water

## Water Quality Problem 2

Cleaning water for cooking and drinking in a third-world country



What contaminants affect this water?



Bacteria



Viruses



Cyanide

What other factors do you need to consider?

- Illiterate population
- Little to no infrastructure to support large-scale water treatment

## Water Quality Problem 3

Cleaning water for irrigation in a second world country



What contaminants affect this water?



Cadmium



Lead



Pesticides

What other factors do you need to consider?

- Semi-literate population
- Intermittent power supply

## Water Quality Problem 4

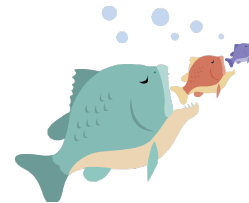
Cleaning water for industrial use in soft drinks



What contaminants affect this water?



Cadmium



Mercury

What other factors do you need to consider?

- Lots of existing systems
- Removal of taste and odor is key

## Water Quality Problem 5

Cleaning water for vaccines



What contaminants affect this water?



Viruses



Bacteria

What other factors do you need to consider?

- Lots of existing systems
- No shortage of energy
- Removal of all contaminants is key

## Water Quality Problem 6

Desalinating ocean water to use as drinking water



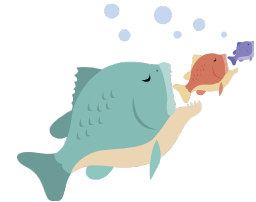
What contaminants affect this water?



Salt



Bacteria



Mercury

What other factors do you need to consider?

- Requires a lot of energy
- We typically extract salt out of the water – is there a way to extract only the water?

## Water Quality Problem 7

Treating water for livestock to drink



What contaminants affect this water?



Pesticides



Bacteria

What other factors do you need to consider?

- Electrostatic interactions are sensitive to pH and salt concentrations

## Water Quality Problem 8

Reclaiming water contaminated by industrial use



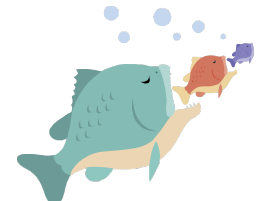
What contaminants affect this water?



Lead



Cadmium



Mercury

What other factors do you need to consider?

- Lots of existing systems

## Water Contaminant B

Coliform bacteria



**Source:** Soils, plants, animals, human sewage, agricultural and storm runoff.

**Hazards:** Most coliform bacteria do not cause disease, but are an easy way to monitor for more dangerous organisms that cause diarrhea.

**Size:** 500–2000 nm

**Dominant force(s):** Gravity, Electrostatic (influenced weakly by charge)

## Water Contaminant V

Viruses



**Source:** Fecal matter, animal urine

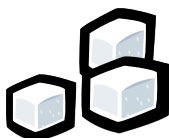
**Hazards:** Hepatitis, meningitis, colds, gastroenteritis.

**Size:** 27–35 nm

**Dominant force(s):** Electrostatic (influenced by charge) electrostatic interactions, which are sensitive to pH and salt concentrations

## Water Contaminant S

Salts



**Source:** Contact with the ground and oceans in the water cycle

**Hazards:** Drinking salt water causes dehydration, nausea, weakness, and possible death.

**Size:** 0.09–0.206 nm

**Dominant force(s):** Electrostatic, (influenced strongly by charge)

## Water Contaminant P

Pesticides



**Source:** Runoff from pesticide application, water treatment plants (chlorine), pesticide disposal.

**Hazards:** Effects depend on type, amount, and duration of exposure, but include cancer, nerve damage, and hormone disruption.

**Size:** 0.240–0.750 nm

**Dominant force(s):** Electrostatic (influenced by charge), larger ones are influenced by gravity

## Water Contaminant L

Lead  
( $\text{Pb}^{++}$  or  $\text{Pb}^+$ )



**Source:** Lead pipes and solder that is used to connect pipes. Leaded gasoline.

**Hazards:** Mental and physical development delays in children; high blood pressure and kidney problems in adults.

**Size:** 0.079–0.143 nm

**Dominant force(s):** Electrostatic (influenced heavily by charge)

## Water Contaminant C

Cadmium  
( $\text{Cd}^{++}$ )



**Source:** Corrosion of galvanized pipes; erosion of natural deposits; discharge from metal refineries; runoff from waste batteries and paints.

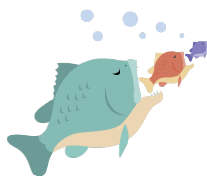
**Hazards:** Kidney damage

**Size:** 0.092–0.124 nm

**Dominant force(s):** Electrostatic (influenced heavily by charge)

## Water Contaminant M

Methylmercury



**Source:** Industrial processing of coal and metals contributes mercury to aquatic systems where it complexes to form methylmercury. It is biomagnified in these food chains.

**Hazards:** Impaired neurological development in children (for example, thinking, memory, attention, language, and fine motor and visual spatial skills)

**Size:** 0.110–0.133 nm

**Dominant force(s):** Electrostatic (influenced by charge)

## Water Contaminant Y

Cyanide ( $\text{CN}^-$ )



**Source:** Industrial contamination from production of paper, textiles, plastics

**Hazards:** Lowers B12 levels, prevents cells from using oxygen, disrupts thyroid and nervous system function.

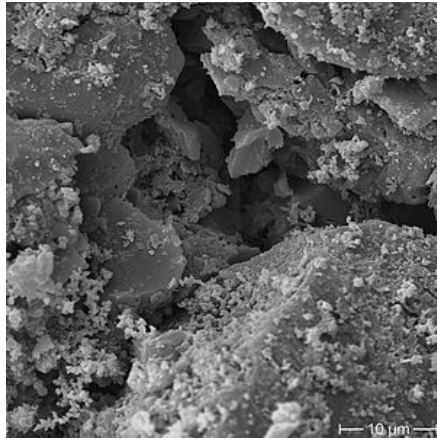
**Size:** 0.090 nm

**Dominant force(s):** Electrostatic (influenced heavily by charge)

## Nanotechnology: Activated Carbon / Activated Charcoal

### Potential

- Effective adsorbent of a variety of non-polar substances—typically taste and odor.



### Risks

- Non-toxic
- Does not affect metals, Cyanide, or alcohol; destroyed by high or low pH

**Subject to forces:** Electrostatic (influenced weakly by charge)

**Size:** 2–50 nm

**Image:** Activated Charcoal through a Scanning Electron Microscope

**Credit:** [http://upload.wikimedia.org/wikipedia/commons/thumb/6/65/Activated\\_Charcoal.jpg/300px-Activated\\_Charcoal.jpg](http://upload.wikimedia.org/wikipedia/commons/thumb/6/65/Activated_Charcoal.jpg/300px-Activated_Charcoal.jpg)

## Nanotechnology: Solar Water Disinfection

### Potential

- Sunlight through clear water for 6+ hours will treat most water for bacterial pathogens.
- This process is sped up by adding a catalyst such as silver and titanium dioxide nanoparticles.



### Risks

- Creation of free radicals and silver chloride in the body
- Cellular death
- Accumulation in the liver
- Smaller particles are more reactive than larger ones

**Subject to forces:** Electrostatic (influenced heavily by charge)

**Size:** 1–100 nm

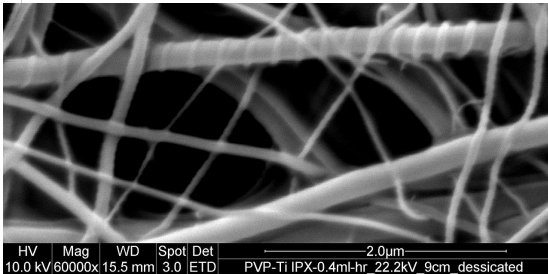
**Image:** A Solar Water Disinfection table

**Credit:** <http://upload.wikimedia.org/wikipedia/commons/thumb/6/67/Indonesia-sodis-gross.jpg/320px-Indonesia-sodis-gross.jpg>

## Nanotechnology: Nanobiocides

### Potential

- Nanobiocides such as metal nanoparticles and engineered nanomaterials are successfully incorporated into nanofibers showing high antimicrobial activity and stability in water.
- Titanium dioxide nanoparticles break down common organic contaminants such as hormones, pharmaceuticals, or manure—when exposed to light
- Nano-sized Silver is used to kill bacteria.



**Credit:** [http://upload.wikimedia.org/wikipedia/commons/thumb/2/25/Titanium\\_dioxide\\_nanofiber\\_spiral.jpg/220px-Titanium\\_dioxide\\_nanofiber\\_spiral.jpg](http://upload.wikimedia.org/wikipedia/commons/thumb/2/25/Titanium_dioxide_nanofiber_spiral.jpg/220px-Titanium_dioxide_nanofiber_spiral.jpg)

### Risks

- Creation of free radicals and silver chloride in the body
- Cellular death
- Accumulation in the liver
- Smaller particles are more reactive than larger ones

**Subject to forces:** Electrostatic (influenced heavily by charge)

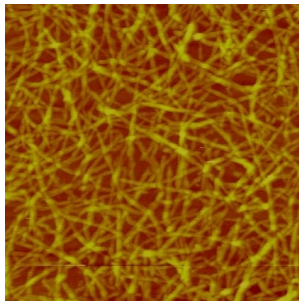
**Size:** 1 to 100 nm

**Image:** Titanium Dioxide nanofiber spiral

## Nanotechnology: Nano-cellulose filter

### Potential:

- From plants or bacteria
- Transparent, electrically conductive, stronger than steel
- Nanocellulose can filter out nitrates in city water, paint dye residues in printing and textiles, and metal ions from industrial waste. It has also been shown to remove virus particles.



[https://en.wikipedia.org/wiki/Nanocellulose#/media/File:AFM\\_Inventia\\_nanocellulose.JPG](https://en.wikipedia.org/wiki/Nanocellulose#/media/File:AFM_Inventia_nanocellulose.JPG)

### Risks:

- Biodegradable
- Inert
- Non-toxic
- Stable in wide range of pH

**Subject to forces:**

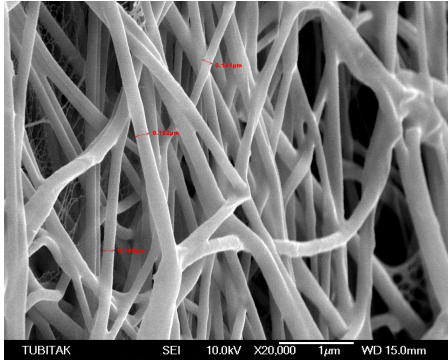
**Size:** 5-20 nm x 10 nm to several micrometers

**Image:** Carboxymethylated nanocellulose

## Nanotechnology: Electrospun Nanofibers

### Potential

- Useful in filtration—these fibers can be built into membranes with a high surface area. Also, the material being spun can be manipulated to give control over pore size and modification of the surface chemistry.



Credit: [http://upload.wikimedia.org/wikipedia/commons/thumb/d/db/N2\\_2.kesit.JPG/220px-N2\\_2.kesit.JPG](http://upload.wikimedia.org/wikipedia/commons/thumb/d/db/N2_2.kesit.JPG/220px-N2_2.kesit.JPG)

### Risks

- May carry similar risks to manufacturers as asbestos fibers: cancer

**Subject to forces:** Varies depending on composition.

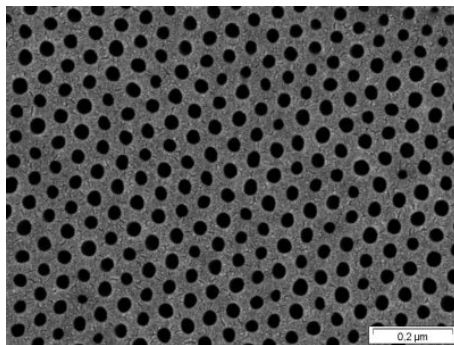
**Size:** 10–1,000 nm

**Image:** Electrospun Nanofibers

## Nanotechnology: Nanoporous membrane

### Potential:

These self-assembled membranes are used to desalinate saltwater, and remove molecules such as micro-pollutants and pharmaceutical residues from water.



[www.nanowerk.com/news2/newsid=29282.php](http://www.nanowerk.com/news2/newsid=29282.php)

### Risks:

- 

**Subject to forces:**

**Size:** 3 - 15 nm

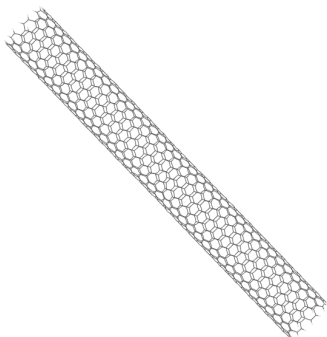
**Image:** Nanopore membrane



## Nanotechnology: Carbon Nanotubes

### Potential

- Water can travel through the tube, while other materials cannot.



### Risks

- Can cross the cell membrane.
- May interfere with or damage DNA
- May carry similar risks as asbestos fibers: cancer

**Subject to forces:** Electrostatic (variably influenced by charge)

**Size:** 1.0 nm diameter

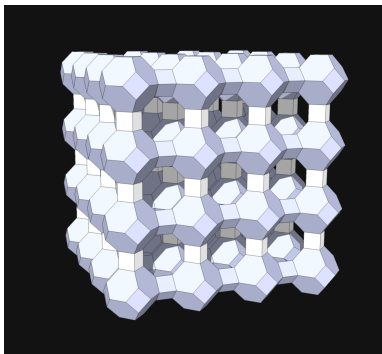
**Image:** Simulated Carbon NanoTube

**Credit:** Source=Wikipedia commons |Date=03 03 2007 |Author=[User:Gmdm](#) |

## Nanotechnology: Zeolyte

### Potential

- Utilizes the existence of nanoscopic pores in zeolite to act as a molecular sieve
- Exchanges  $\text{Na}^+$  for other ions, such as  $\text{K}^+$ ,  $\text{Ca}^+$ ,  $\text{Mg}^{2+}$ ,  $\text{Cu}^{2+}$



### Risks

- Naturally occurring substance.
- None known when used in processing water. Not recommended to take internally.

**Subject to forces:** Electrostatic (influenced heavily by charge)

**Size:** 0.003–0.010 nm

**Image:** Zeolite

**Credit:** [http://commons.wikimedia.org/wiki/Category:Zeolite#mediaviewer/File:Struttura\\_molecolare\\_di\\_una\\_zeolite.png](http://commons.wikimedia.org/wiki/Category:Zeolite#mediaviewer/File:Struttura_molecolare_di_una_zeolite.png)