

Thermoelectric Energy Conversion: Science and Engineering Challenges and Opportun

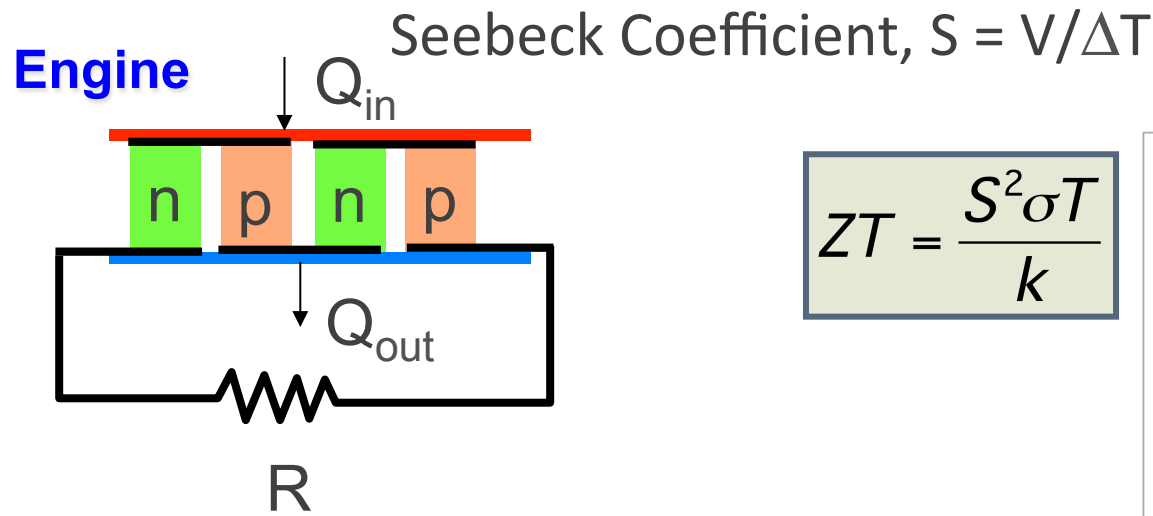
Arun Majumdar

Departments of Mechanical Engineering &

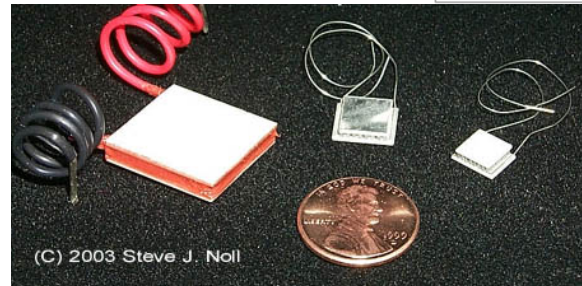
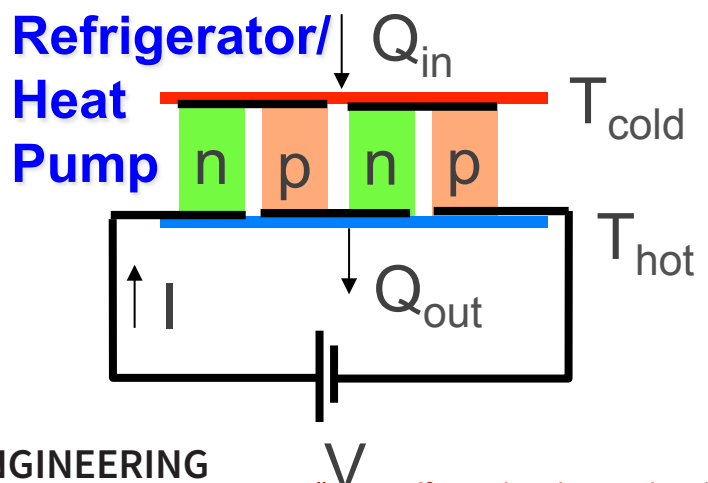
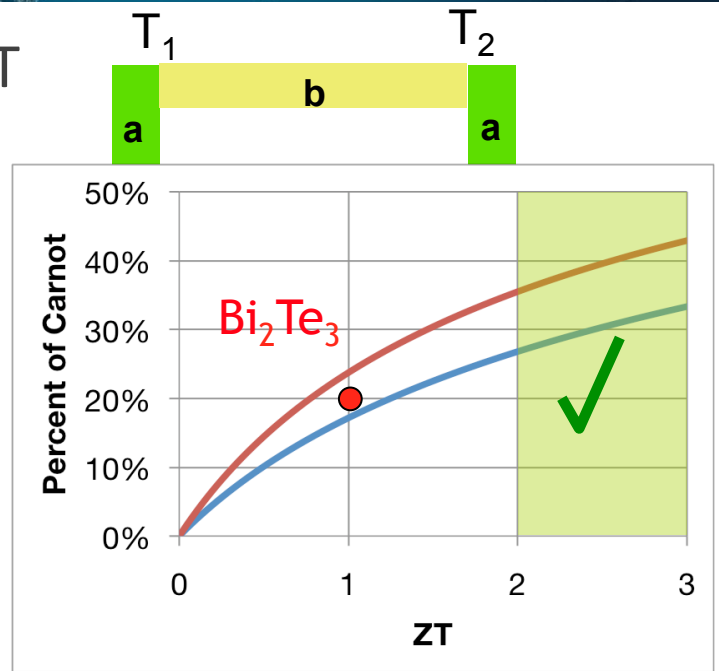
Materials Science & Engineering (courtesy), Stanford University

Department of Photon Science, SLAC

Thermoelectricity & Energy Conversion

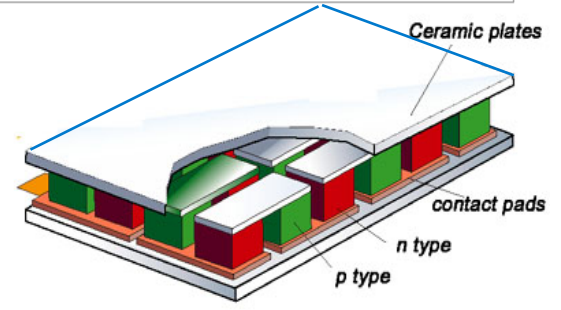


$$ZT = \frac{S^2 \sigma T}{k}$$



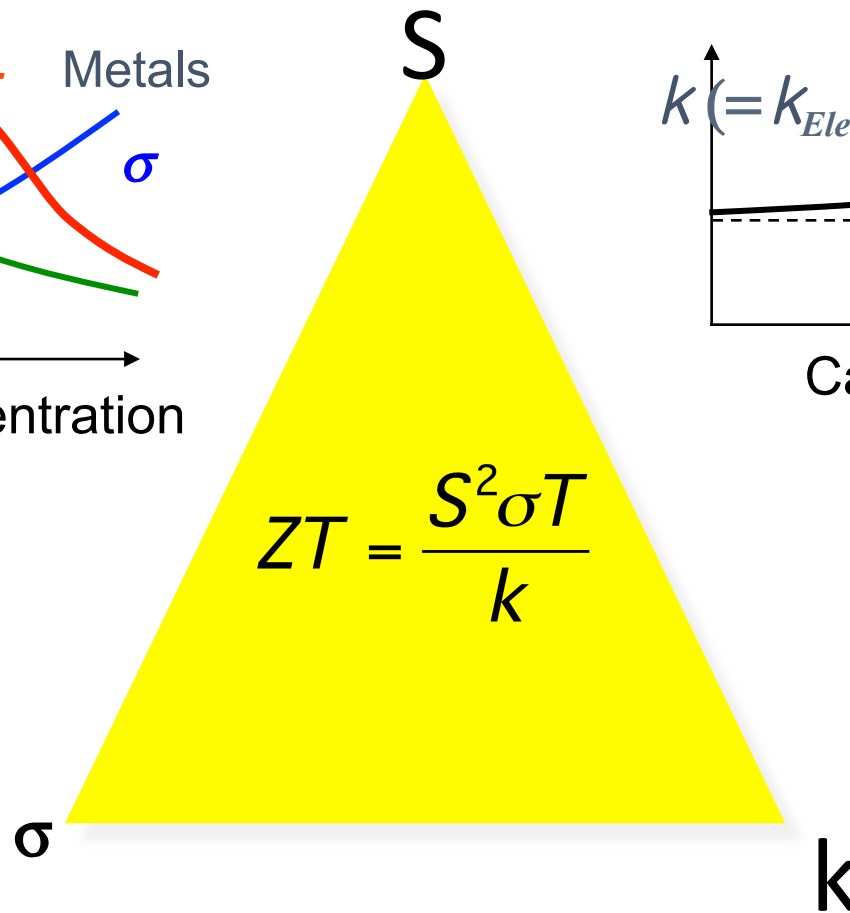
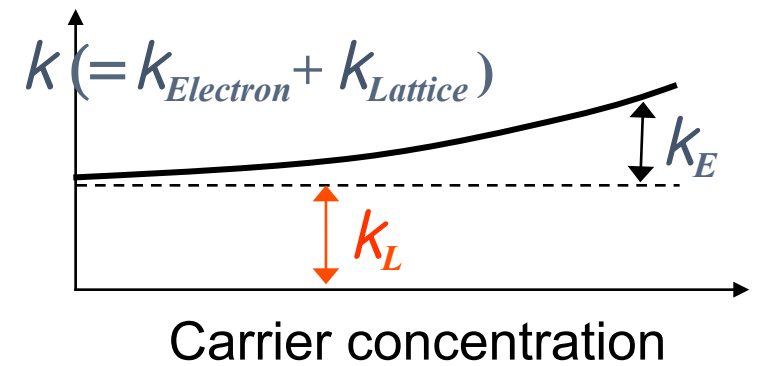
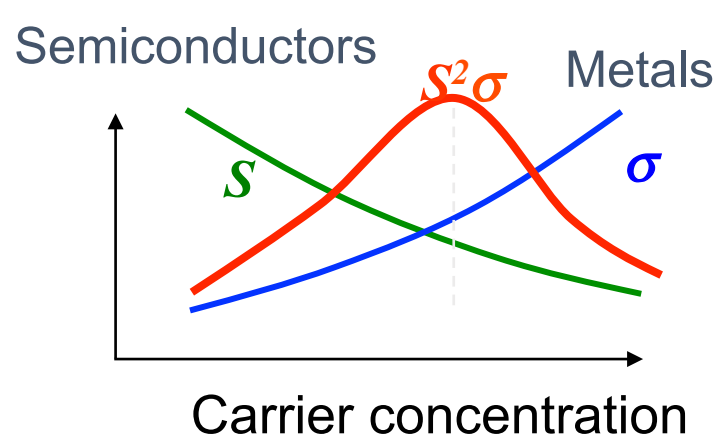
Bismuth Telluride

(low efficiency, low abundance, high cost)



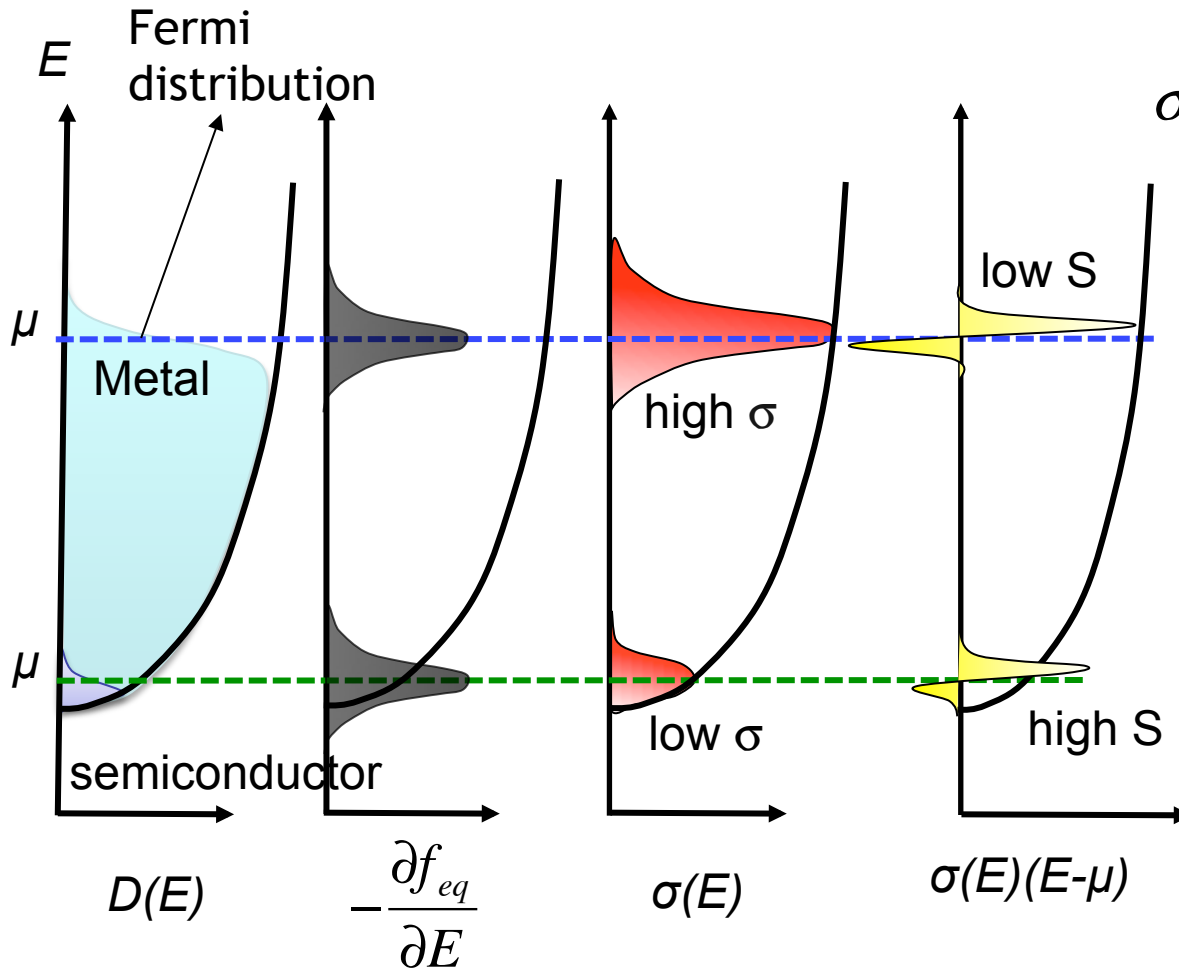
"Any sufficiently advanced technology is indistinguishable from magic." Arthur C. Clarke

Bermuda Triangle of Thermoelectrics





Band Picture of Thermoelectric Parameters



Electrical Conductivity, σ

$$\sigma = \int \sigma(E) dE$$

$$\sigma(E) = q^2 D(E) \tau(E) v(E)^2 \left(-\frac{\partial f_{eq}}{\partial E} \right)$$

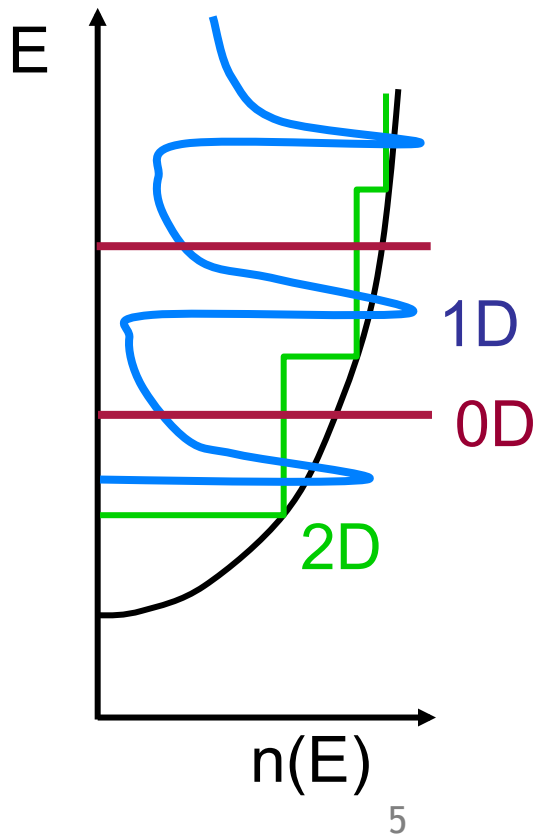
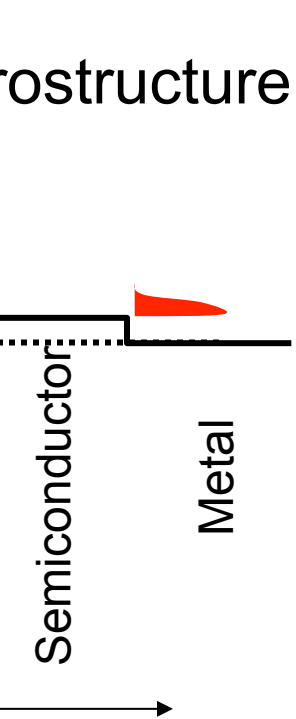
Thermopower, S

$$S = \frac{1}{eT} \frac{\int \sigma(E)(E-\mu) dE}{\int \sigma(E) dE}$$

$$S \sim m^* \sim \frac{1}{\partial^2 E / \partial k^2}$$



How do we increase $S^2\sigma$



Quantum
Confinement

(Hicks & Dresselhaus, 1993) Thermoelectrics Program Review
August 2009

How do we reduce k

$$k = Cvl/3$$

$C = 2 \times 10^6 \text{ J/m}^3\text{-K}$ - Dulong-Petit

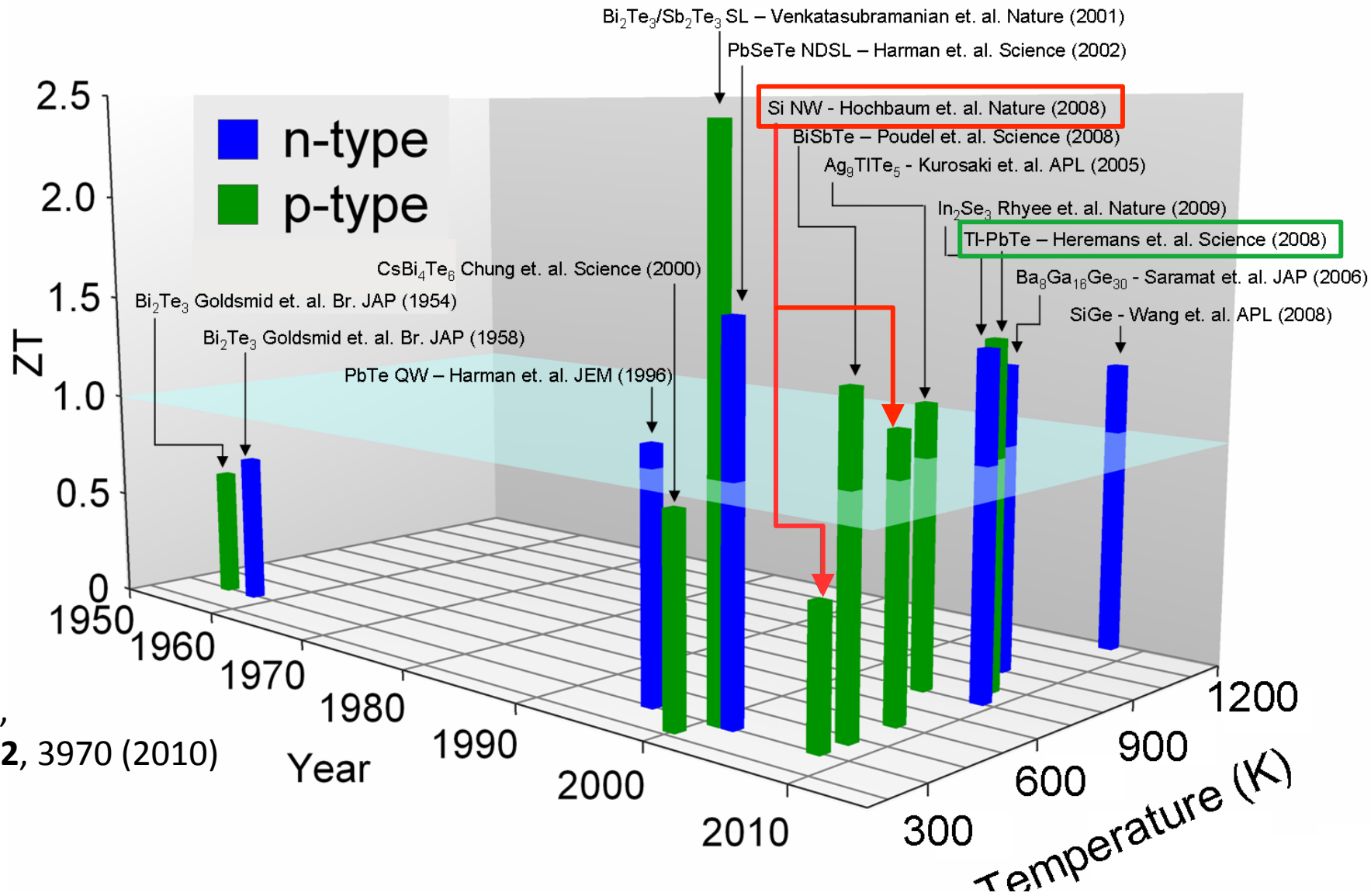
$v = 2000\text{-}4000 \text{ m/s}$

$l = 0.2\text{-}0.5 \text{ nm}$ (atomic scale)

$k_{\min} = 0.25\text{-}1 \text{ W/m-K}$

Vineis, Shakouri, Majumdar,
Kanatzidis, *Adv. Materials* **22**, 3970 (2010)

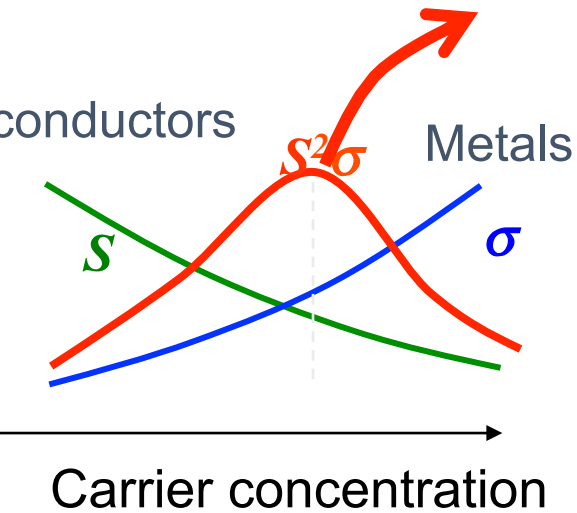
ZT Map



nkouri, Majumdar,
Adv. Materials **22**, 3970 (2010)

Viabile Options

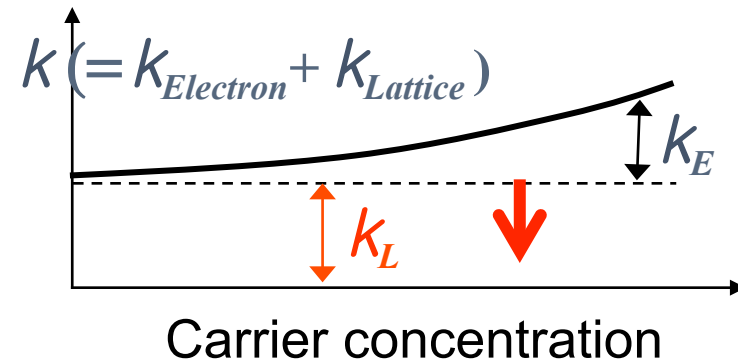
Materials with large DOS
 $k_B T$ s from E_F



Degenerate Semiconductors
 ultra-low k (e.g., localization)

S

$$ZT = \frac{S^2 \sigma T}{k}$$



$$k_{min} < 1 \text{ W/m-K}$$

k

USA
 1996
 Sciences

is part of a special series of Inaugural Articles by members of the National Academy of Sciences
 1995.

thermoelectric

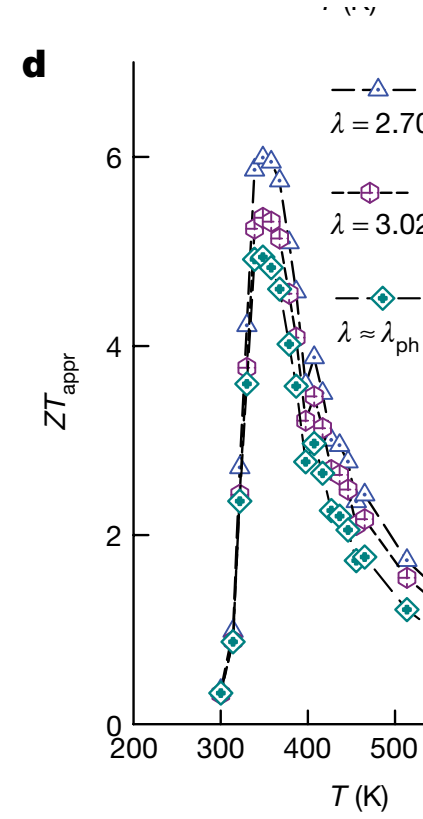
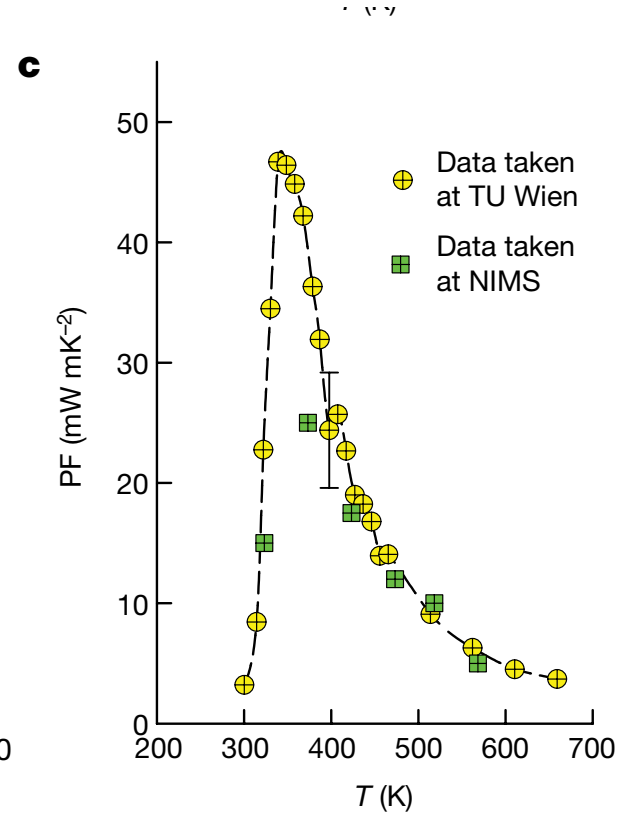
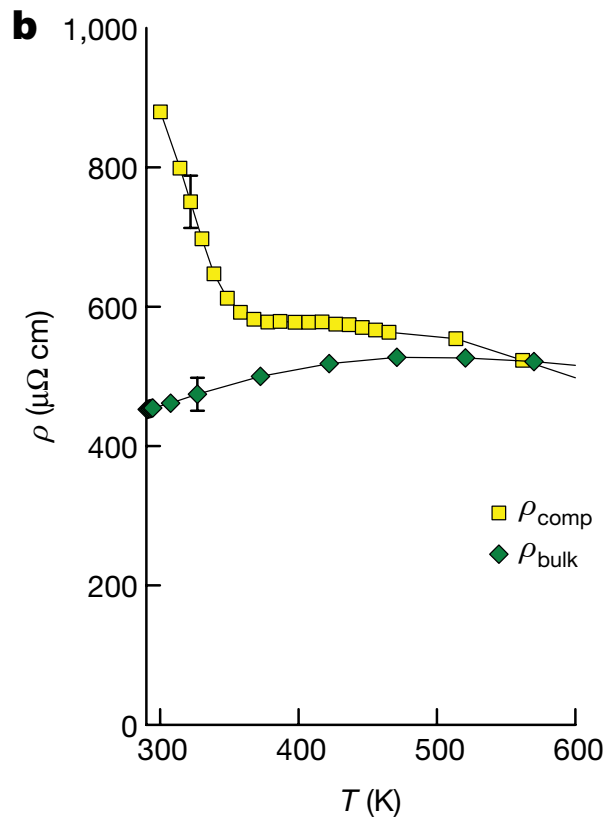
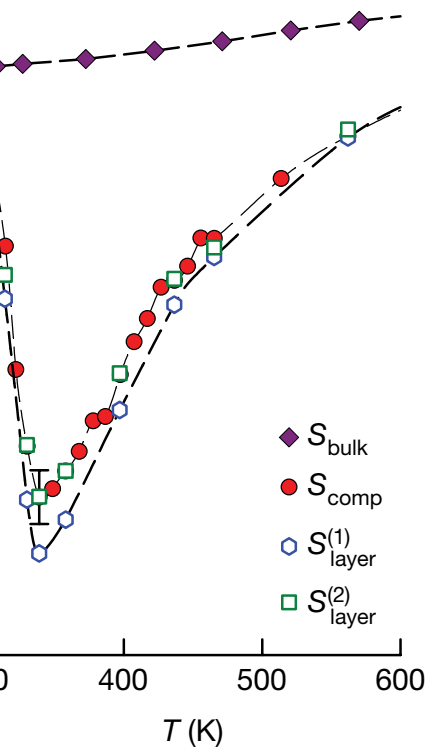
AND J. O. SOFO[‡]

and Astronomy, The University of Tennessee, Knoxville, TN 37996-1200; [†]Solid State Division, Oak Ridge National Laboratory, P.O. Box
 7831-6030; and [‡]Instituto Balseiro, Centro Atomico Bariloche, (8400) Bariloche, Argentina

istinguishable from magic." Arthur C. Clarke

Full-Heusler Alloy

Group



Hinterleitner et al., Nature (2019)



Is there a killer application for thermoelectrics?

Power Generation from Waste Heat Recovery (100 °C – 800 °C)

Refrigeration & Heat Pumps (10-60 °C)

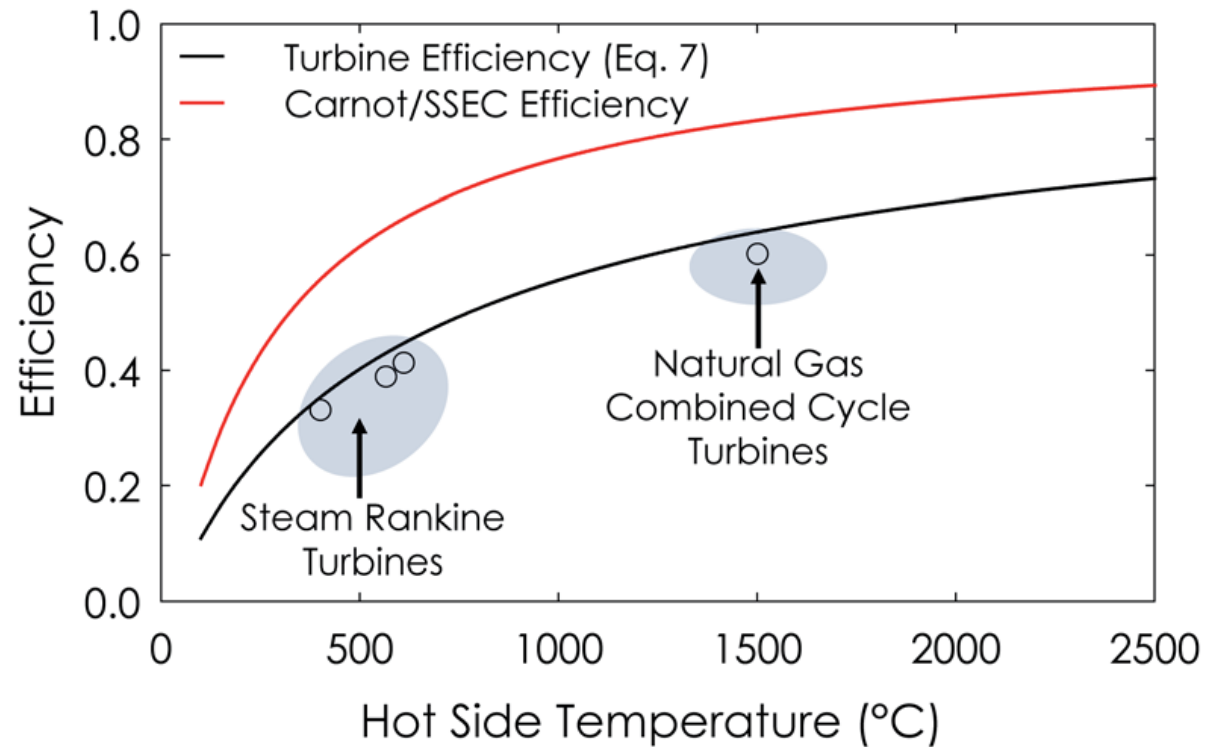
Efficiency of Current Heat Engines

Group

$$dW_{\max} = dH - T_{\text{amb}}dS$$

$$C_P dT \quad dS = \frac{dQ}{T} = \frac{C_P dT}{T}$$

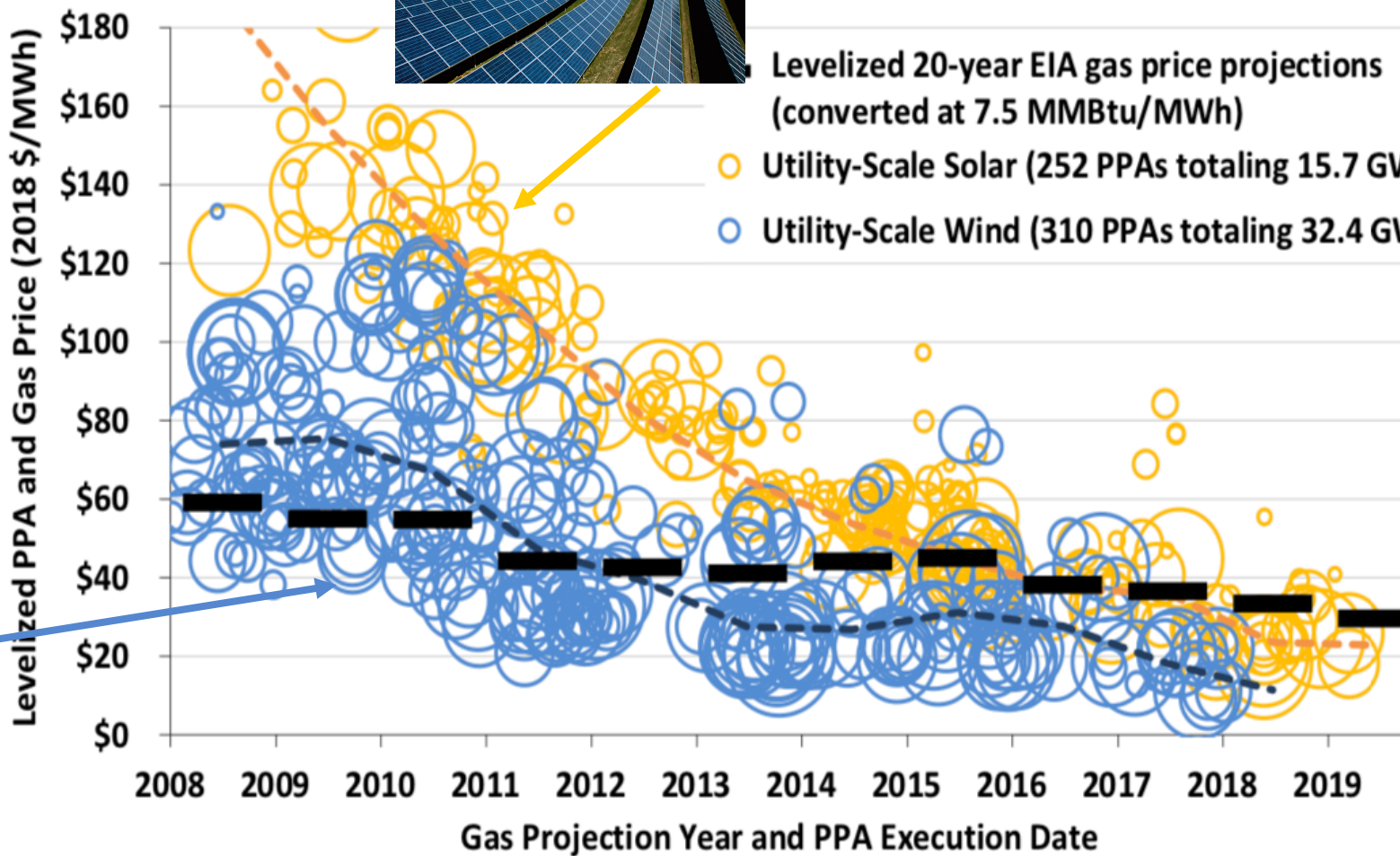
$$\eta_{\max} = \frac{\int_{T_{\text{amb}}}^{T_H} dW_{\max}}{\int_{T_{\text{amb}}}^{T_H} dQ} = 1 - \frac{T_{\text{amb}}}{T_H} \ln\left(\frac{T_H}{T_{\text{amb}}}\right)$$



Henry & Prasher, *Energy & Env. Science* (2014)

Carbon-Free Power Generation

r Group

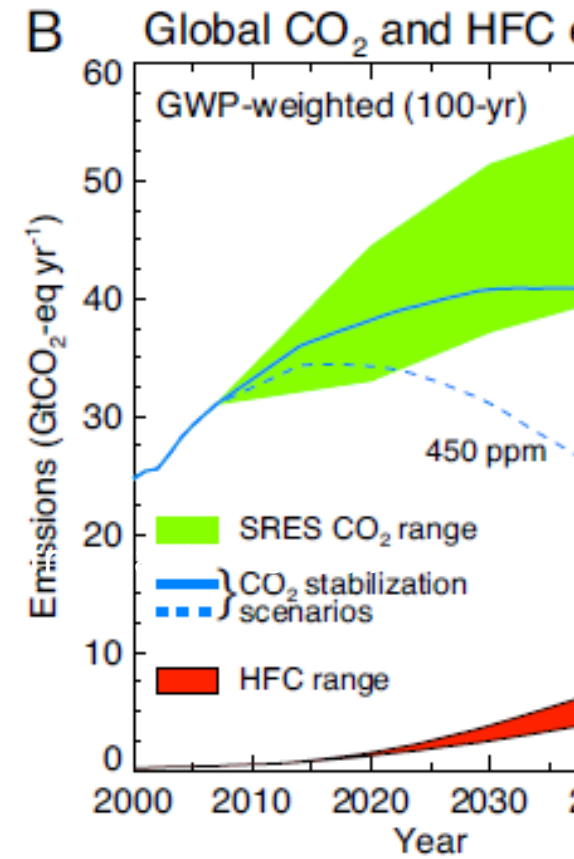
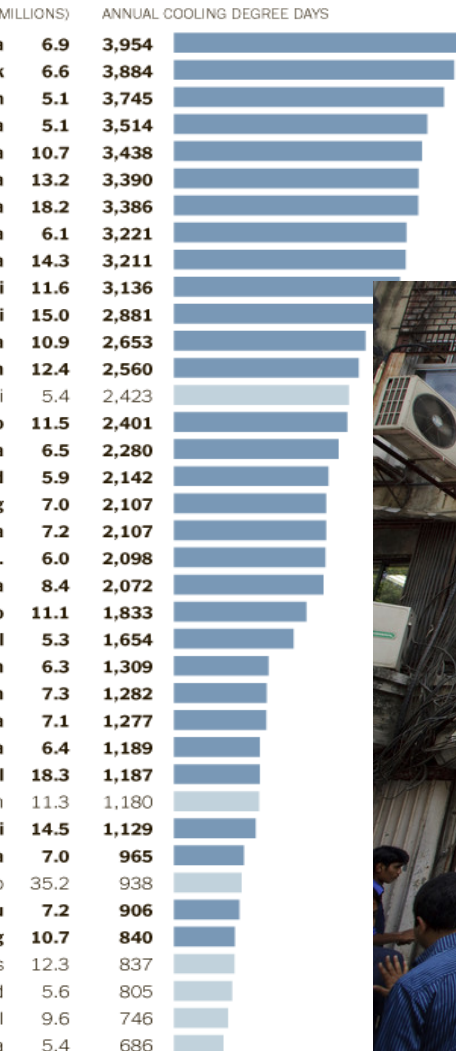


"Any suffi

Global Warming Potential of Refrigerants

an oddly named index that measures (but not in actual
conditioning — in some of the world's largest urban areas.
bold.

**Finding the right
refrigerant with GWP < 1
is a grand challenge**



Velders et al., PNAS (200

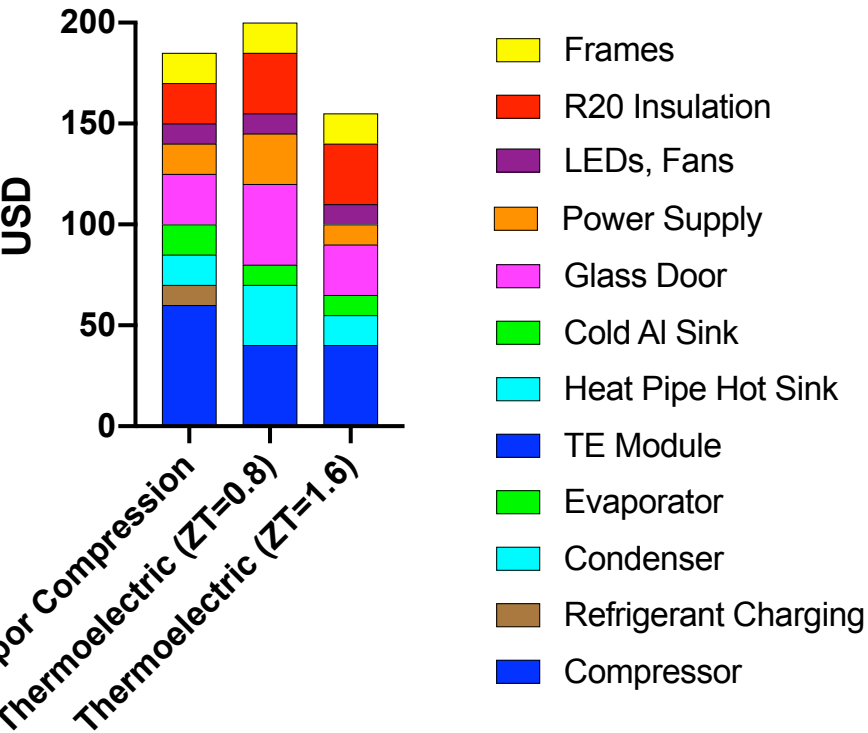


The Kigali Amendment to the Montreal Protocol: HFC Phase-down



Cost Breakdown

BOM DL Beverage/Wine Coolers



True®



Godrej

Courtesy: Sheetak Inc.

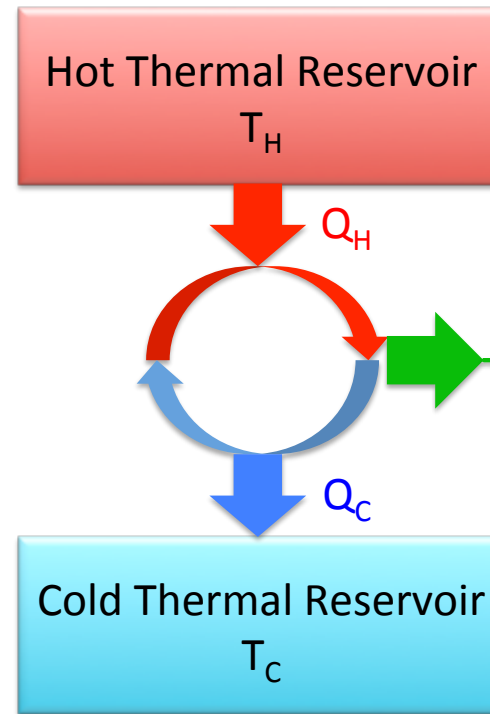
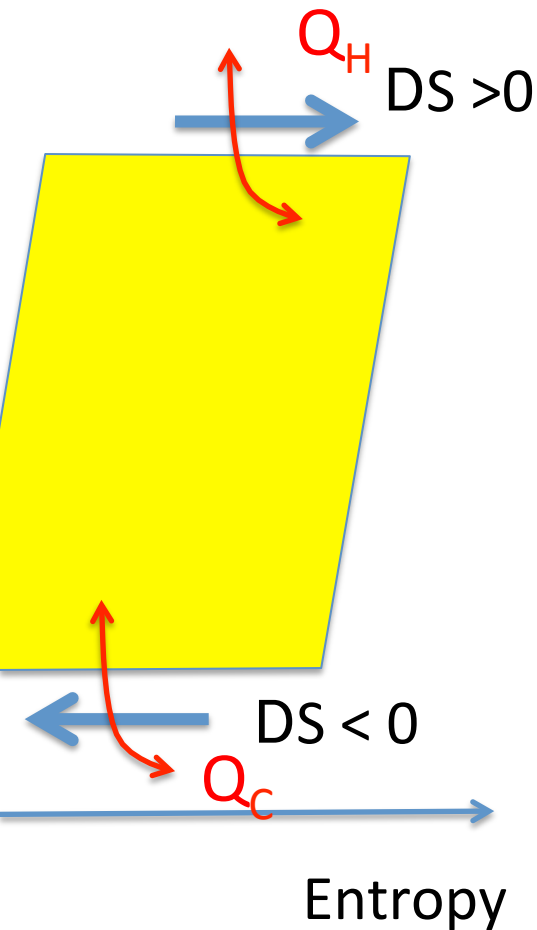
Refrigeration Technology

Engineering

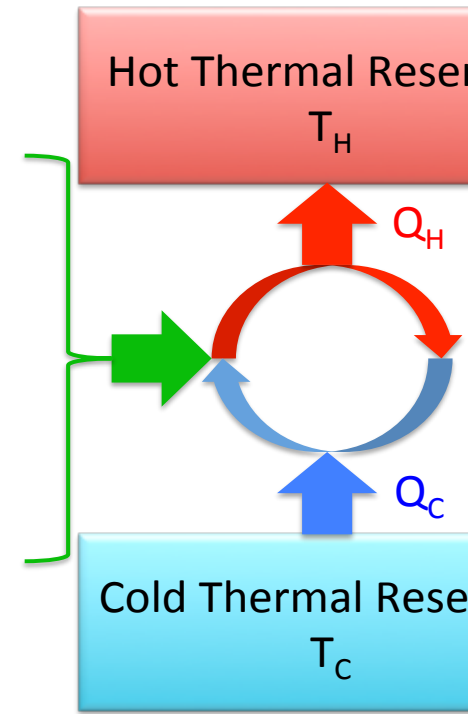
"Any sufficiently advanced technology is indistinguishable from magic." Arthur C. Clarke

Heat Engines

Power Generation & Cooling



WORK
 Mechanical
 Electrical
 Chemical
 Electrochemical
 Magnetic
 Electromagnetic



Liquid-Vapor:
 Thermoelectric:
 Electrocaloric:
 Magnetocaloric:
 Electrochemical:
 Chemical:

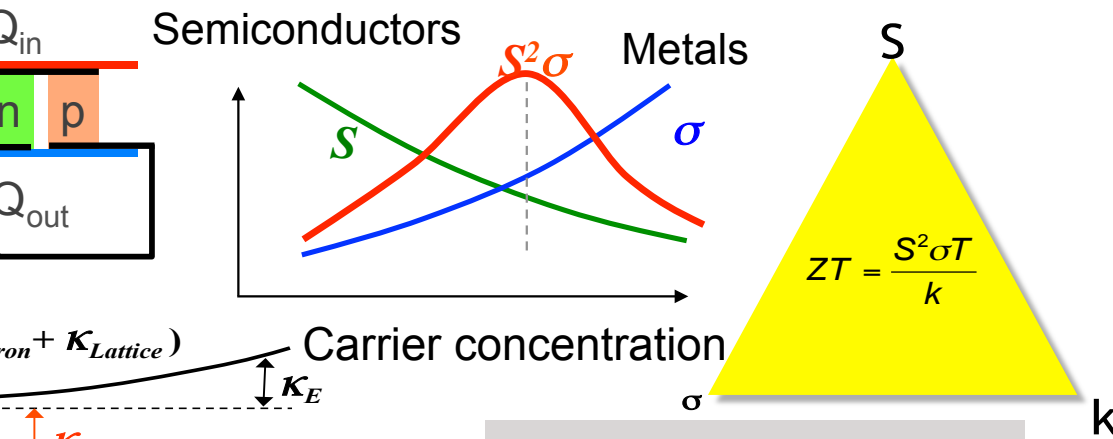
DS , configurational entropy of gas
 DS , configurational entropy of electrons in
 DS , orientational entropy of electric dipoles
 DS , orientational entropy of magnetic dipoles
 DS , entropy of solvent-solute in redox reactions
 DS , entropy of oxidizing/reducing medium

Entropy of Energy Carriers & Excitations

- Vapor:** DS , configurational entropy of gas
- Electric:** DS , configurational entropy of electrons in k-space
- Caloric:** DS , orientational entropy of electric dipoles
- Magnetic:** DS , orientational entropy of magnetic dipoles
- Chemical:** DS , entropy of solvent-solute in redox reaction
- Electrochemical:** DS , entropy of oxidizing/reducing medium

Water Liquid-Vapor:
 $DS = 6.3 \text{ kJ/kg-K} \approx 1.17 \text{ mV/K}$

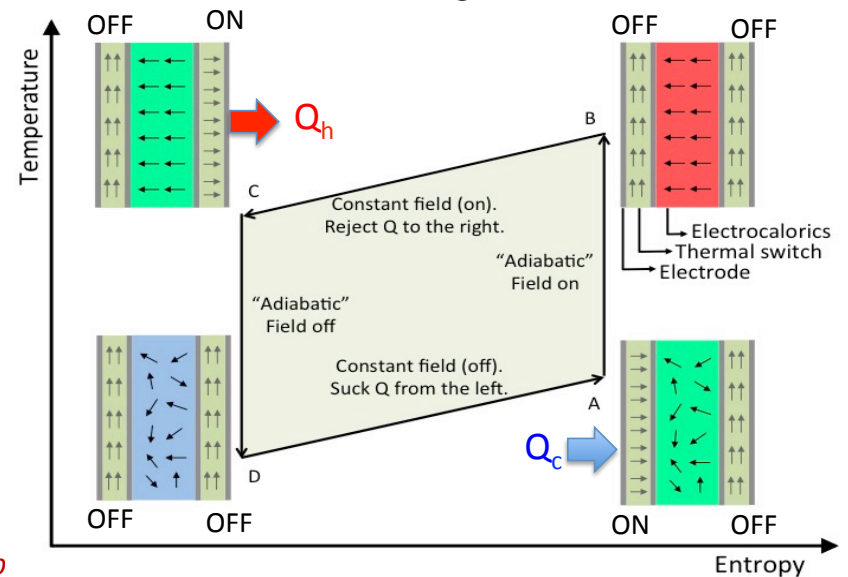
Is this the theoretical limit for



Very hard to decouple S , s , k and improve ZT to 3

"Any sufficiently advanced technology is indistinguishable from magic"

Electro/Magnetocaloric



Temp
 Sepa
 Unip

Ther
 swit

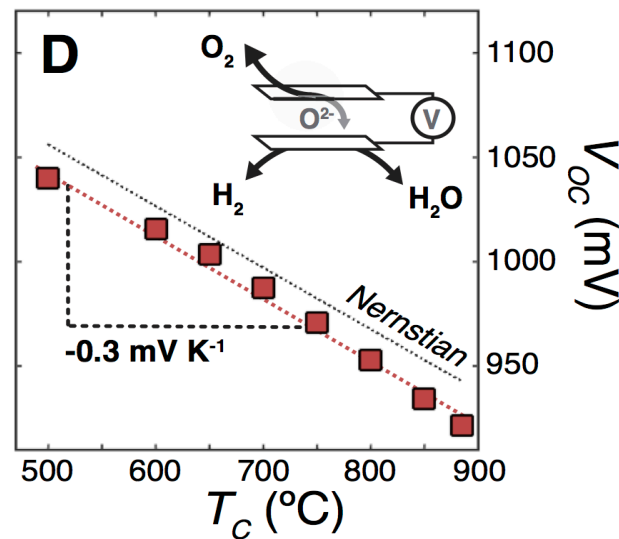
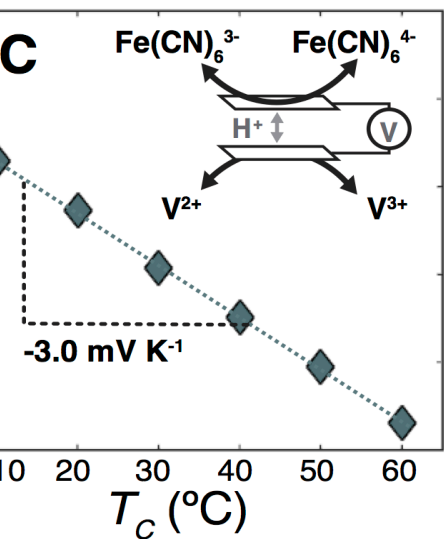
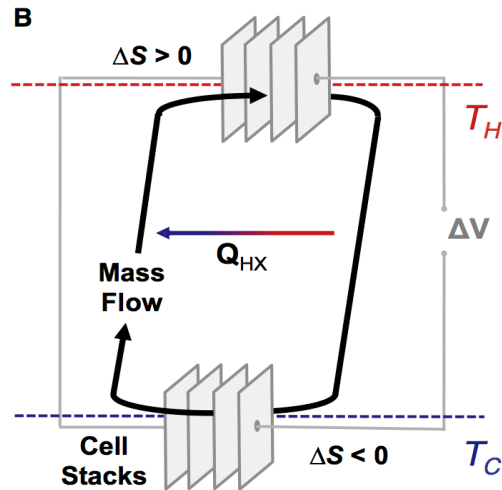
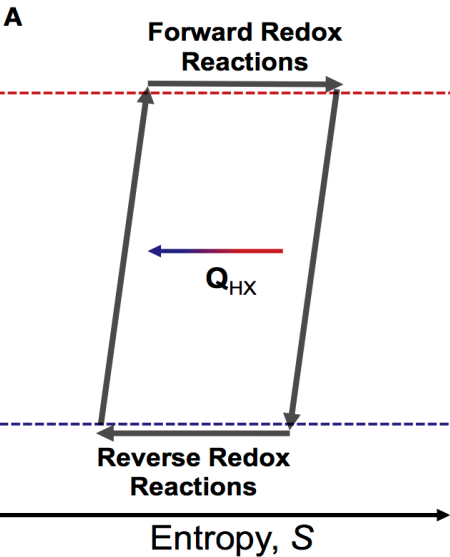
Continuous Electrochemical Heat Engines



Ian McKay



Andre

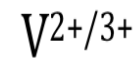
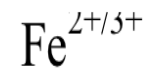


Redox Couple



DS > 1.17 mV/K
More than liquid-vapor phase trans

Methyl viologen (2+/1+)



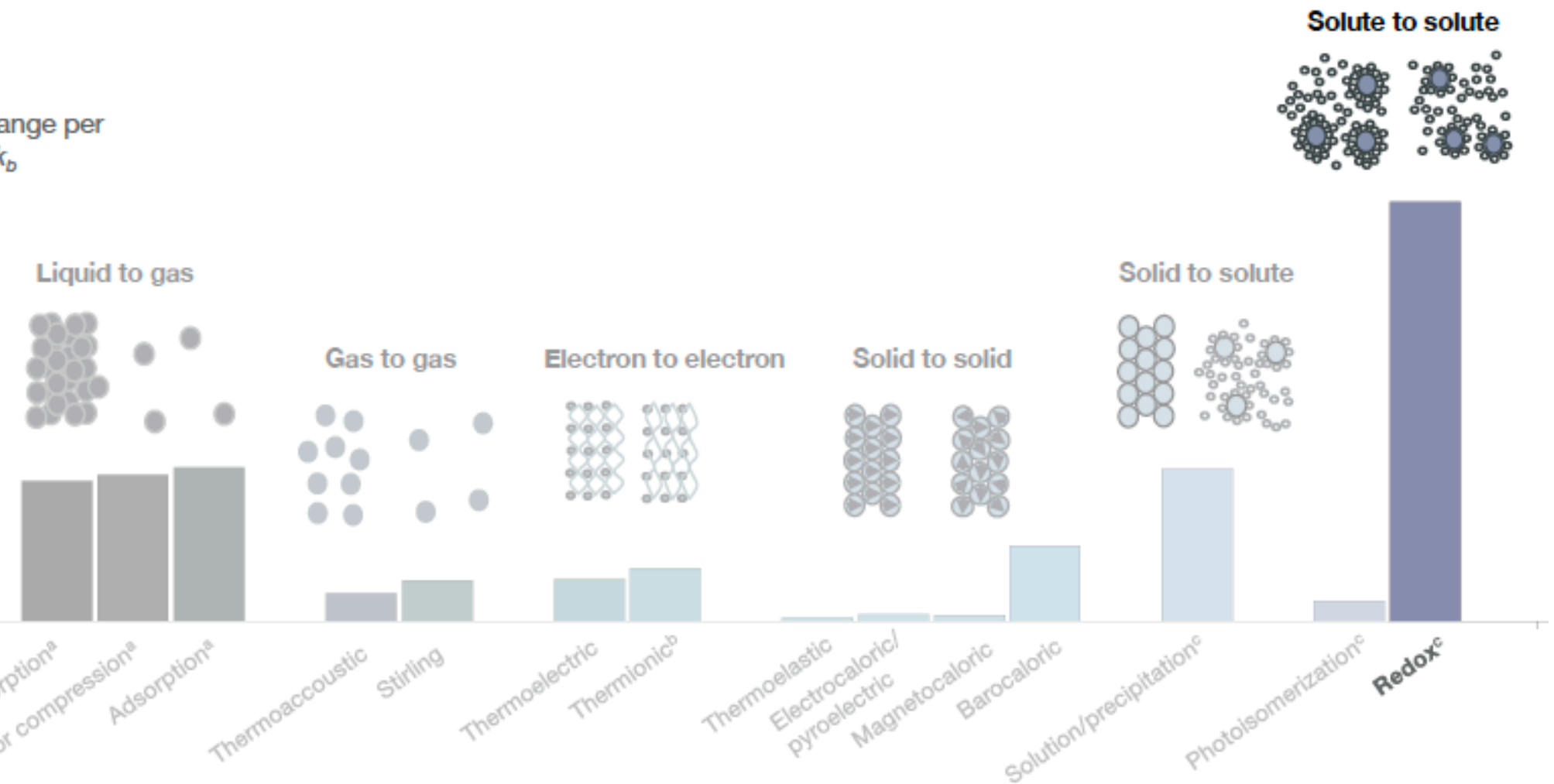
Poletayev, McKay, Chueh, Majumdar,
Energy & Env Sci (2018)

ble from magic." Arthur C. Clarke

Entropy Changes in Nature

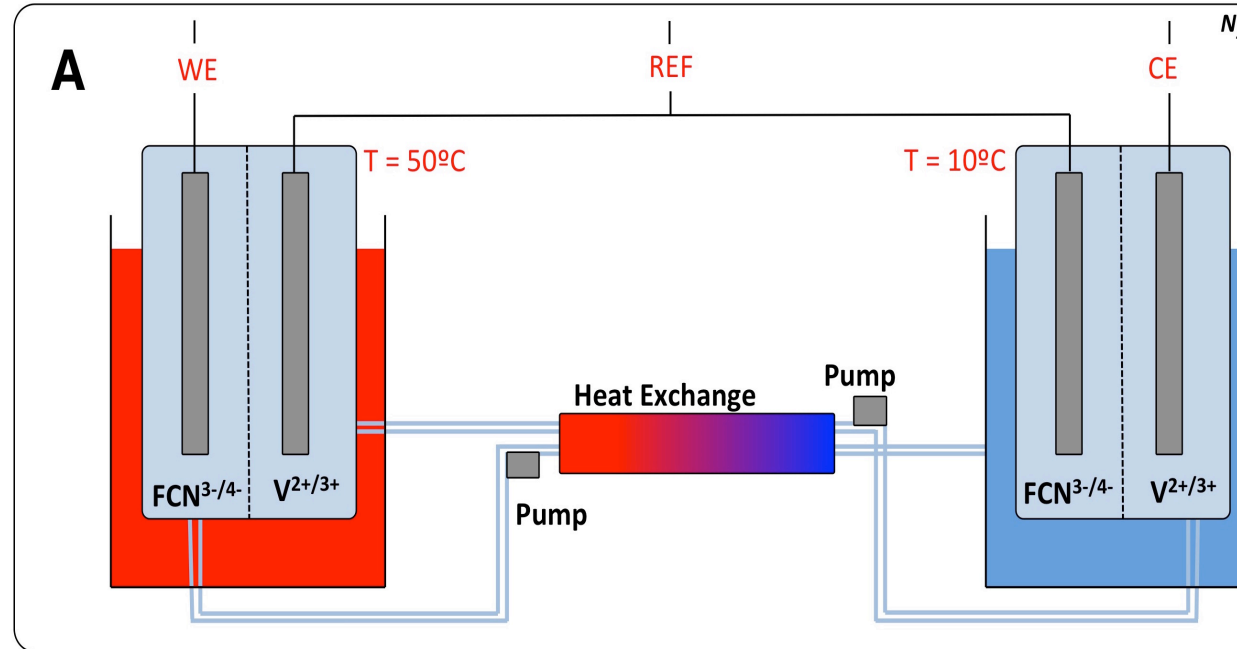
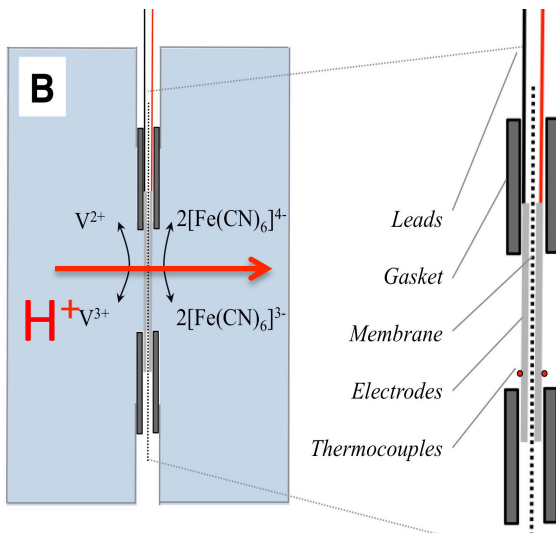
Group

Change per
 k_B



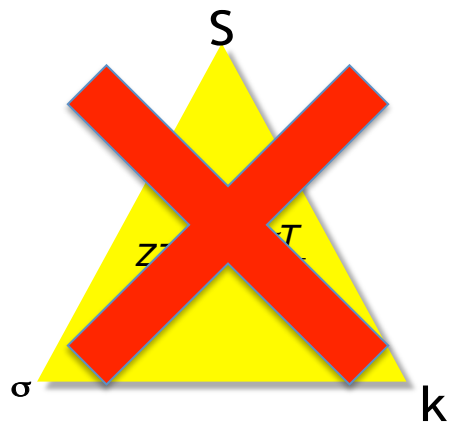
"Any sufficiently advanced technology is indistinguishable from magic." Arthur C. Clarke

Experimental Set Up

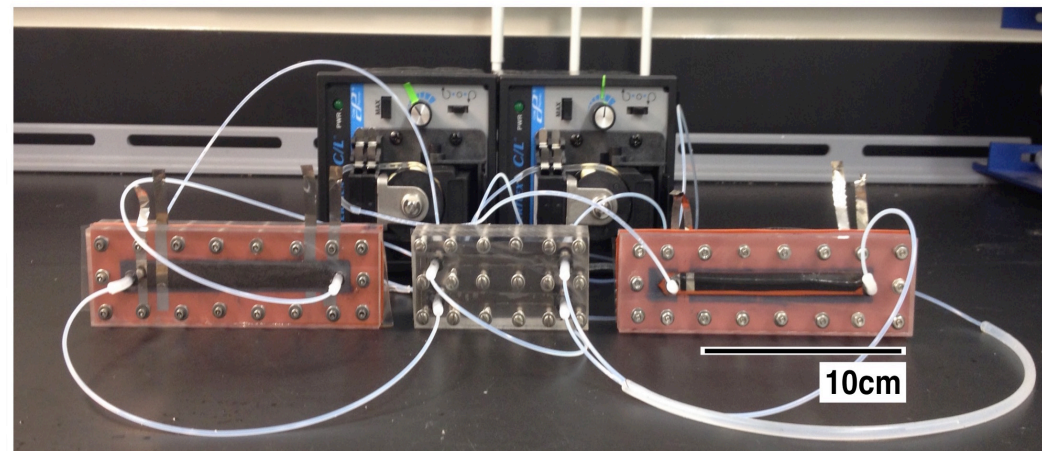


COUPLING OF TRANSPORT & ENTROPY

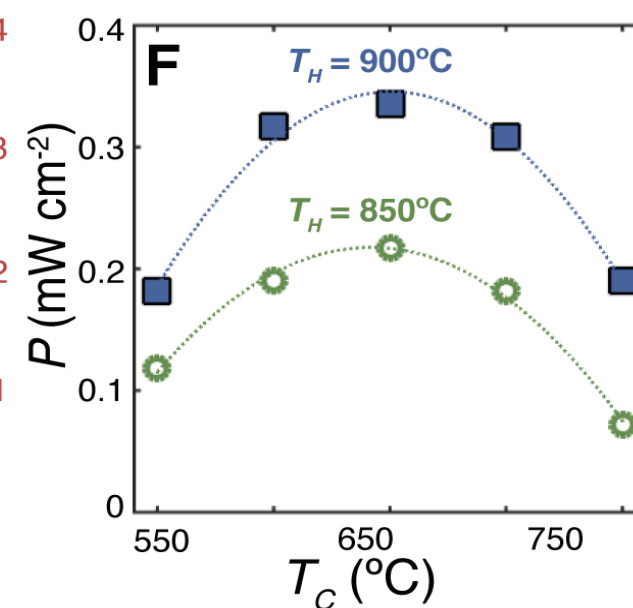
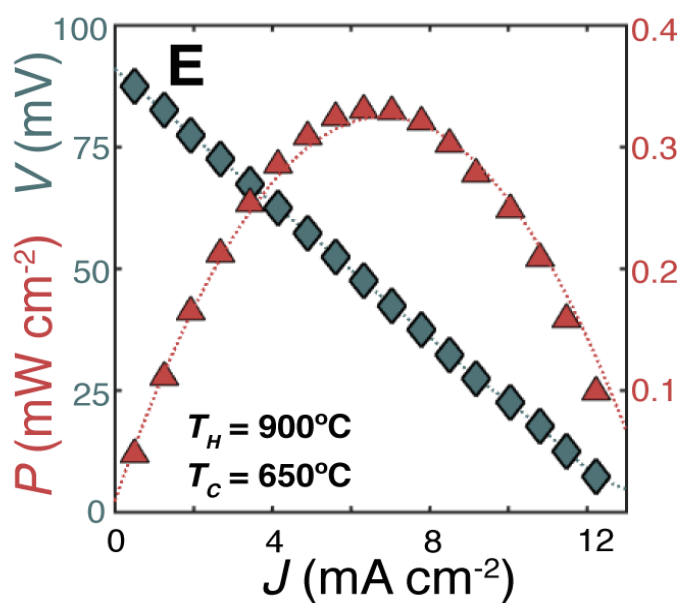
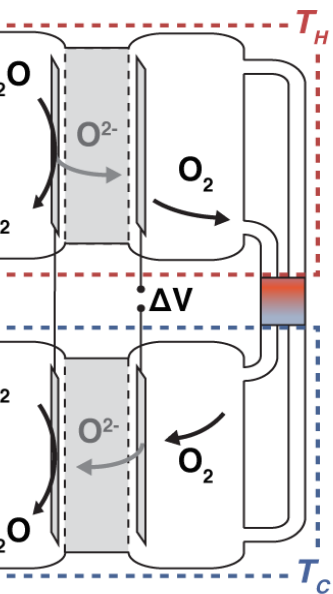
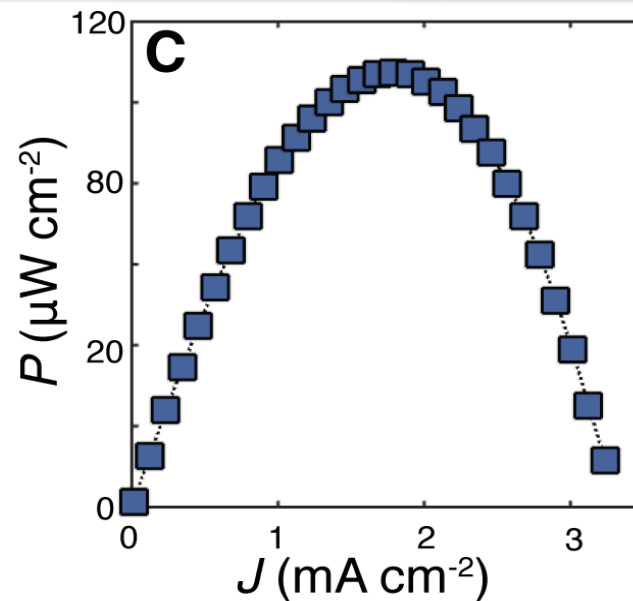
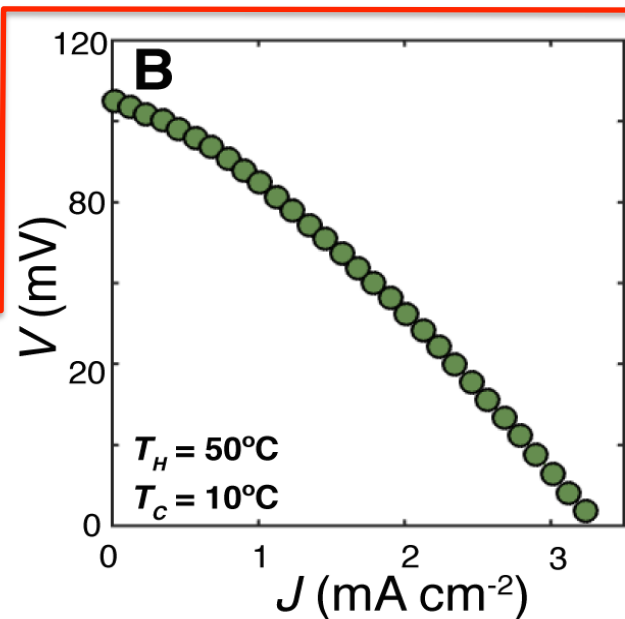
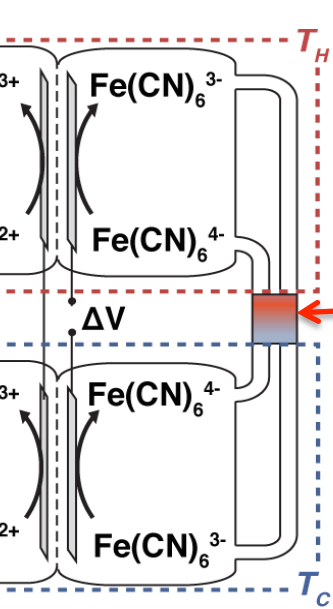
Microchemical redox reactions
 resistance of proton exchange membrane
 cooperative heat exchanger



B



Continuous Electrochemical Heat Engines



Recuperative
Heat Exchange

Currently about
20% of Carnot
limit

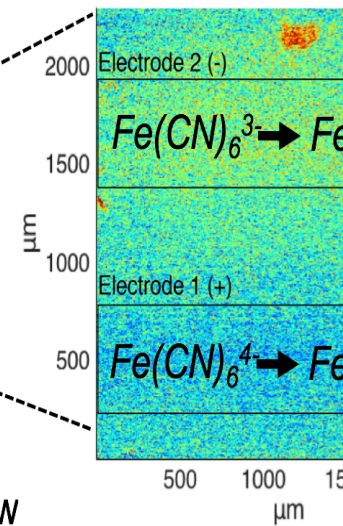
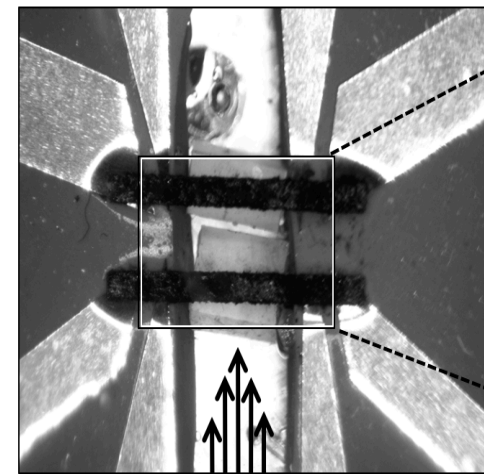
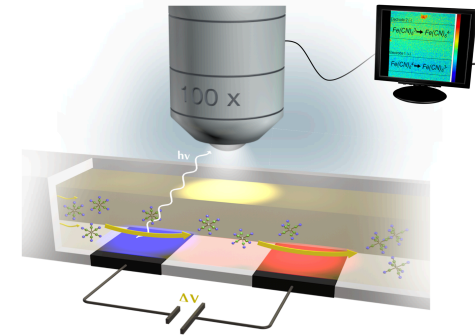
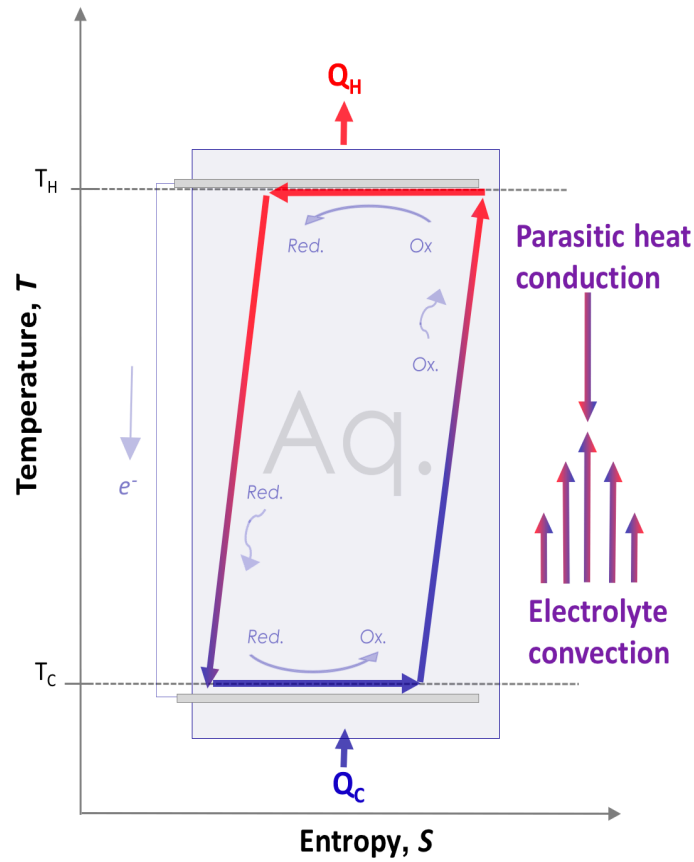
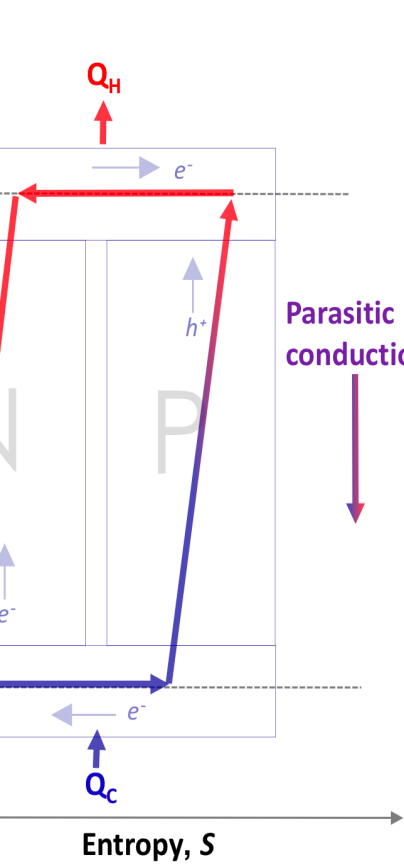
Estimates of
60-70% of
Carnot limit with
better heat
exchangers

Redox Refrigeration



Ian McKay

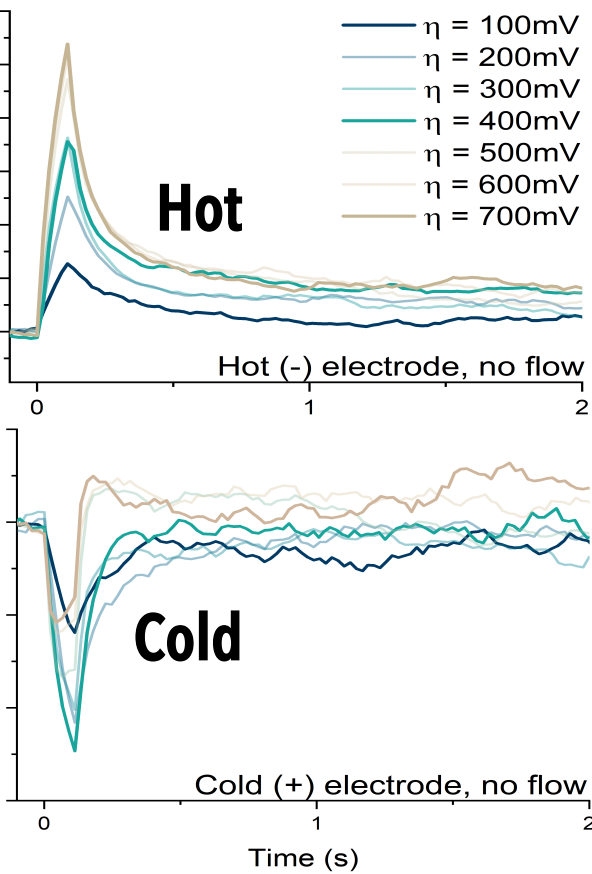
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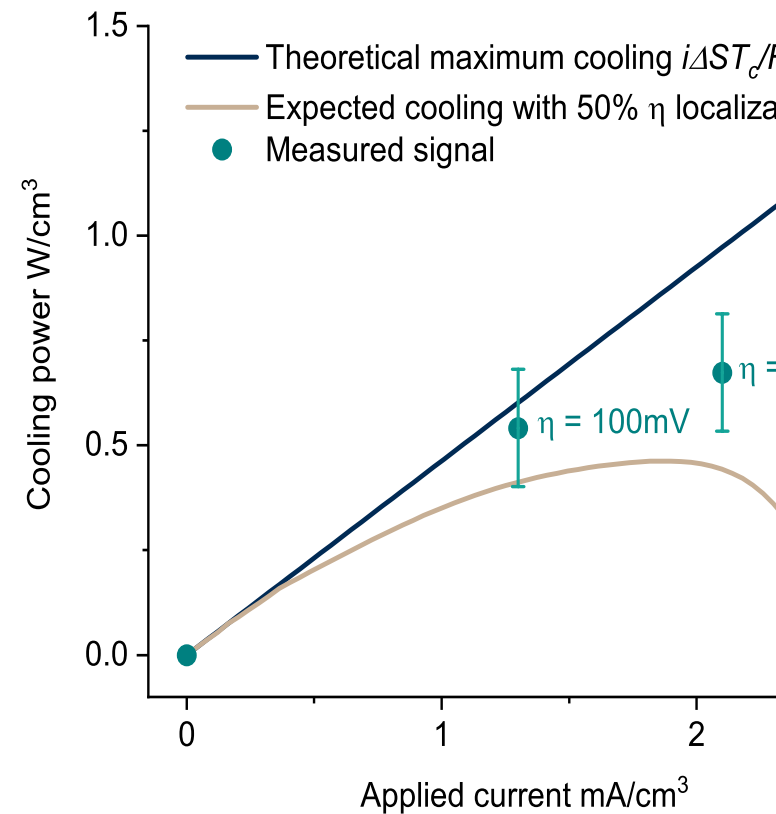
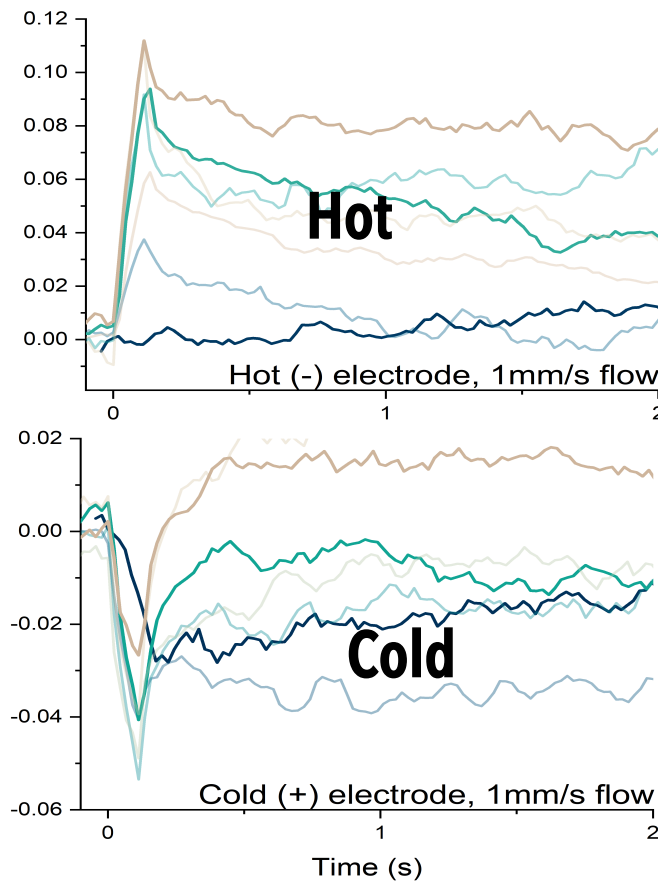
McKay, Kunz, Majumdar, Redox Refrigeration, *Scientific Reports* (2019).

Redox Refrigeration with Flow

No Flow



1 mm/s Flow



Classification of Redox Refrigerants

$$= [\Delta S (\text{J/mol}\cdot\text{K}) T (\text{K}) - E \downarrow a (\text{J/mol})] \times C (\text{mol/kg}) / c \downarrow p (\text{J/kg}\cdot\text{K})$$

able Thermodynamics

$$/ \text{mol}\cdot\text{K}) \cdot C (\text{mol/kg}) / c \downarrow p (\text{J/}$$

$$YT = DT_{max}$$

$$= \Delta S (\text{J/mol}\cdot\text{K}) T (\text{K}) / E \downarrow a (\text{J/mol})$$

$$> 1$$

Redox reaction	α , mV/K	YT= ΔT_{max} @300K, K	$Q_{g/b}$ (
$2\text{H}^+ + 2\text{e}^- \rightarrow \text{H}_2 (\text{g})$	0.8	307	$4.0^{9,2}$
$\text{V}^{3+} + \text{e}^- \rightarrow \text{V}^{2+}$	1.2-1.9*	36	6.0^{27}
$\text{V}^{5+} + \text{e}^- \rightarrow \text{V}^{4+}$	-0.2*	6	2.4^{27}
$\text{Br}_2 (\text{l}) + \text{e}^- \rightarrow 2\text{Br}^-$	0.3*	14	1.7^{28}
$\text{Fe}(\text{CN})_6^{3-} + \text{e}^- \rightarrow \text{Fe}(\text{CN})_6^{4-}$	-1.5*	10.6	78.4^2
$\text{Fe}^{3+} + \text{e}^- \rightarrow \text{Fe}^{2+}$	1.1*	36	95.7^3
$\text{Cr}^{3+} + \text{e}^- \rightarrow \text{Cr}^{2+}$	2.2	7	219.5
$\text{Ce}^{5+} + \text{e}^- \rightarrow \text{Ce}^{4+}$	2.3	13	36.0^9
$\text{S}_2^{2-} + 2\text{e}^- \rightarrow 2\text{S}^{2-}$	-.7	9.4	$1.3^{9,3}$



Is there a killer application for thermoelectrics?

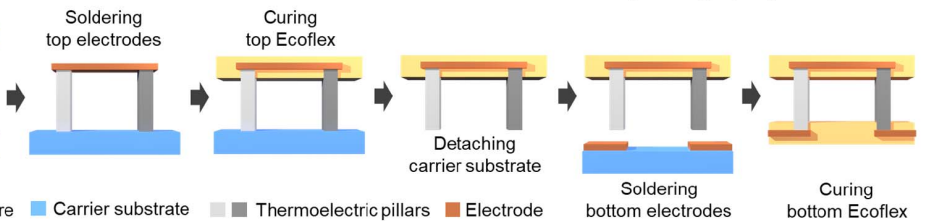
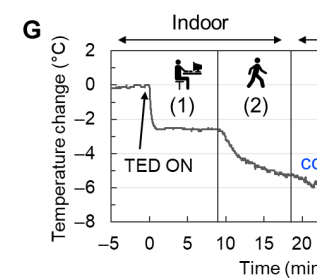
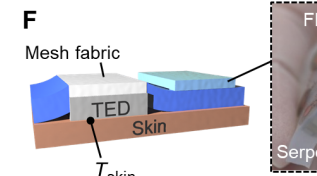
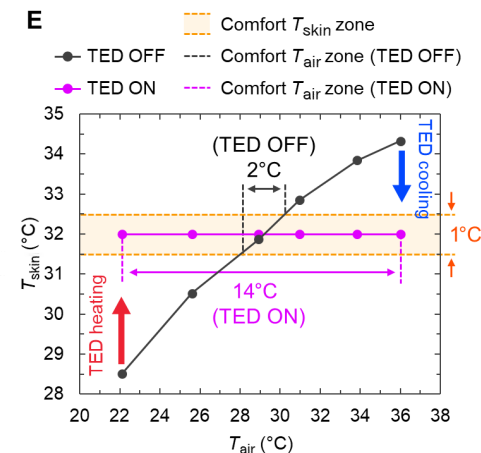
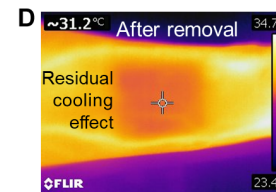
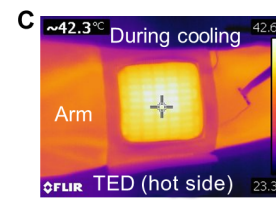
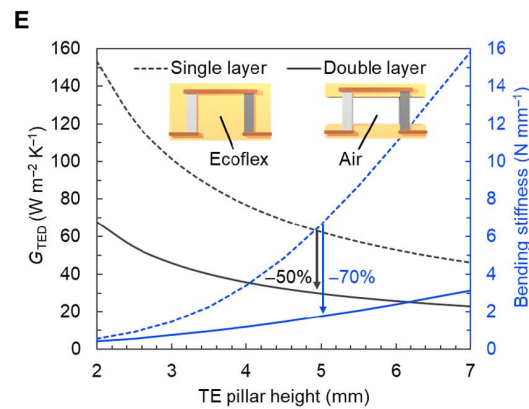
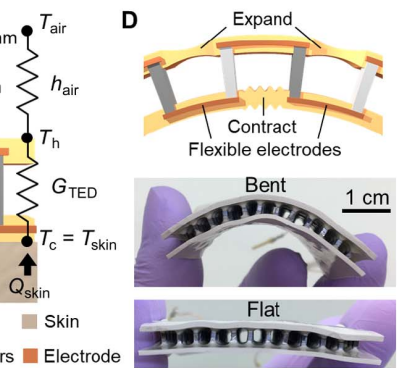
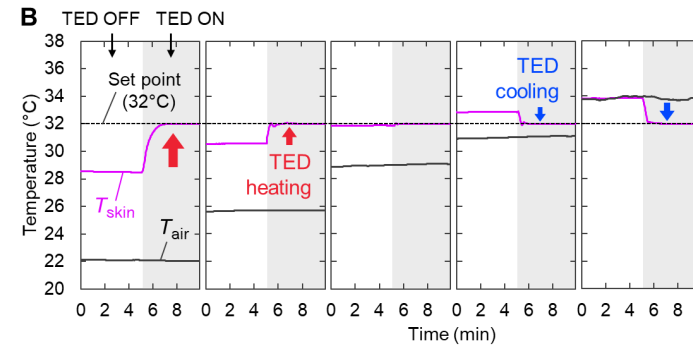
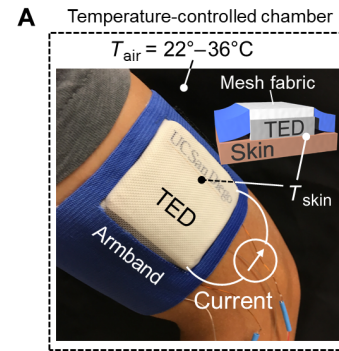
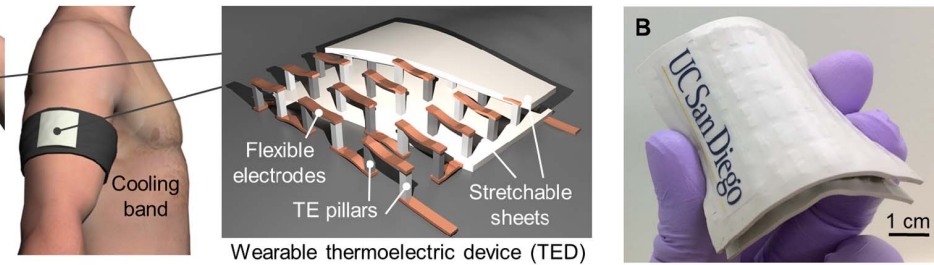
Power Generation from Waste Heat Recovery (100 °C – 800 °C)

Refrigeration & Heat Pumps (10-60 °C)

ENGINEERING

Wearable thermoelectrics for personalized thermoregulation

Sahngki Hong^{1,2}, Yue Gu^{1,3}, Joon Kyo Seo^{1,3}, Joseph Wang^{1,3}, Ping Liu^{1,3}, Y. Shirley Meng^{1,3,4}, Sheng Xu^{1,3*}, Renkun Chen^{1,2,4*}



"Any sufficiently advanced technology is indistinguishable from magic." Arthur C. Clarke