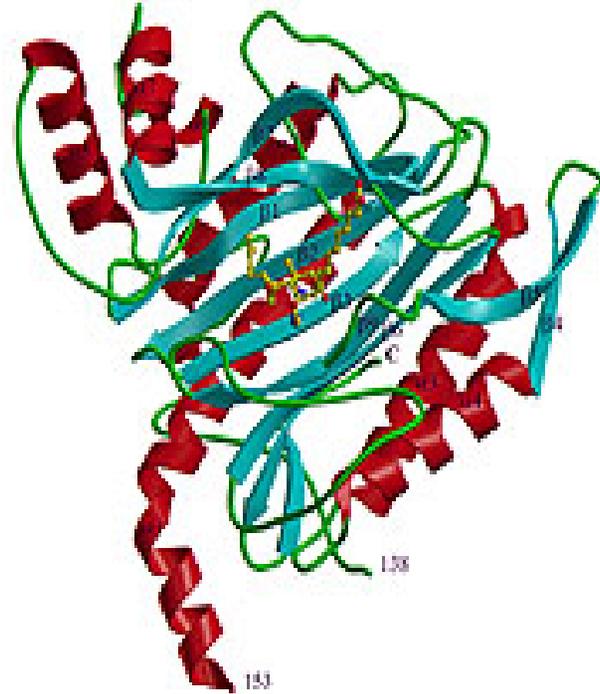
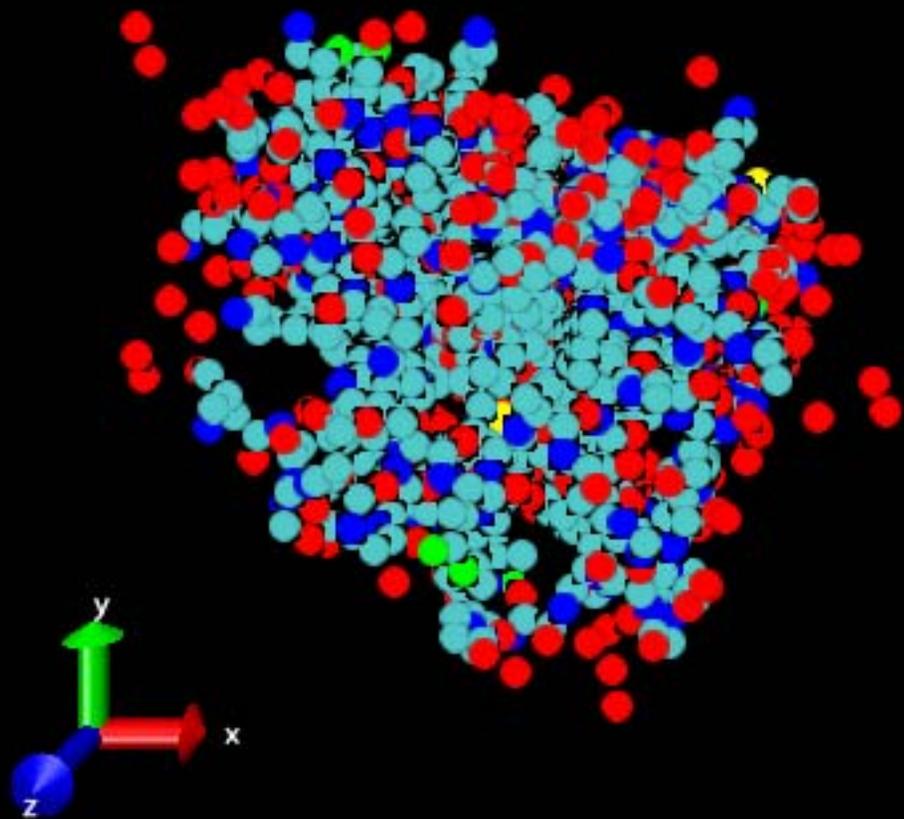
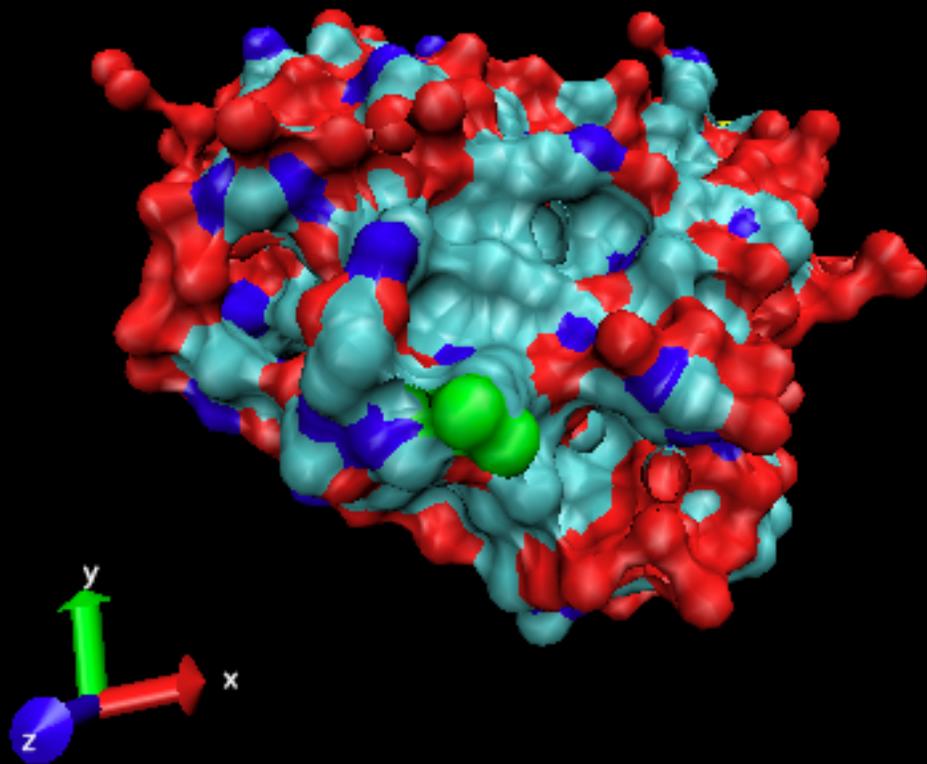


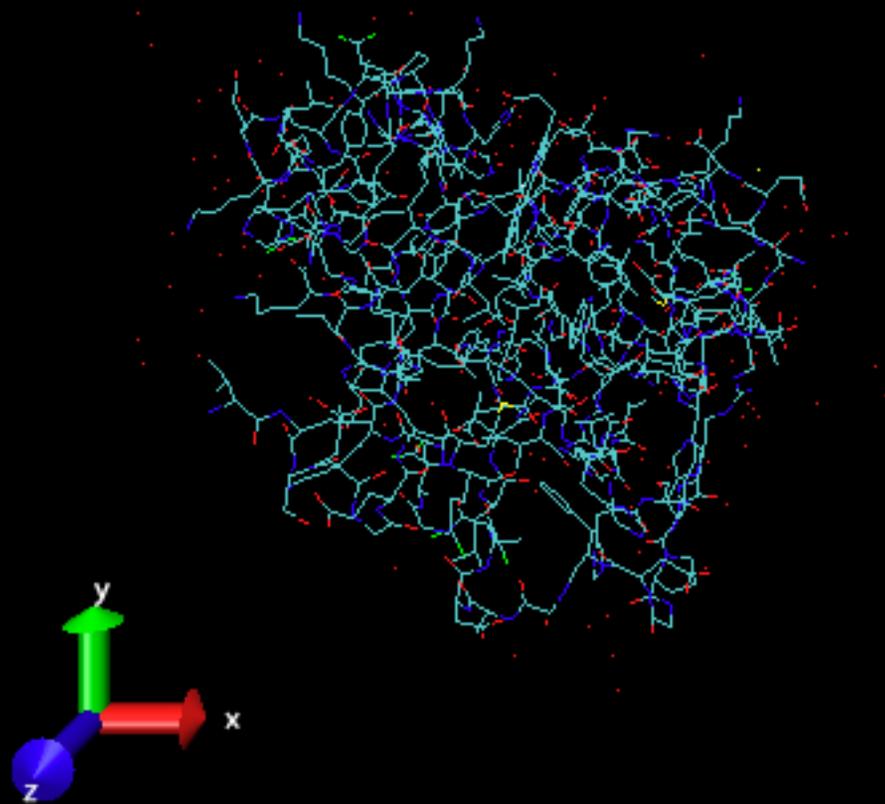
# Some Physics for Proteins

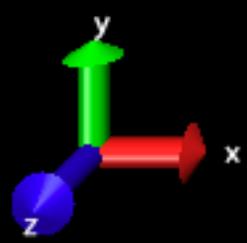
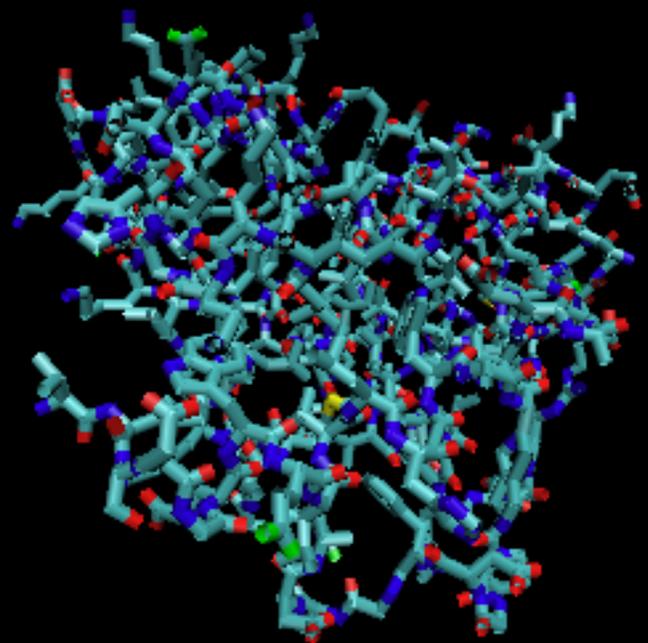


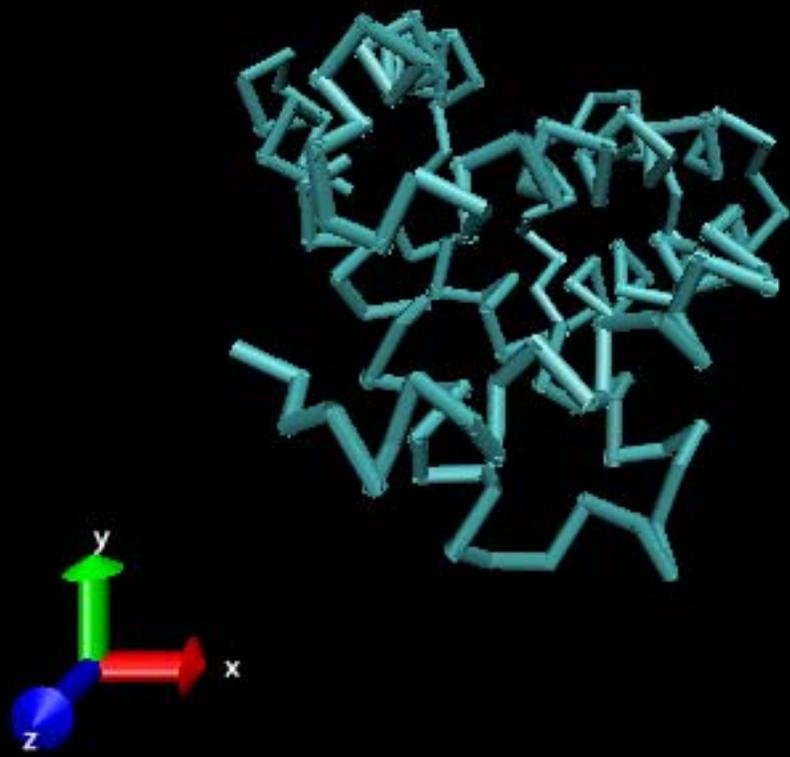
Stephen M. Durbin  
Department of Physics  
Purdue University  
Friday, 27 July 2007

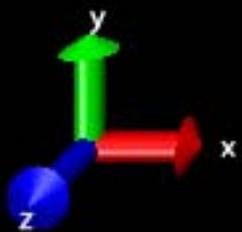


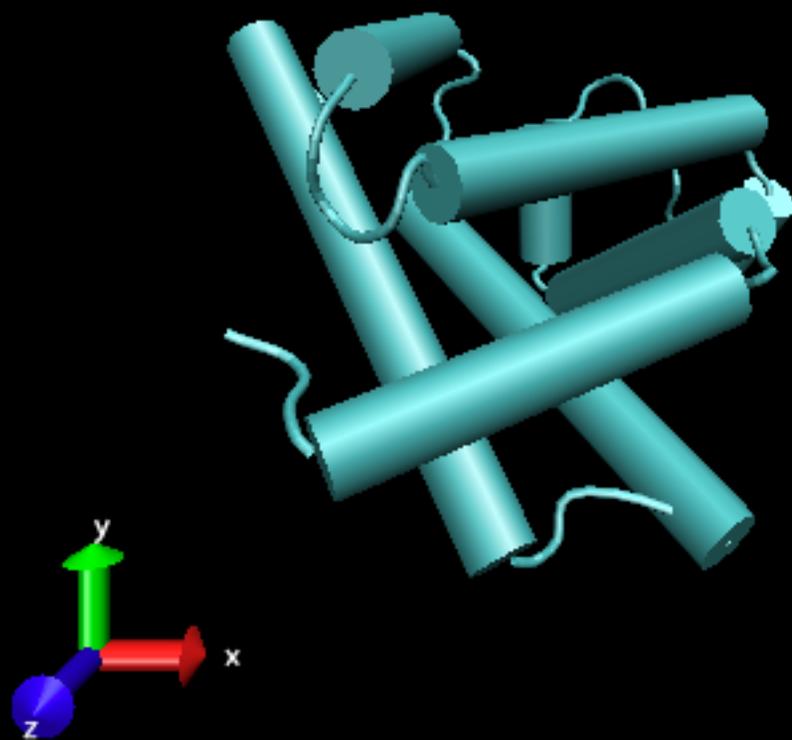


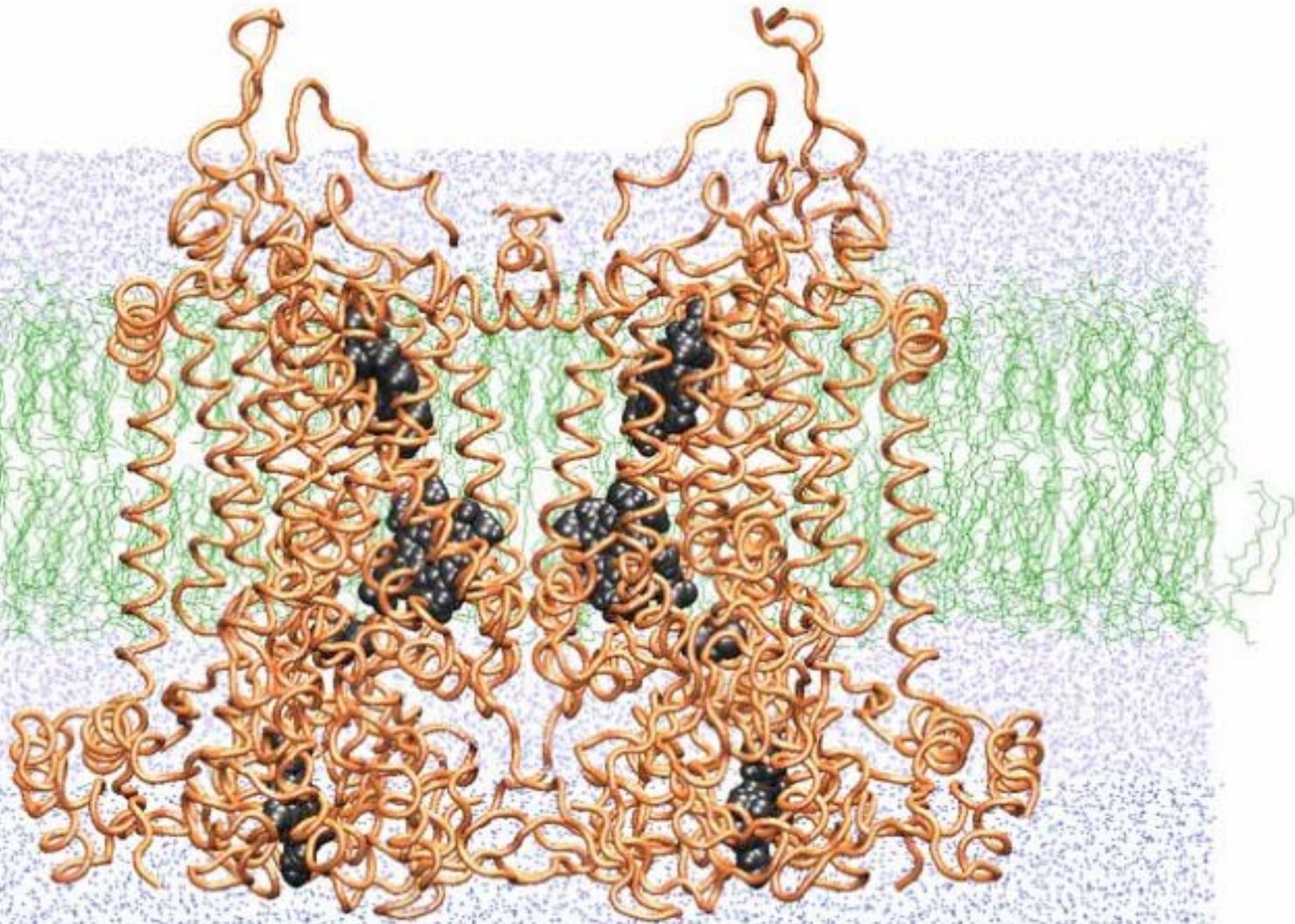




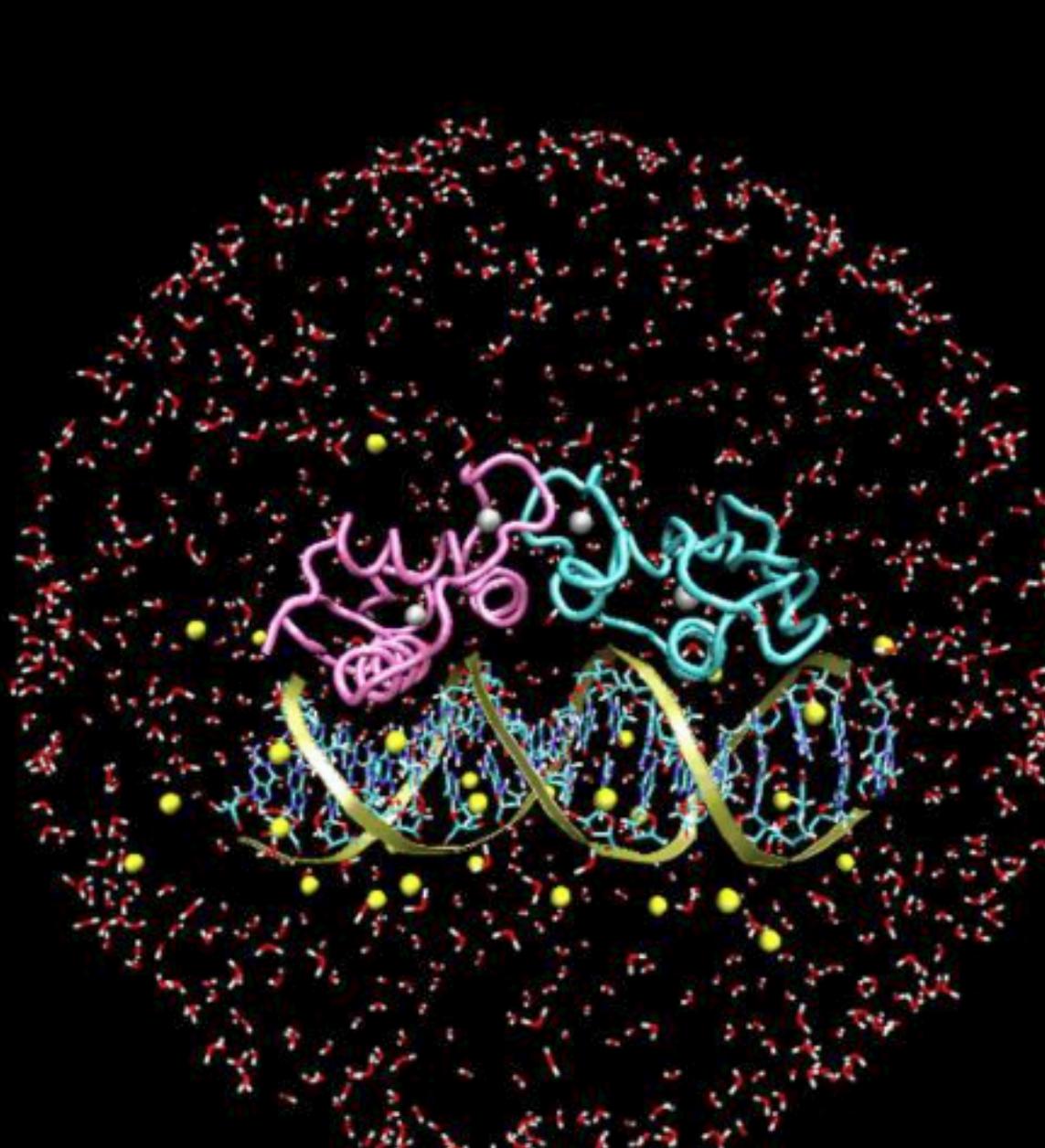




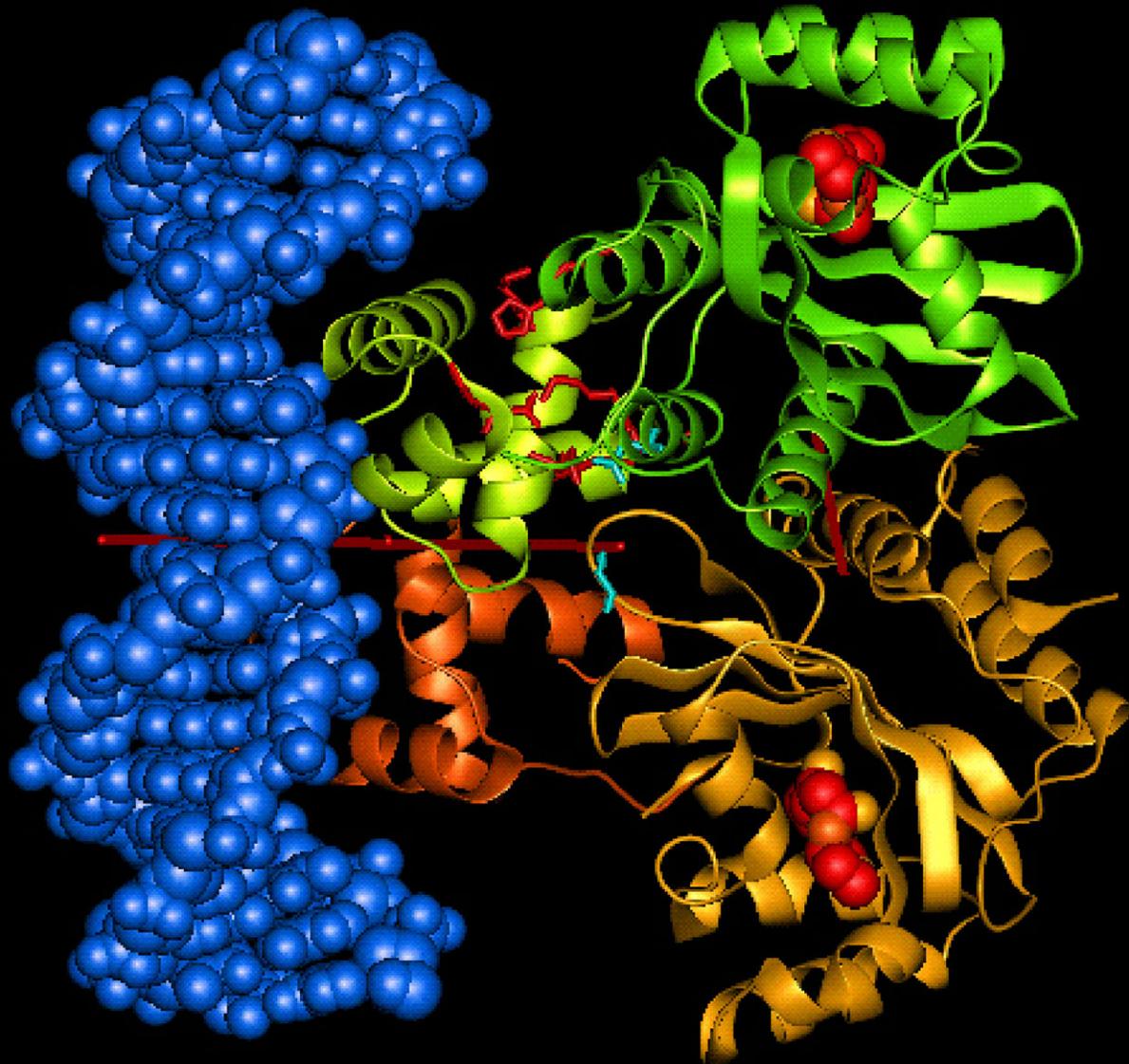


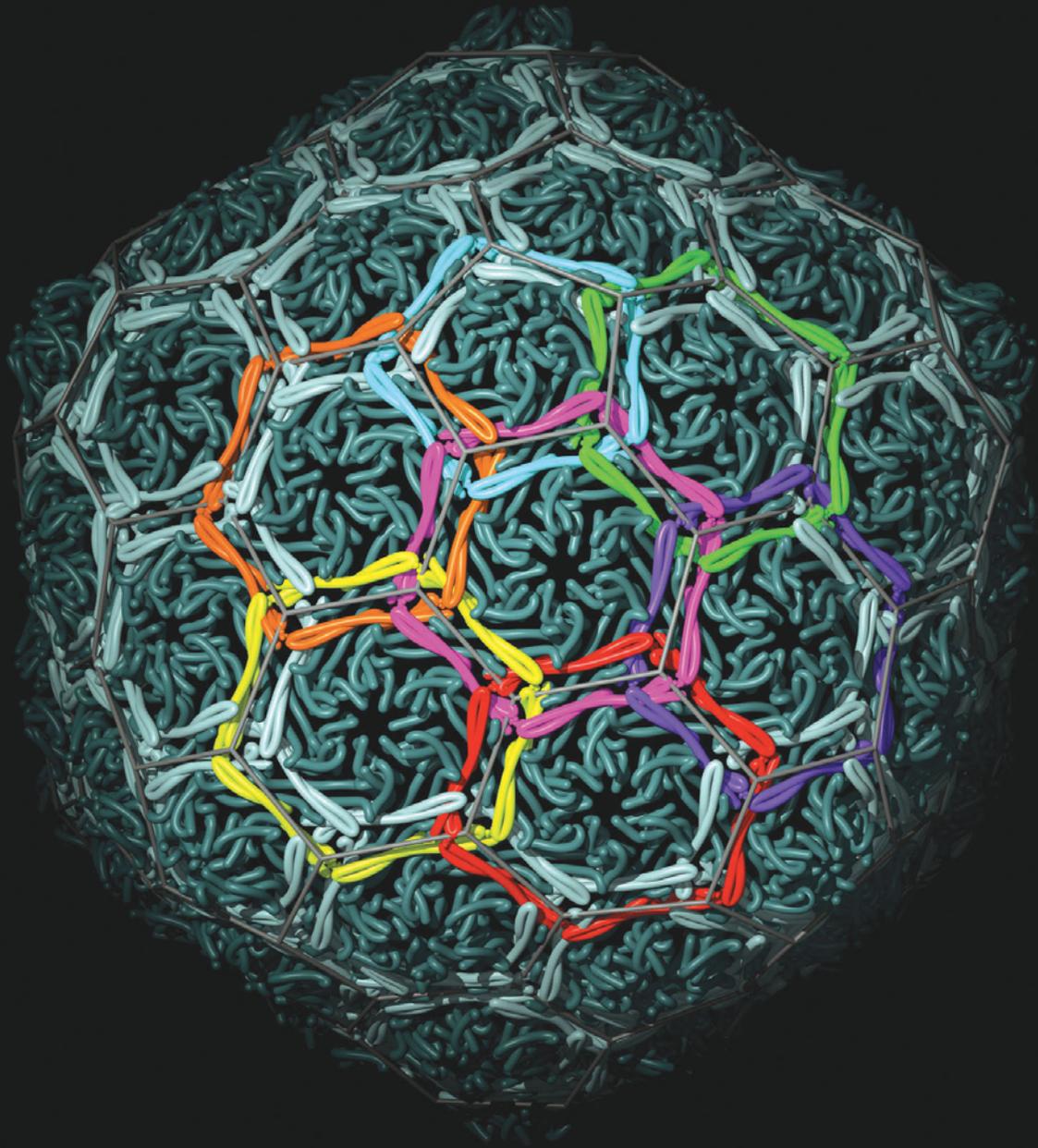


**Theoretical Biophysics Group**  
University of Illinois at Urbana-Champaign

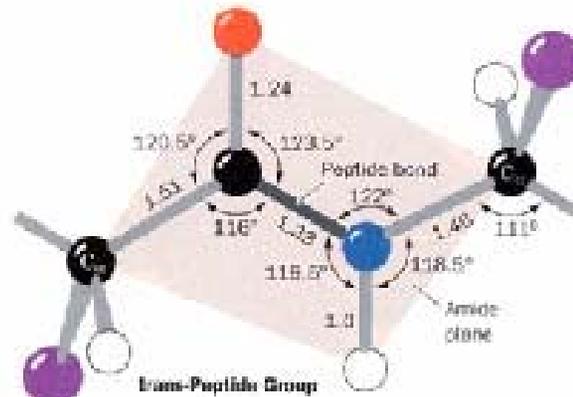


**Theoretical Biophysics Group**  
Beckman Institute  
University of Illinois at Urbana-Champaign

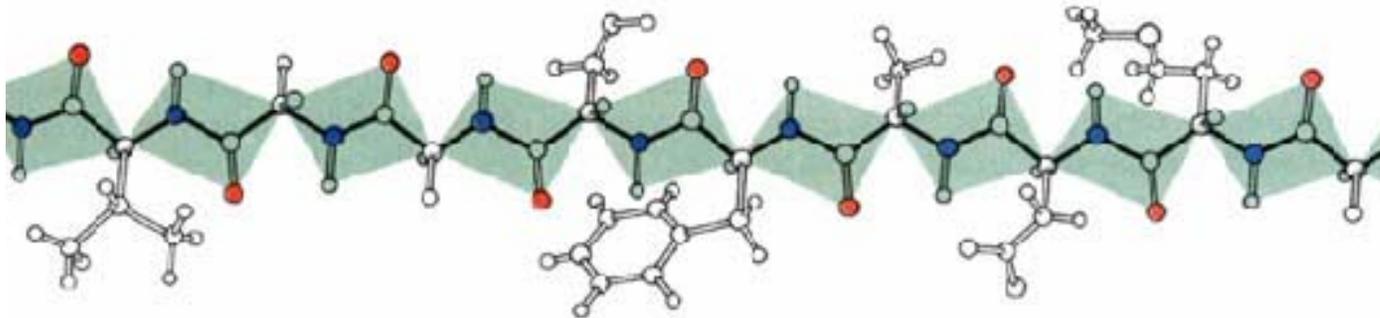


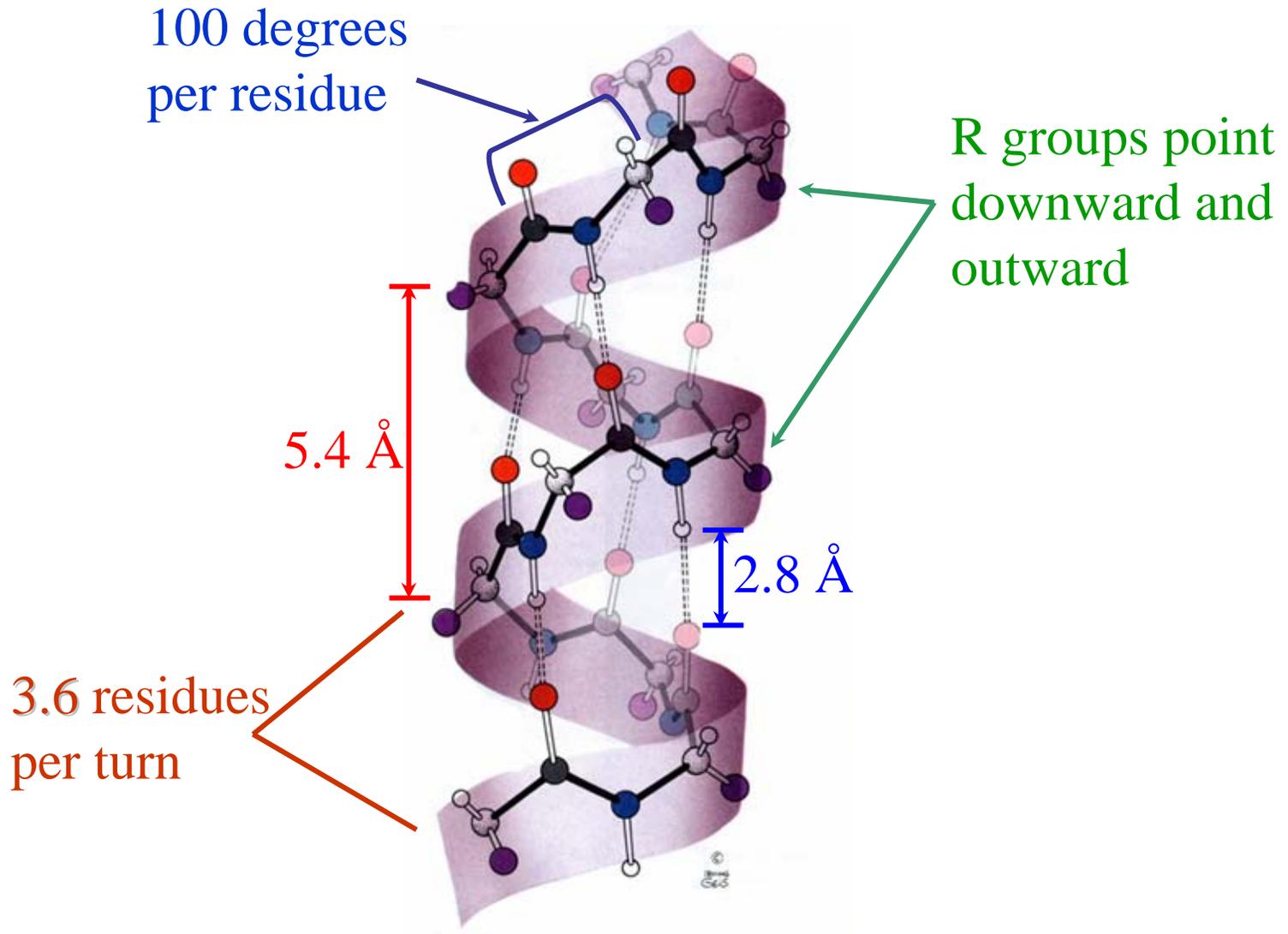


# Peptide group



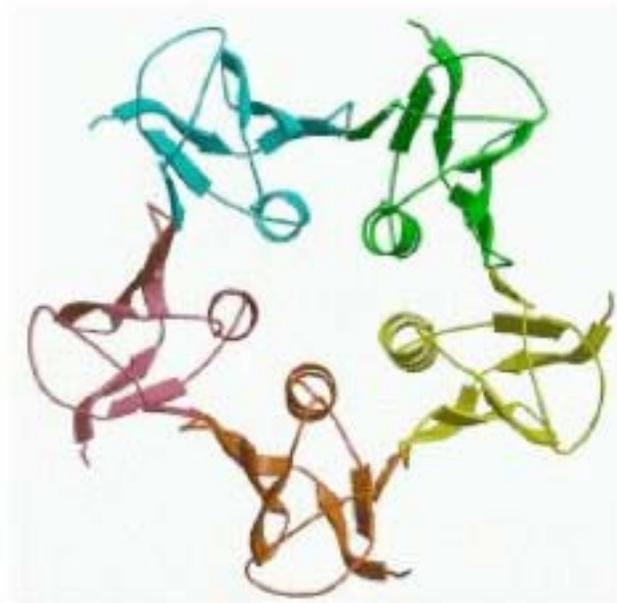
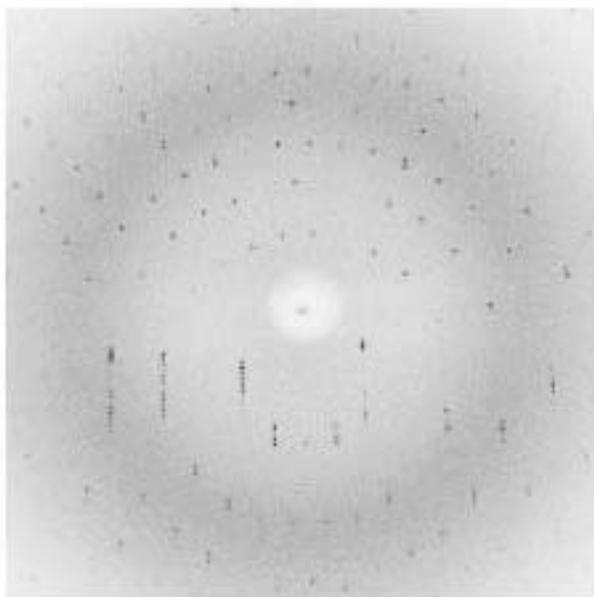
# Fully extended polypeptide chain

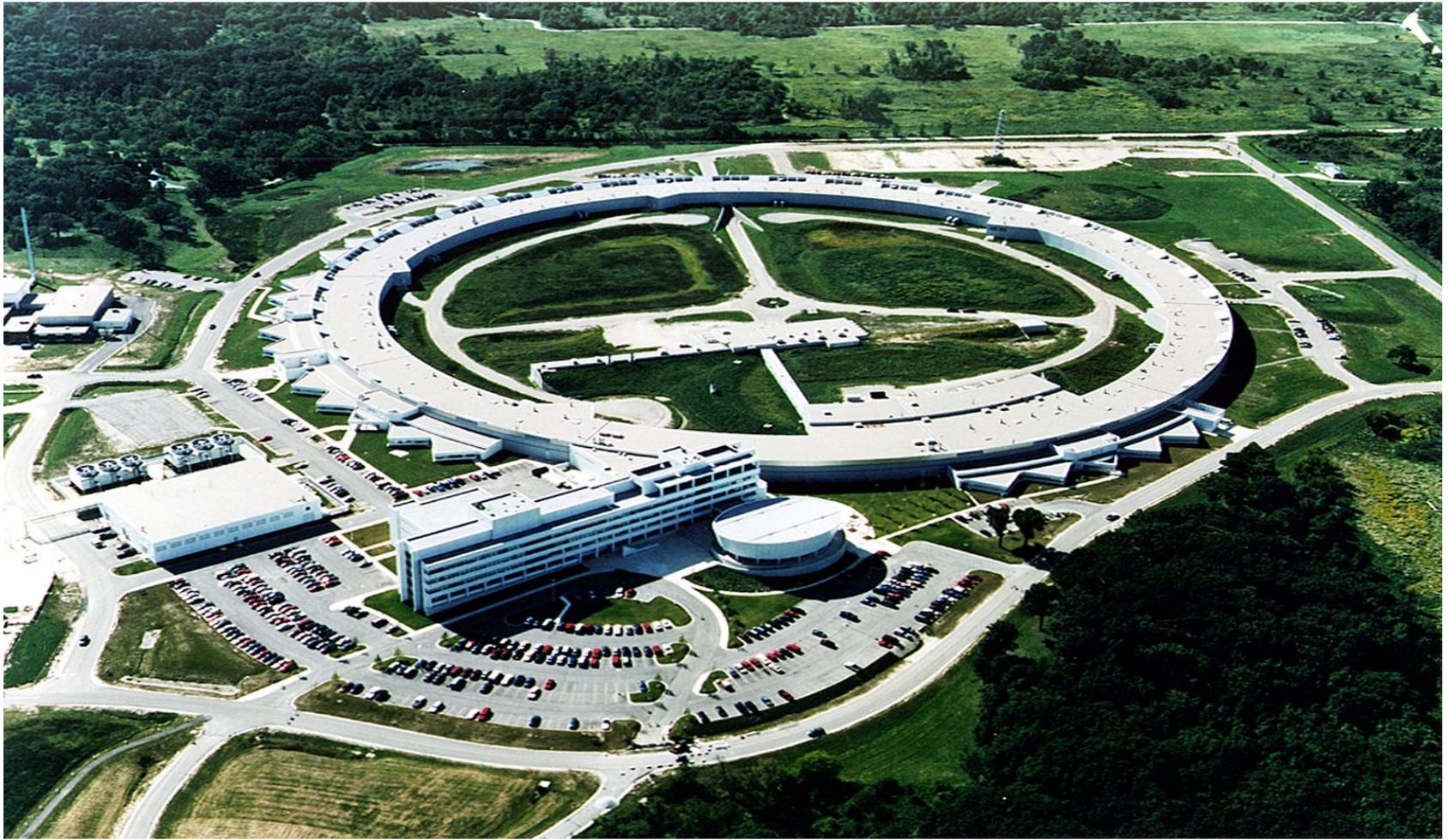




other types of helices exist...







The Advanced Photon Source

James Maxwell 1860's

Shake electrons to make light!





# 1905: Einstein's Theory of Special Relativity

Time and Space are different!!

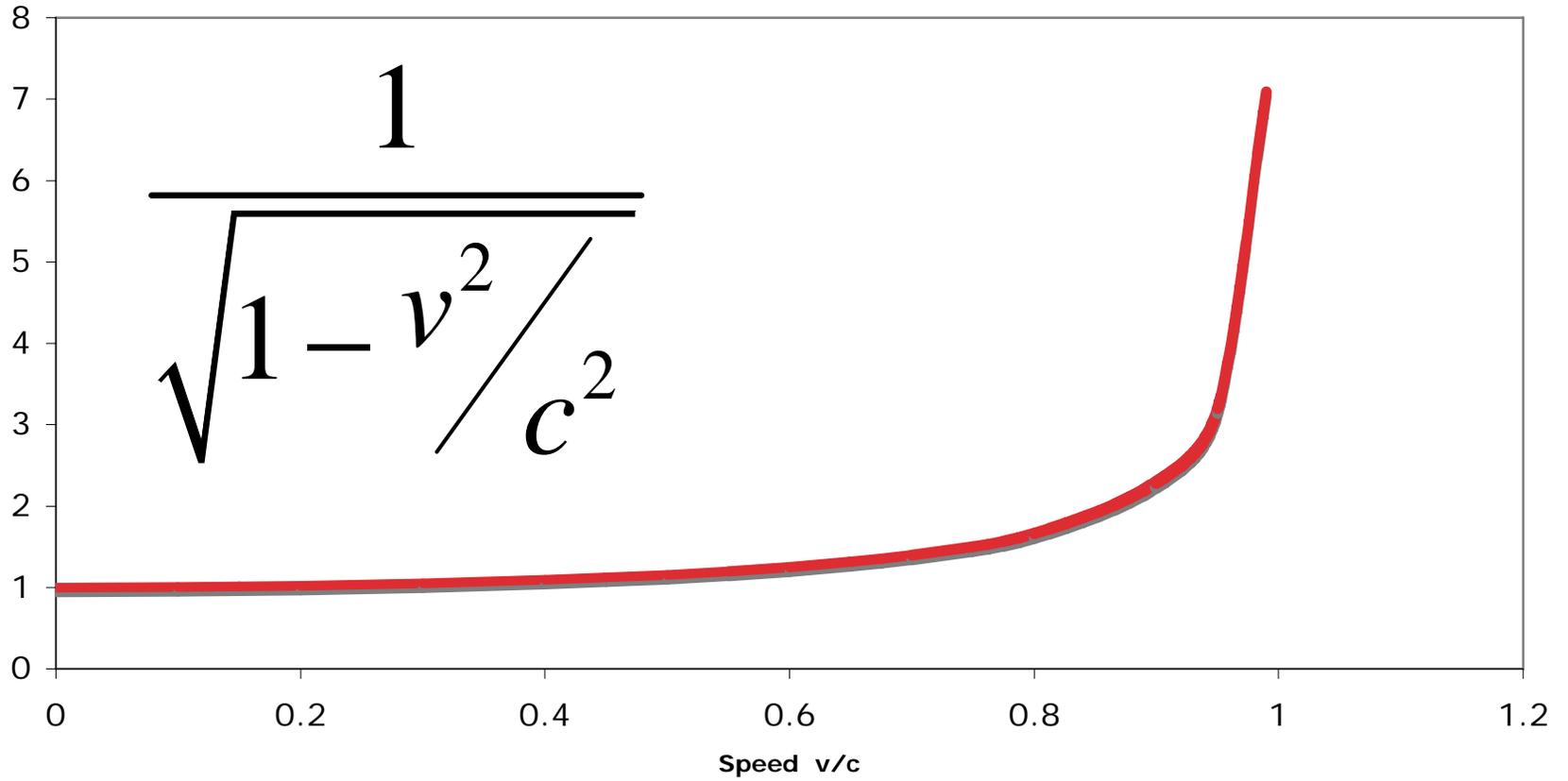
An object gets more massive when moving.

A clock slows down when moving.

A meter stick gets shorter when moving.

Einstein was thinking about Maxwell.....

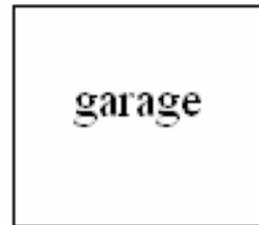
# Gamma



**If this looks BLUE, you're driving too fast!**



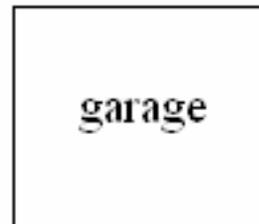
stationary ladder



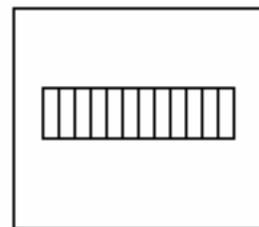
garage



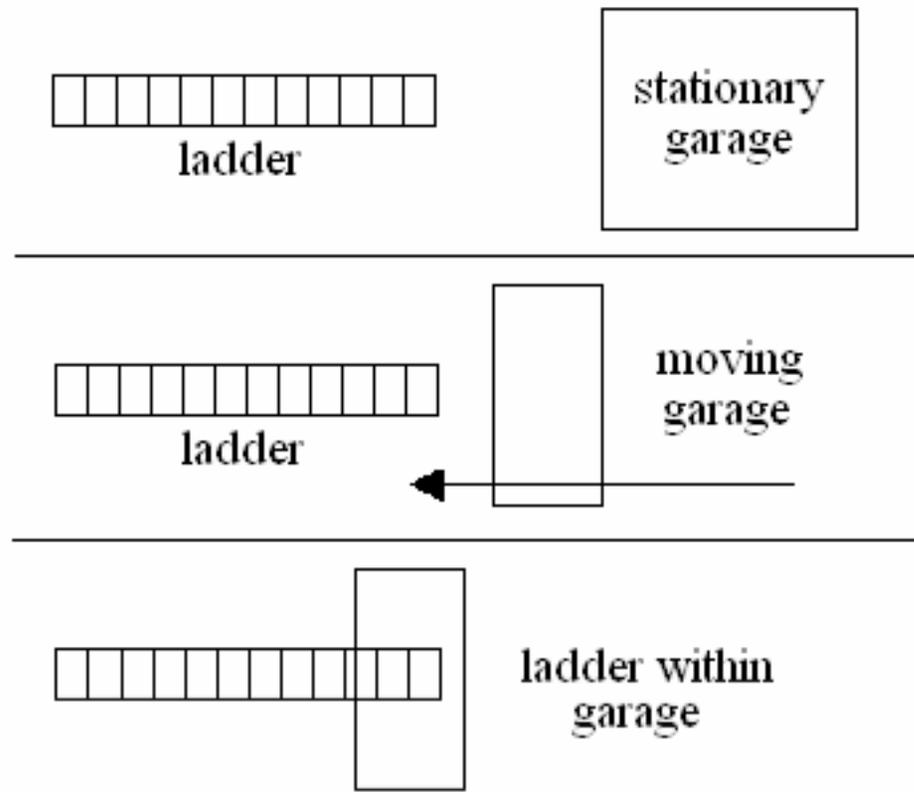
moving ladder

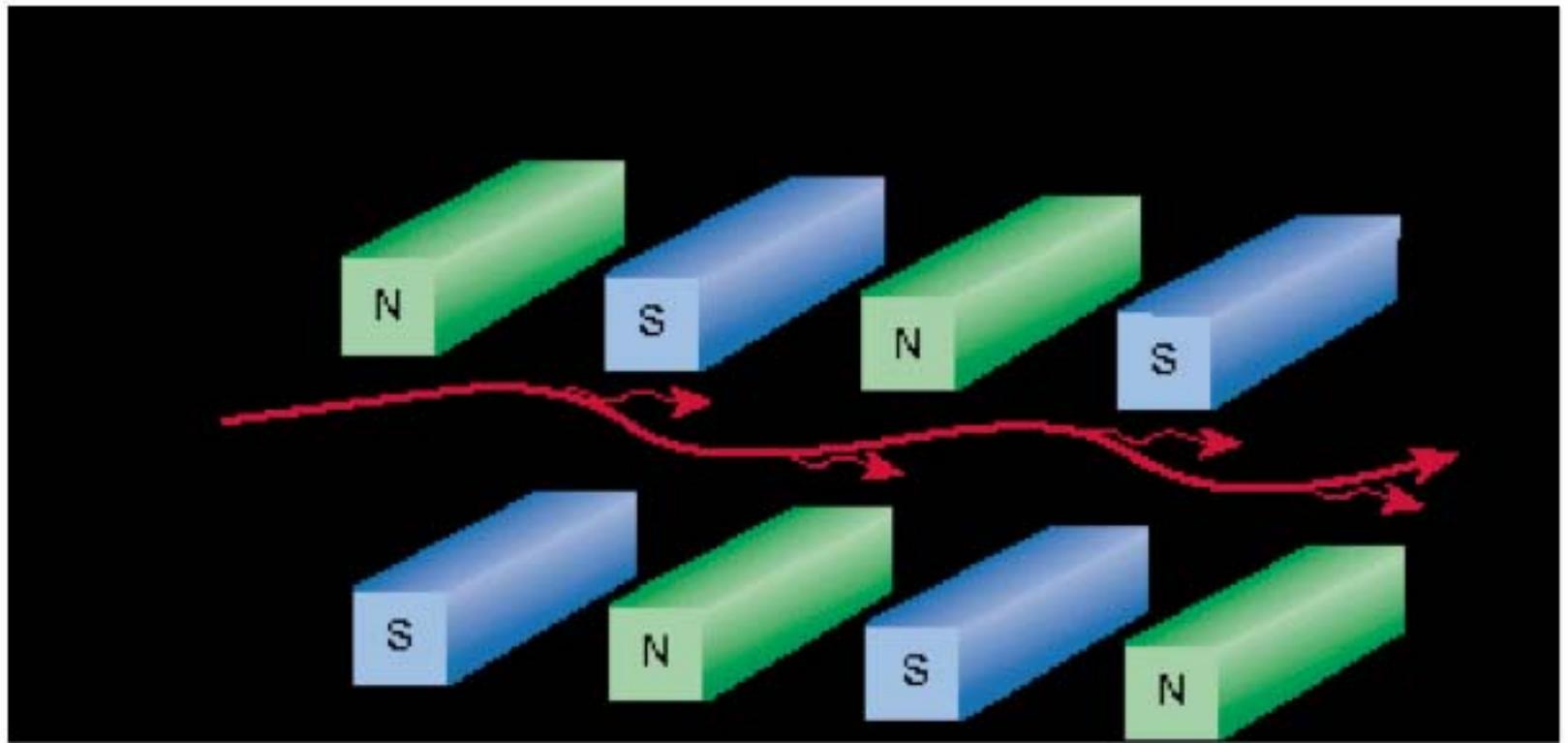


garage



ladder in garage





INSERTION DEVICE  
(Wiggler or Undulator)  
Permanent Magnetic Material  
Nd-Fe-B

Electrons

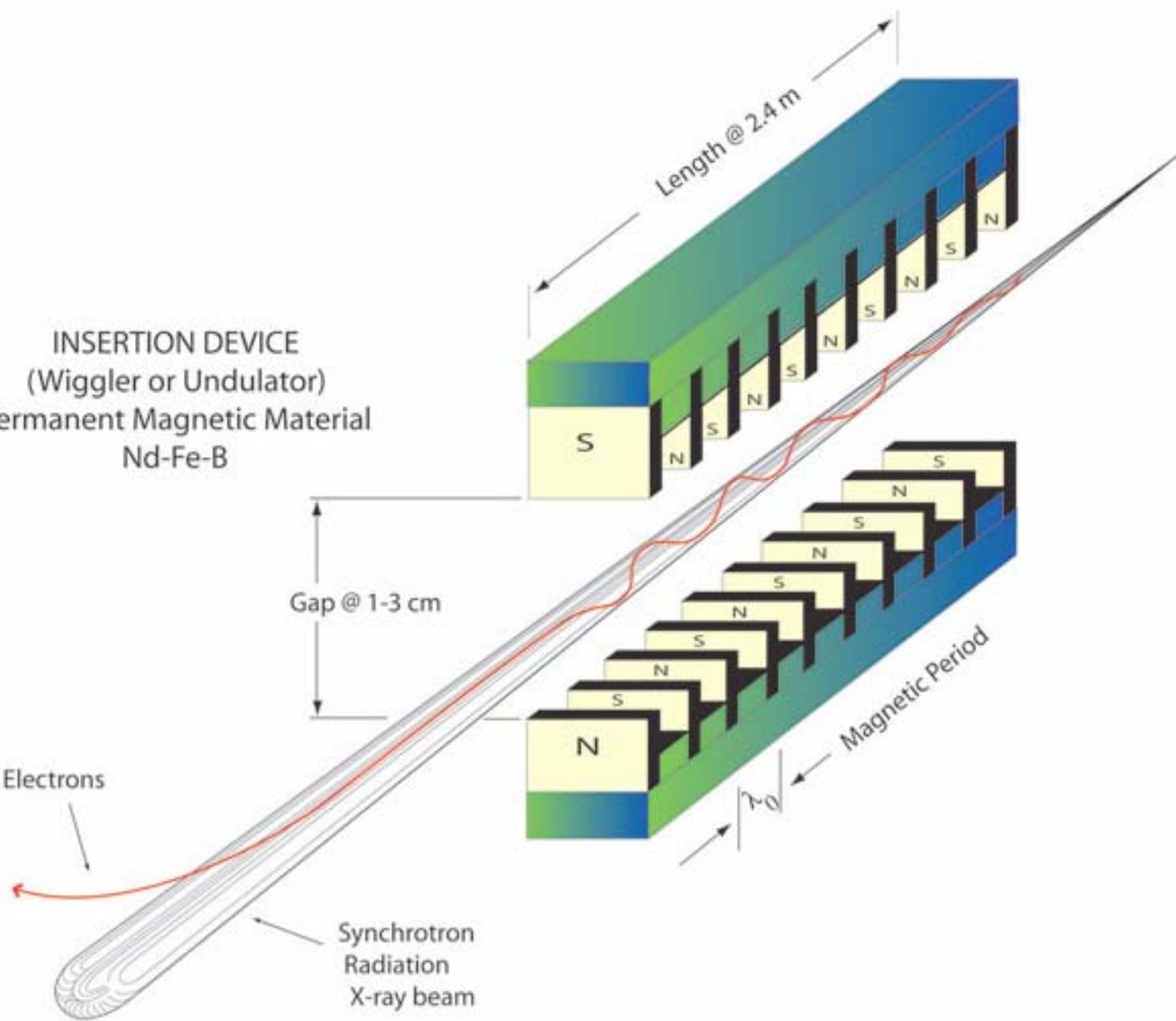
Synchrotron  
Radiation  
X-ray beam

Gap @ 1-3 cm

Length @ 2.4 m

Magnetic Period

$\lambda_0$



If you saw things from the point of view of the electrons, how big are the magnets?

A meter stick is now only 1/10 of a millimeter!!!

(gamma ~ 10,000)

The electron radiates (like Maxwell says) with a wave length of 1/10 millimeter, BUT:

We see a wave length smaller by another factor of 10,000!!

That means, we get X-Rays instead of radio waves.

Because of Maxwell, electrons radiate when we shake them.

Because of Einstein, we can build accelerators and magnets that make X-Rays of incredible power.

Because of this synchrotron radiation, we now learn the atomic structures of hundreds of proteins every year, instead of one every few years!

Theoretical Physics from a century ago gives us the latest scientific advances in biology!

What about Quantum Mechanics,  
and the Heisenberg Uncertainty Principle?

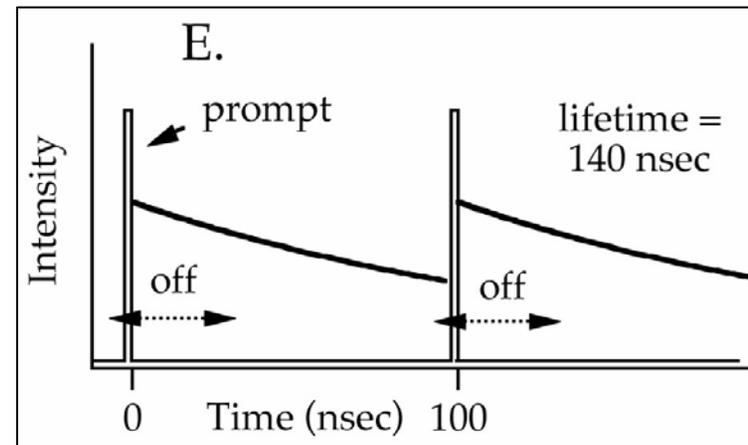
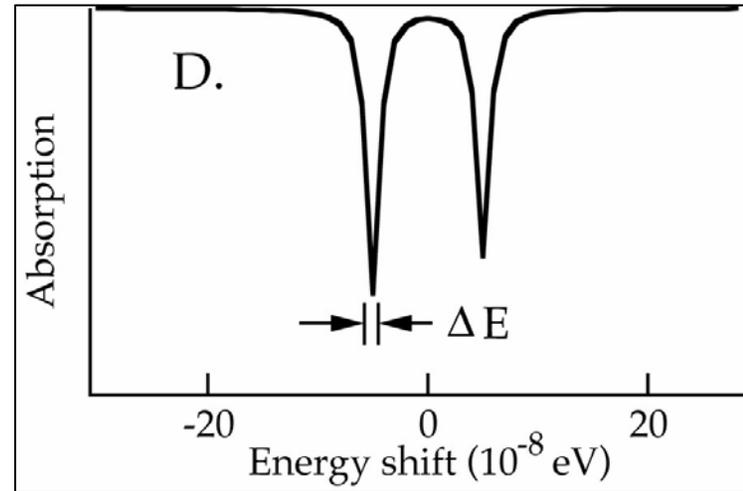
$$\Delta p \Delta x \geq h$$

$$\Delta E \Delta t \geq h$$

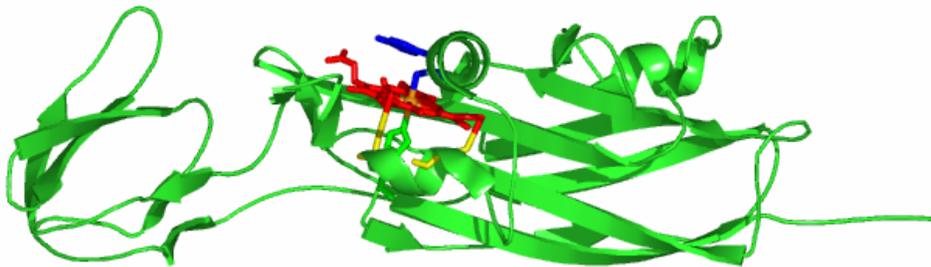
3. Absorption is governed by the Mossbauer energy width, leading to lifetime of the excited states of  $\sim 140$  nanoseconds (Heisenberg uncertainty).

4. Time between x-ray pulses is  $\sim 100$  nsec; APD detectors are gated off during the incident pulse time window.

5. This ensures that any fluorescence detected arise from interaction with the nucleus, and NOT from the much more likely interaction with core electrons.

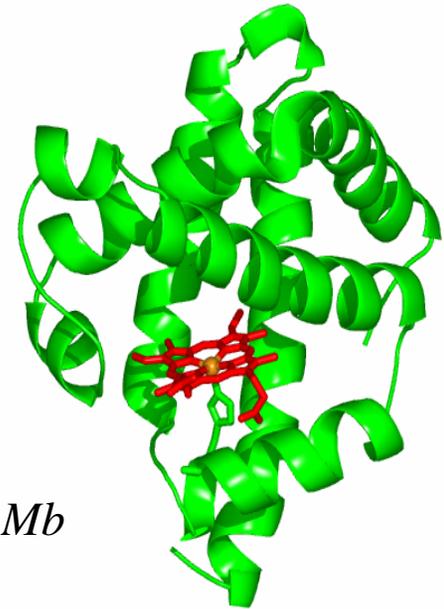


## C. Comparison of heme proteins - cytochrome f vs myoglobin



*Cytochrome f*

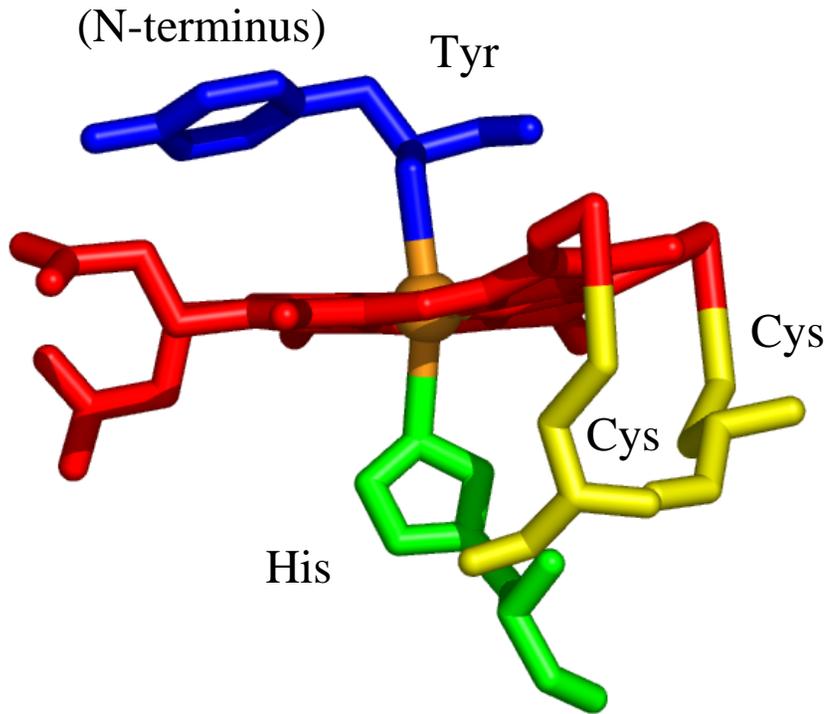
An electron-transfer membrane protein,  
part of the cytochrome *b<sub>f</sub>* complex of  
oxygenic photosynthesis. Has unusual Fe  
linkage to N-terminus of protein.



*Mb*

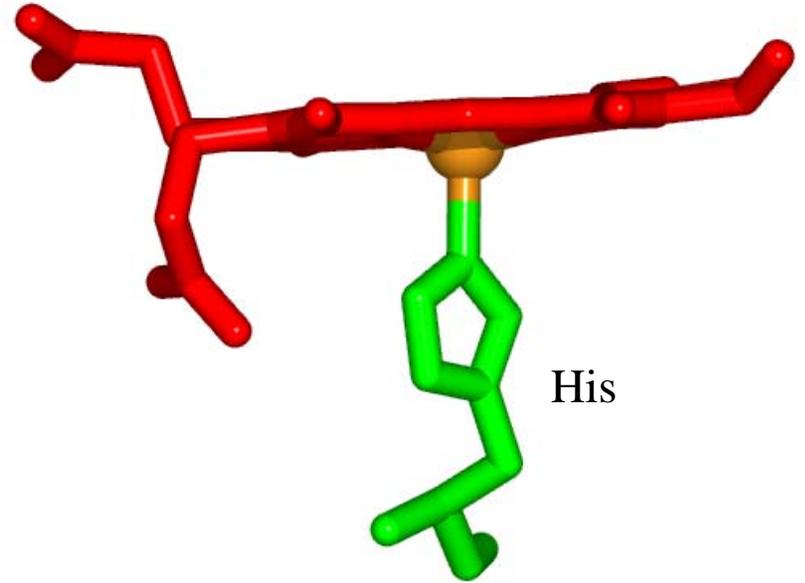
Oxygen ligand-binding  
protein, found in muscle  
tissues, etc.

Cytochrome f



Four strong bonds to protein

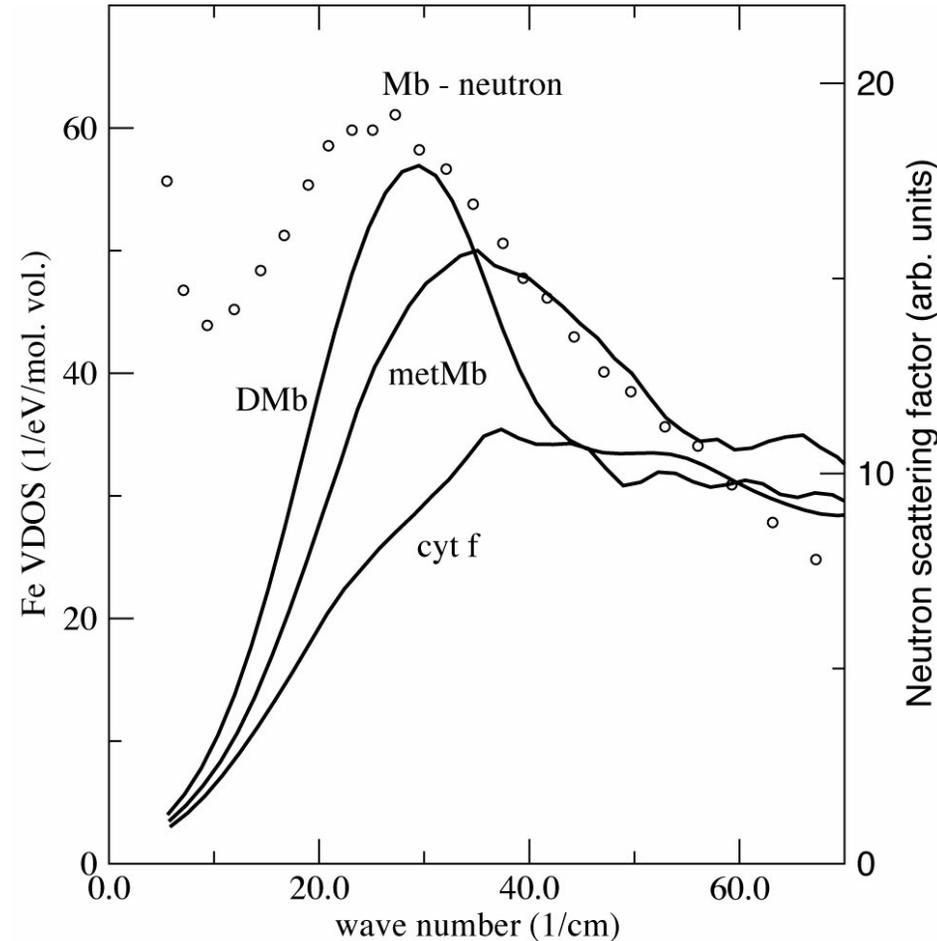
deoxy-Myoglobin



Only one bond to protein

## Protein Peak Comparison:

- *cyt f* has the smallest displacements at low, protein peak frequencies, despite the most direct coupling to protein vibrations (N-terminus)
- existence of protein peaks shows that protein normal modes are long-lived, underdamped, not dissipative.



(Neutron - Cusack & Doster 1990)

What have learned about these two proteins?

The oxygen storage protein, myoglobin, has a heme molecule that is unusually floppy, probably related to the need for motion to accommodate a small ligand like O<sub>2</sub>.

The electron-storage protein, cytochrome f, is unusually rigid. The reason for this is not clear, but one suspects it is related to the electron-storage function.

