

Network for Computational Nanotechnology (NCN)

UC Berkeley, Univ. of Illinois, Norfolk State, Northwestern, Purdue, UTEP

First Time User Guide to **Crystal Viewer Tool** on **nanoHUB.org**

NCN

nanoHUB.org

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Topics discussed

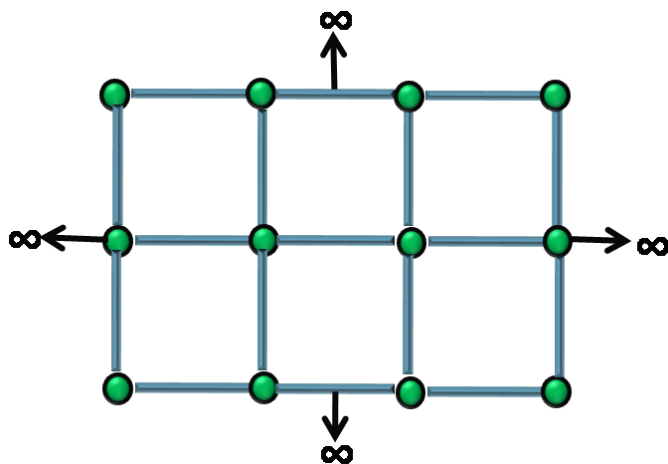
- What are crystals ?
 - » How are they represented ?
- Basics of crystallography :
 - » Lattice vectors, primitive cells, unit cell
 - » Bravais lattices, Miller indices and planes
- What is Crystal Viewer tool ?
 - » Features .
- Capabilities in the tool.
- Types of Inputs.

- What happens when you just hit SIMULATE?
- Example run for a complete simulation.
- Limitations of the tool.
- Few Words About the tool.
- References .
- Appendix

What are crystals ?

- *Crystals are*
 - » Periodic arrangement of atoms in space (**called Lattice**).
- Simplest periodic block can be a single atom
 - » Eg: Copper, Gold, Silver, Iron, etc.

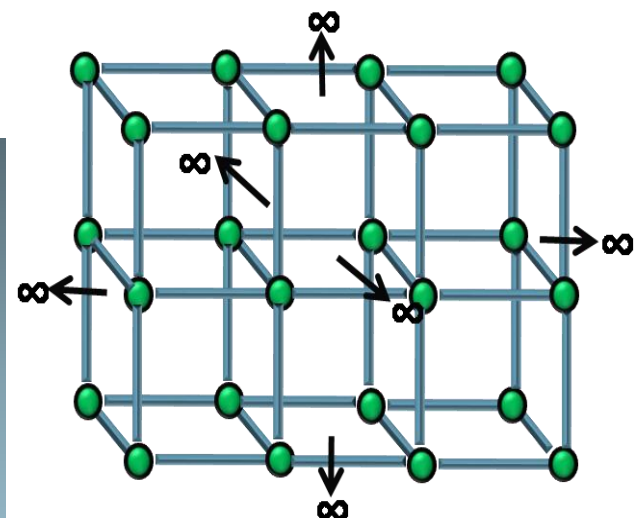
1D Lattice



2D Lattice

Atoms can be
arranged periodically
in 1, 2 and 3 D to
produce

CRYSTALS



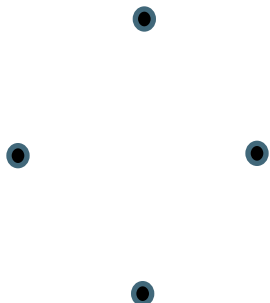
3D Lattice

Images obtained from **Crystal Viewer** tool on nanoHUB.org

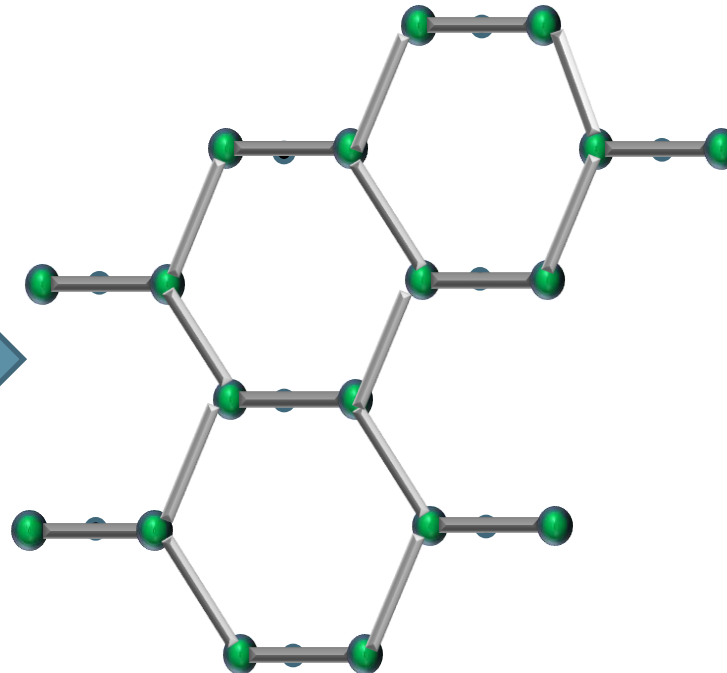
Representation of Crystals

- Crystal is represented by
 - » **Lattice** : Points in space repeated periodically
 - » **Basis** : Group of atoms repeated each lattice point.

LATTICE



BASIS

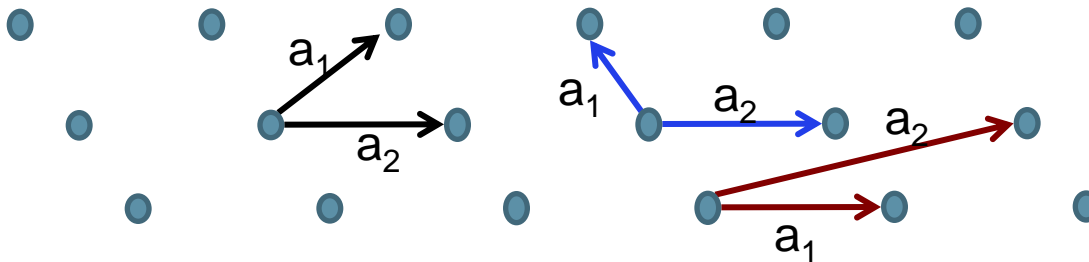


GRAPHENE

CRYSTAL

Steps to construct a crystal

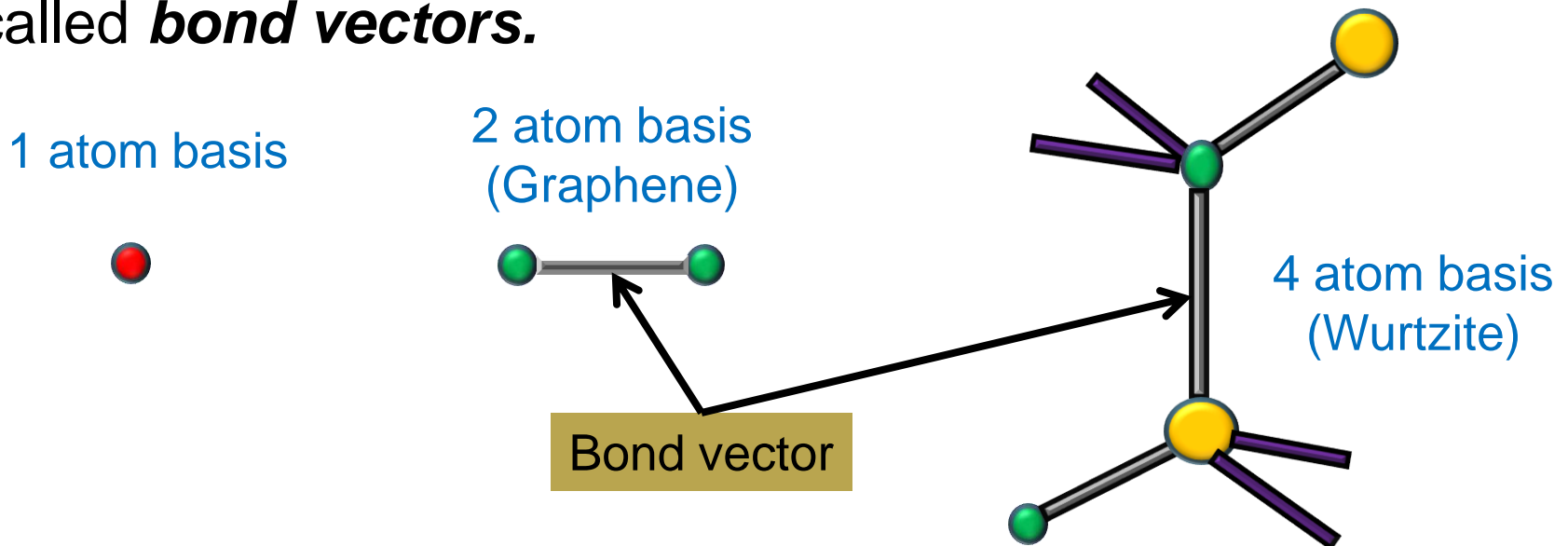
- **Basis** & **Lattice** together form the crystal.
- A basis can have one or more than one atom
 - » In the last slide basis had 2 atoms.
- Translate the basis in the space (at lattice points) to obtain the crystal structure.
 - » $T = La_1 + Ma_2 + Na_3$ T = Translational vector. $L, M, N \Rightarrow$ integers
- a_1, a_2 & a_3 are called **Lattice vectors**.
- Lattice Vectors are not necessarily **unique**.



Three different lattice vectors produce the same lattice structure.

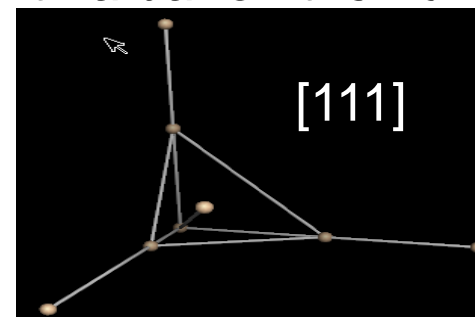
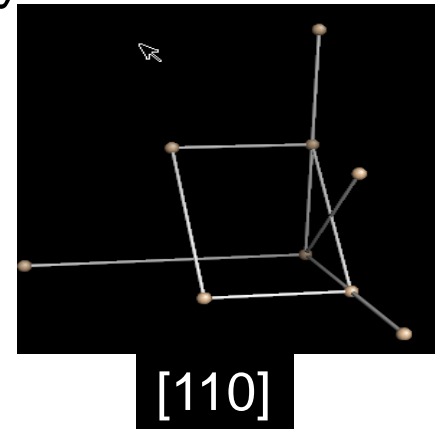
Crystallography : About the Basis...

- Basis : Can be as simple as one atom, or a complex many atom system.
- Basis is also called a ***unit cell*** or simply a ***cell***.
- Unit cell with minimum volume is a ***Primitive cell***.
- In complex cells atoms are linked together using vectors called ***bond vectors***.



Miller Planes and Indices

- Miller indices are method to describe :
 - » planes and directions in crystals.
- Indices are denoted using l, m, n along x,y,z respectively.
 - » These are inverse of intercepts on the x,y,z axis.
 - » Convert these into full integers.
 - » **Eg: intercept (3,2,6) \Rightarrow l,m,n = $6 \times (1/3, 1/2, 1/6) = (2,3,1)$.**
- (lmn) represents a miller plane along direction l,m,n.
- [lmn], denotes a direction in the basis of direct lattice.
- $\langle lmn \rangle$ denotes direction that are eqv. to [lmn].
- {lmn} denotes the family of planes eqv. to (lmn).
- Negative integers are represented by with a bar on the number.

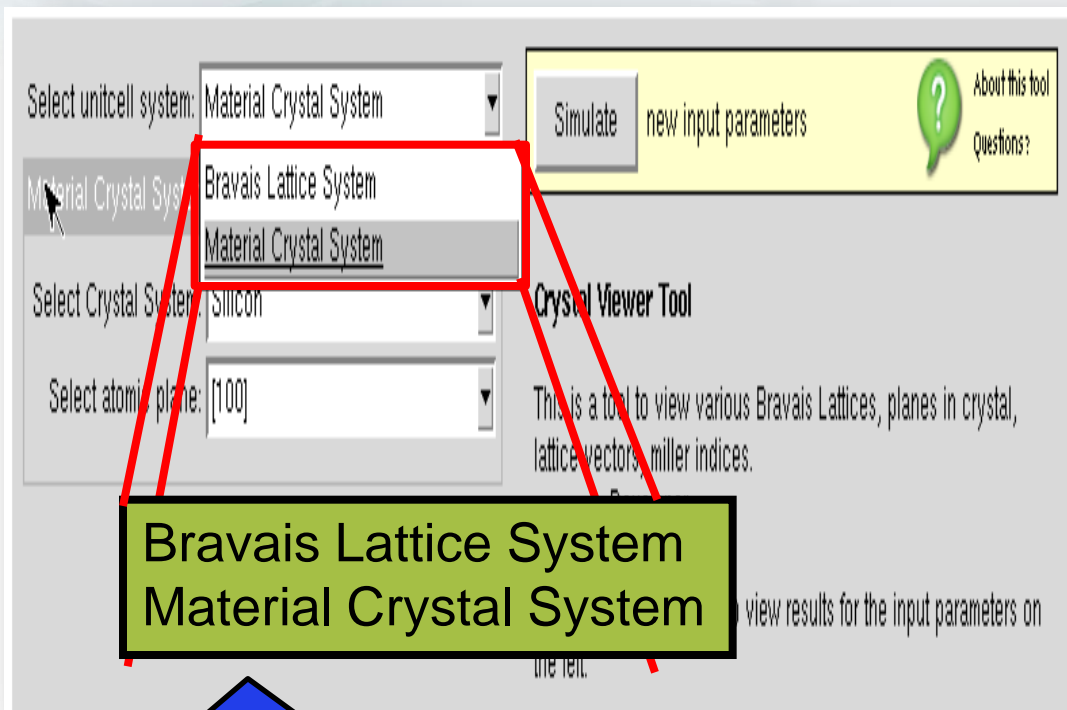


Images obtained from
Crystal Viewer tool on
nanoHUB.org

What is Crystal Viewer Tool ?

- Crystal Viewer Tool :
 - » A MATLAB® based.
 - » Used for visualization of :
 - ✓ Different material crystals.
 - ✓ Different Bravais Lattices.
 - » Provides information about miller indices and planes.
 - » Provides information about crystals.
 - » Tool has been developed at Purdue University.
 - ✓ Part of the teaching tools on nanoHUB.org.
- Developers : (version 1.23)
 - » Abhijeet Paul / Purdue University.
 - » Gerhard Klimeck / Purdue University.

Crystal Viewer Tool : First look



Capabilities of the tool :

- ✓ Visualize Bravais lattices.
- ✓ Visualize larger grids.
- ✓ Visualize miller planes.
- ✓ Visualize material crystals.
- ✓ Visualize atomic planes.
- ✓ Obtain crystal information.

Check out the tool webpage for latest updates, features and bug fixes at :

http://nanohub.org/resources/crystal_viewer

Types of Inputs

Input Type 1

Material Crystal Systems

Select Crystal System: Silicon

Select atomic plane: Silicon

Select atomic plane: [100]

[1] Allowed Material Systems

[2] Allowed Atomic Planes

Input Type 2

Bravais Lattice: Cubic

Cubic

Tetragonal

Orthorhombic

Monoclinic

Triclinic

Trigonal(Rhombohedral)

Hexagonal

Size of s

a: 1

b: 2

c: 3

Simple Cubic (SC)

Body Centered Cubic(BCC)

Face Centered Cubic(FCC)

Angle of the unitcell in degree

alpha: 60

beta: 60

gamma: 60

Grid Size: 5

[1] Select Bravais lattice

[2] Select unit cell

[3] Select unit cell parameters

[4] Select gridsize

Input Type 3

Show Miller Plane: ☒ yes

Miller Indices

l: 1

m: 0

n: 0

[1] Select if you want to see miller planes

[2] Select the miller indices.

Images obtained from
Crystal Viewer tool on
nanoHUB.org

What happens when you just hit SIMULATE?

Default Inputs

Select unitcell system: Material Crystal System

Material Crystal Systems

Select Crystal System: Silicon

Select atomic plane: [100]

Runs with the following default inputs:

[Select unitcell system]: Material Crystal System.

[Select Crystal Material]: Silicon

[Select atomic plane]: [100]

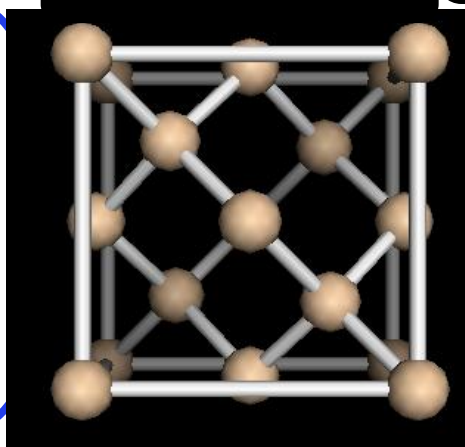
Result: Crystal Information

```

=====
                        CRYSTAL PARAMETER INFORMATION
=====
Material : Si
Crystal Type : Zinc Blende
Relative dielectric constant : 11.7
Band Gap [eV] : 1.12
ax [nm] : 0.5432
ay [nm] : 0.5432
az [nm] : 0.5432
Primitive Vectors [x y z] :
1 : [ 0.50 0.50 0.00 ]
2 : [ 0.50 0.00 0.50 ]
3 : [ 0.00 0.50 0.50 ]
Bond Vectors (anion to cation) [x y z] :
1 : [ 0.25 0.25 0.25 ]
2 : [ 0.25 -0.25 -0.25 ]
3 : [ -0.25 0.25 -0.25 ]
4 : [ -0.25 -0.25 0.25 ]
    
```

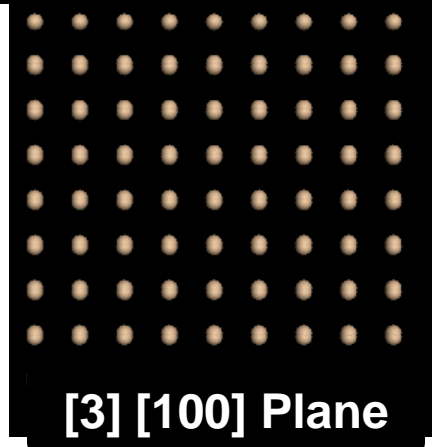
[1] Crystal Information

[2] Unitcell



Default Outputs

[3] [100] Plane



Bravais lattice simulation : Input to Output

INPUTs

Bravais Lattice

Bravais Lattice: Triclinic

[1] Select type of Bravais lattice

Select the Cell type

Choice: Simple Primitive Cell (PC)

[2] Select the type of unitcell

Size of sides of the unitcell(a ,b, c)

a: 1

b: 2

c: 3

[3] Select a,b,c sides of cell

Angle of the unitcell in degrees

alpha: 60

beta: 60

gamma: 60

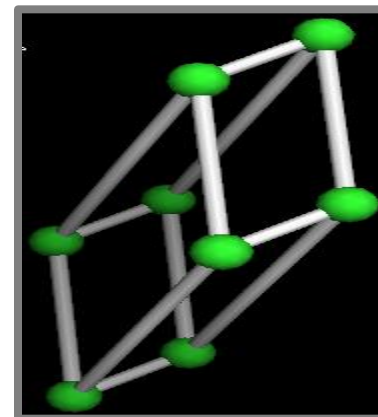
[4] Select cell angles (α, β, γ)

Grid Size: 5

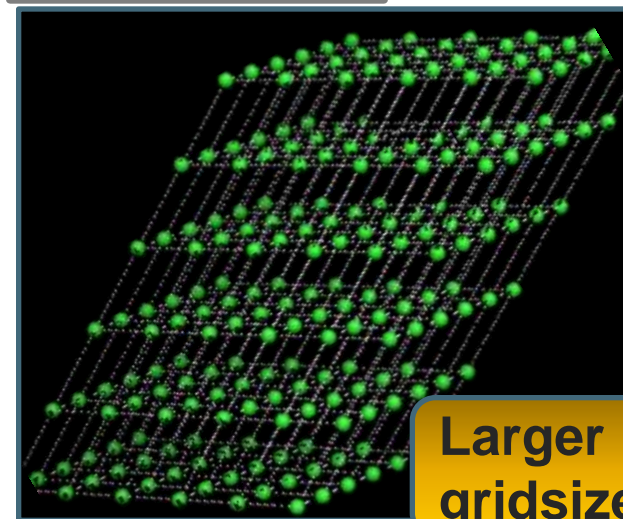
[5] Select if larger (>1) grid size needed.



OUTPUTs



Triclinic unitcell



Larger gridsize

Limitations of the tool

- Visualization of miller planes is not very good
 - » normal vector and plane are not very helpful.
 - ✓ work is underway to improve this.
- Cannot take cut planes in the larger crystal systems.
- Improvement in atomic plane visualization needed.

Always keep checking the tool web-page for latest features, releases and bug-fixes at : http://nanohub.org/tools/crystal_viewer

Few words regarding the tool

- Use this tool to learn about crystals.
- Feel free to post about
 - »the bugs
 - »new features you want.
- Contact the developers in case you want to collaborate for some work using this tool.

Always keep checking the tool web-page for latest features, releases and bug-fixes at : http://nanohub.org/tools/crystal_viewer

References

- URLs on crystal information, miller index & Bravais lattice
 - » <http://cst-www.nrl.navy.mil/lattice/>
 - » http://en.wikipedia.org/wiki/Bravais_lattice
 - » http://serc.carleton.edu/research_education/crystallography/xldatabases.html
 - » <http://webmineral.com/>
 - » http://en.wikipedia.org/wiki/Miller_index
- Books to read about crystals
 - » [Kittel, Charles \(1996\) \[1953\].Introduction to Solid State Physics \(Seventh Edition ed.\). New York: John Wiley & Sons. pp. 10. ISBN 0-471-11181-3. http://www.wiley.com/](#)

References

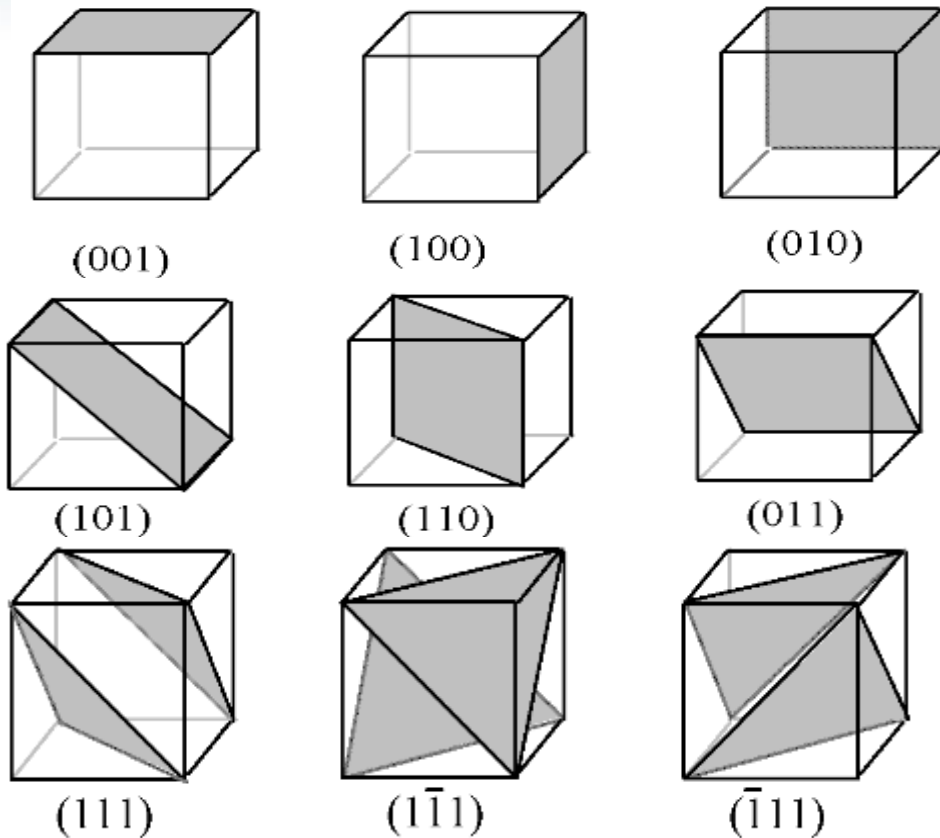
- Homework assignment using the tool.
 - » <https://nanohub.org/resources/4200>
- Link for the tool :
 - » http://nanohub.org/resources/crystal_viewer

Appendix A : Bravais lattices

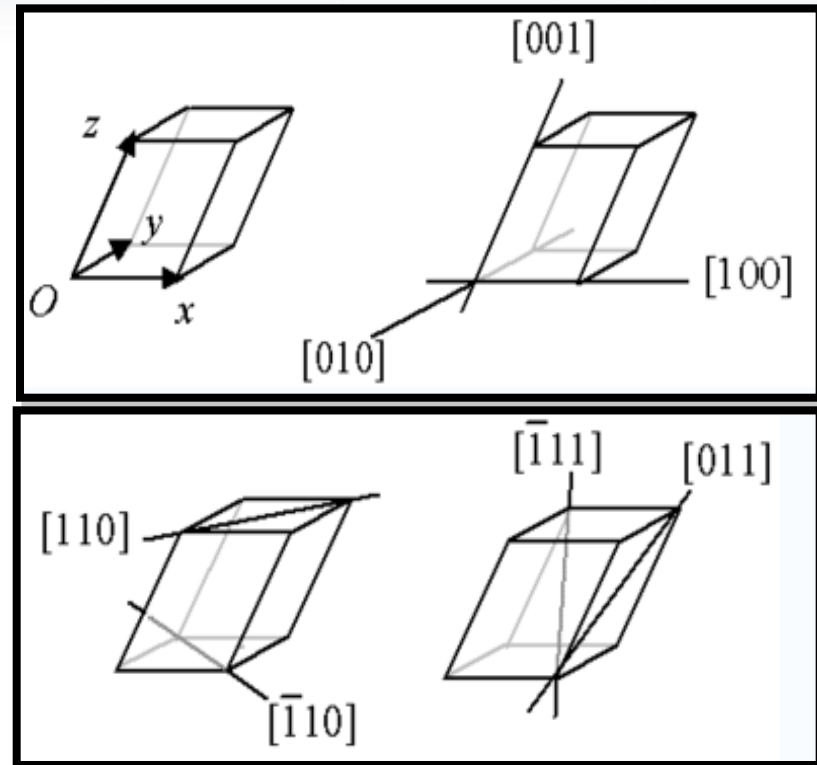
Distance Parameters (a, b, c)	Angle Parameters (α, β, γ)	Name	No. of Lattices / Types	Example
$a = b = c$	$\alpha = \beta = \gamma = 90^\circ$	Cubic	3 S,B,F	CsCl, NaCl
$a = b \neq c$	$\alpha = \beta = \gamma = 90^\circ$	Tetragonal	2 S,B	TiO ₂ (Rutile), SnO ₂ (Cassiterite)
$a \neq b \neq c$	$\alpha = \beta = \gamma = 90^\circ$	Orthorhombic	4 S,B,E,F	KNO ₃ , BaSO ₄ (Baryte)
$a = b = c$	$\alpha = \beta = \gamma < 120^\circ$ $\neq 90^\circ$	Rhombohedral (Trigonal)	1 S	Ice, Al ₂ O ₃
$a = b \neq c$	$\alpha = \beta = 90^\circ$; $\gamma = 120^\circ$	Hexagonal	1 S	Mg, Zn, PbI ₂
$a \neq b \neq c$	$\alpha = \gamma = 90^\circ$; $\beta \neq 90^\circ$	Monoclinic	2 S,E	Monoclinic Sulphur, Na ₂ SO ₄ ·10H ₂ O
$a \neq b \neq c$	$\alpha \neq \beta \neq \gamma \neq 90^\circ$	Triclinic	1 S	CuSO ₄ ·5H ₂ O

S: simple B: body centered F: face centered E: end face centered

Appendix B: Miller Planes & Directions



Some Important Miller planes



Some Important Miller directions

Images from http://en.wikipedia.org/wiki/Miller_index Author: Christophe Dang Ngoc Chan