

Fundamentals of Nanoelectronics

Prof. Supriyo Datta
ECE 453
Purdue University

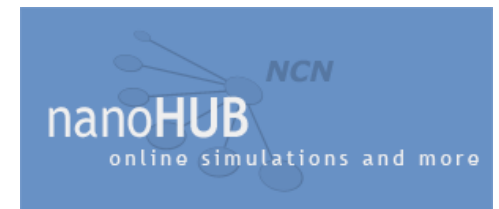
10.29.2004

Lecture 25: Density Of States: General Approach

Ref. Chapter 6.2

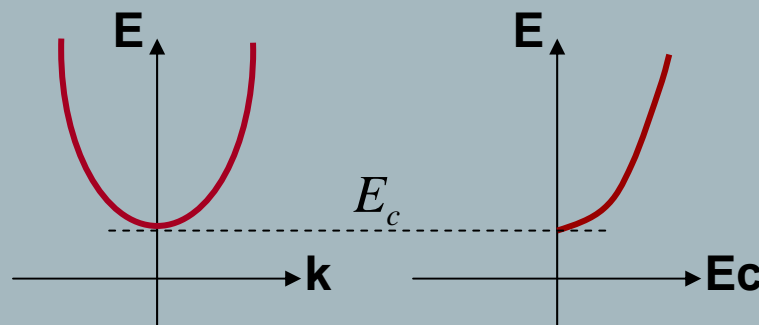


Network for Computational Nanotechnology



- We like to be able to learn DOS in a more general sense. The dispersion relation can be used to derive DOS with this fact in mind that the dispersion relation exists only for periodic structures. Note that even without dispersion relation, DOS still exists. In that sense DOS is a more fundamental concept than dispersion relation.
- Remember what we've done last day:

3-D

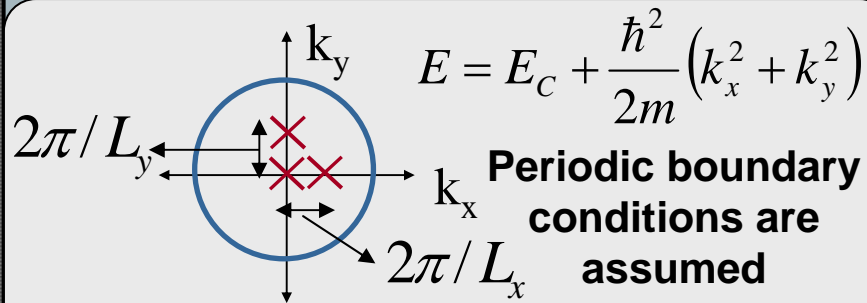


$$E = E_c + \frac{\hbar^2 k^2}{2m} \Rightarrow D(E) \propto \sqrt{E - E_c}$$

From E-k to DOS $N(k) \rightarrow N(E) \rightarrow D(E)$

02:30

- For 2D we have:



- The question to be asked is that how many states we have inside the circle above? In 1D we looked at the total length. In this case we look at the area of the circle considering the fact that each state occupies the area of:

$$(2\pi/L_x)(2\pi/L_y)$$

- We need to know how many of above can be fit into the circle. That will be the total number of states:

$$N_T(k) = \frac{\pi(k_x^2 + k_y^2)}{\frac{2\pi}{L_x} \frac{2\pi}{L_y}}$$

- If we substitute: $L_x L_y = S$, then

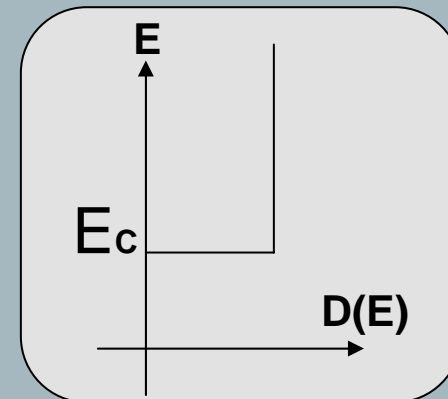
$$N_T(k) = \pi k^2 \frac{S}{4\pi^2} \quad k^2 = k_x^2 + k_y^2$$

- For $D(E)$ we have:

$$D(E) = \frac{S}{4\pi^2} \pi \frac{d}{dE}(k^2) = \frac{S}{4\pi^2} \pi \frac{2m}{\hbar^2}$$

$$D(E) = \frac{Sm}{2\pi\hbar^2}$$

- One point to notice is that in this case, DOS does not depend on energy.



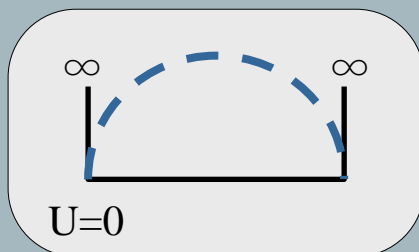
Periodic BC versus Infinite Wall BC

05:02

$$N_T(k) = \frac{\pi (k_x^2 + k_y^2)}{2\pi \frac{2\pi}{L_x} \frac{2\pi}{L_y}}$$

- Notice that the $2\pi/L_x$ in the above relation is the result of Periodic Boundary Conditions (PBC). This may not be necessarily true for solids but the physical argument is that what happens at the boundaries does not affect the center and we are interested in the center. Assuming PBC means: $k_x L_x = 2\pi \nu_x$

- Another approach would be to assume infinite potential wall Boundary Conditions (BC).



- In this case wavefunctions have to go to zero at the two ends. If so instead of exponentials, “sin” functions are the form of solutions.

$$\sin k_x x \Rightarrow k_x L_x = \pi \nu \Rightarrow k_x = (\pi/L_x) \nu$$

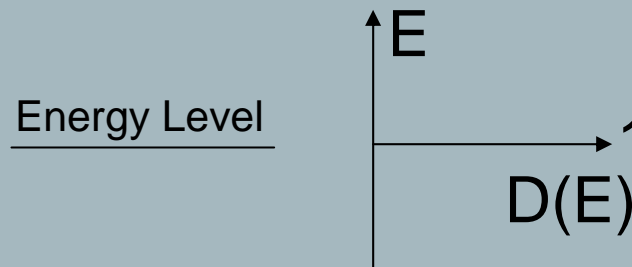
- This means that k states will be π/L_x apart instead of $2\pi/L_x$ of periodic boundary conditions. Does this mean that we have twice as many states?
- The answer is no. For PBC solutions are exponential and $\exp(ikx)$ is linearly independent of $\exp(-ikx)$. So there are actually two solutions for k and $-k$. However in the case of “sin” function: $\sin(-kx) = -\sin(kx)$. You can clearly see that these are two linearly dependent answers. In summary again:

$$N(k) \rightarrow N(E) \rightarrow D(E)$$

Overview: DOS (General Approach)

11:29

- Given any piece of solid its energy levels can be described by the eigenvalues of the Hamiltonian matrix $[H]$ for that solid. These eigenvalues can be denoted as ϵ_{α} . Alpha is the subscript that indicates which eigenvalue we are considering.
- What would DOS be if we had only one energy level? It would be a delta function.



- So for each energy level we have a delta function of strength 1. So we can write DOS as:

$$D(E) = \sum_{\alpha} \delta(E - \epsilon_{\alpha})$$

- Let's try to do this for 1-D. If we have dispersion relation like: $E = E_c + \frac{\hbar^2 k^2}{2m}$

- Then we can write DOS in terms of k states: $D(E) = \sum_k \delta(E - \varepsilon_k)$

- If the states are really closely spaced then we can write the summation as an integral:

Useful identities for the integration below

$$\left\{ \begin{array}{l} \int \delta(E) dE = 1 \\ \int \delta(E - \varepsilon) dE = 1 \\ \int \delta(\varepsilon - E) d\varepsilon = 1 \end{array} \right.$$

$$\varepsilon_k = E + \frac{\hbar^2 k^2}{2m} \Rightarrow \frac{d\varepsilon_k}{dk} = \frac{\hbar^2}{m} \frac{\sqrt{2m(\varepsilon_k - E_c)}}{\hbar} = \frac{\hbar \sqrt{2m(\varepsilon_k - E_c)}}{m}$$

$$D(E) = \sum_k \delta(E - \varepsilon_k) = 2 \frac{L}{2\pi} \int dk \delta(E - \varepsilon_k) = 2 \int \frac{d\varepsilon_k \cdot L}{2\pi} \frac{dk}{d\varepsilon_k} = \int \frac{d\varepsilon_k \cdot L}{2\pi} \frac{m}{\hbar \sqrt{2m(\varepsilon_k - E_c)}} \delta(E - \varepsilon_k)$$

Identity: $\int dE \delta(\varepsilon - E) f(\varepsilon) = f(E)$

$$D(E) = 2 \frac{L}{2\pi} \frac{m}{\hbar \sqrt{2m(E - E_c)}} \quad 1D$$

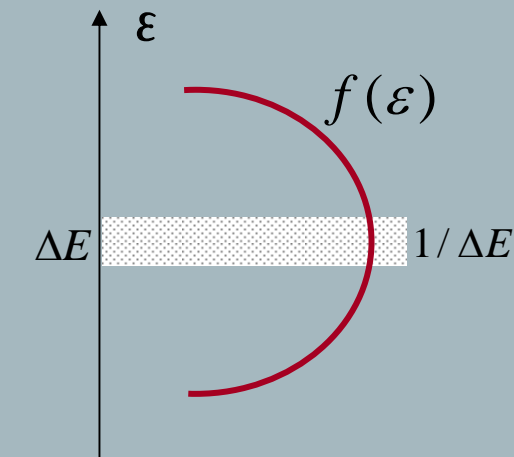
A Note About Delta Function

32:52

- To derive DOS in 1D we used the following identity:

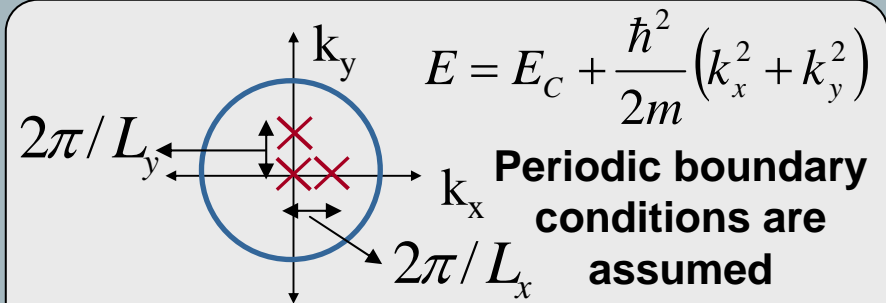
$$\int dE \delta(\varepsilon - E) f(\varepsilon) = f(E)$$

- Consider the picture below to see why:



$$\int dE \delta(\varepsilon - E) f(\varepsilon) = f(E)$$

- Let's derive DOS for a 2D solid:



$$E = E_C + \frac{\hbar^2}{2m} (k_x^2 + k_y^2)$$

Periodic boundary conditions are assumed

$$D(E) = \sum_{k_x, k_y} \delta(E - \varepsilon_{k_x, k_y}) = \int \frac{dk_x L_x}{2\pi} \int \frac{dk_y L_y}{2\pi} \delta(E - \varepsilon_k) = \frac{S}{4\pi^2} \int dk_x dk_y \delta(E - \varepsilon_k)$$

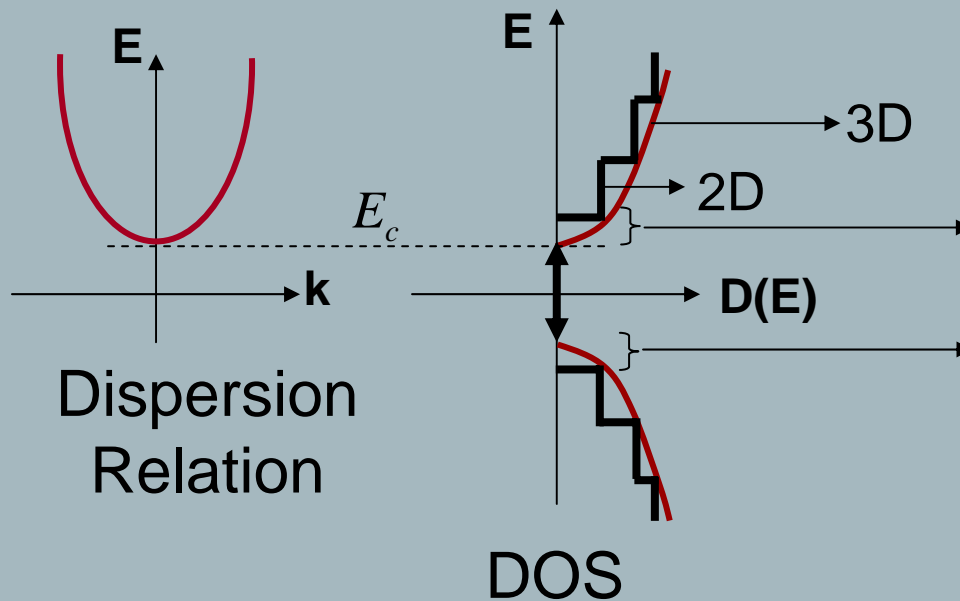
- Putting this in circular coordinates...

$$D(E) = \frac{S}{4\pi^2} \int 2\pi k dk \delta(E - \varepsilon_k) = \frac{S}{2\pi} \int \frac{k d\varepsilon_k}{d\varepsilon_k / dk} \delta(E - \varepsilon_k) = \frac{Sm}{2\pi\hbar^2} \int d\varepsilon_k \delta(E - \varepsilon_k)$$

$$D(E) = \frac{Sm}{2\pi\hbar^2} \quad 2D$$

Blue Shift

43:30



This gap is due to confinement. As one goes from 3D to 2D the solid is confined in 1 dimension and the energy levels become discretized in that direction. The ground state energy will not be the bottom of conduction band. This has confirmed to be the case by optical experiments. Whereas the gap between the conduction band edge and valence band edge is 1.5eV, the photon that is emitted by the material has higher energy indicating that the first energy levels occur higher than the conduction band edge and lower than valence band edge. For 3D the light is in infrared range and for 2D the light shifts towards shorter wavelengths of blue range. Because of this, people call this type of confinement the blue shift.